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HOMOGLUTATHIONE SYNTHETASE AND THE PLANT

THIOL-REDOX PROTEOME

By

Ashley L. Galant

A dissertation presented to the
Graduate School of Arts and Sciences
of Washington University in
partial fulfillment of the
requirements for the degree
of Doctor of Philosophy

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ABSTRACT OF THE DISSERTATION

Homoglutathione Synthetase and the Plant Thiol-Redox Proteome

by

Ashley Galant

Doctor of Philosophy in Biology and Biomedical Science (Plant Biology)

Washington University in St. Louis, 2011

Professor Joseph M. Jez, Chairperson

In the plant cell, redox regulation and redox responsiveness are governed by a series of mechanisms that hinge upon the use of small molecule redox-couples and reversible, thioredoxin- or glutaredoxin-mediated protein disulfides. This work examines first the structural basis for synthesis of these small molecules and second how plants are able to adapt and respond to changes in environmental redox state.

Among the major redox-couples, glutathione (GSH) is maintained at the highest cellular concentrations, and is furthermore employed in a protective capacity as an anti-xenobiotic and anti-oxidation protein thiol-modification. Almost all eukaryotes utilize GSH, but some legumes additionally synthesize homoglutathione (hGSH), which is a GSH analog that contains a terminal β -alanine residue instead of a terminal glycine. This alternate reaction is catalyzed by hGSH synthetase, which is related to GSH synthetase; however, the specific features that alter substrate specificity are unknown. To understand the molecular basis for the synthesis of the legume-specific molecule, the three-dimensional structure of hGSH synthetase from *Glycine max* (soybean) was solved by x-

ray crystallography in three forms - apoenzyme, bound to γ -glutamylcysteine, and with hGSH, ADP, and a sulfate ion bound in the active site. Comparison of these structures with those of GSH synthetase suggest that two residues - a leucine and a proline in the Ala-rich loop region of the enzymes - dictate the use of β -alanine instead of glycine in hGSH synthetase. Site-directed mutagenesis studies and kinetic analysis further support this conclusion.

As a means of regulating activity, many plant proteins limit access to their active sites and control the aggregation of catalytic oligomeric complexes through the formation of redox-reversible disulfide bonds. In order to identify plant proteins and pathways that utilize such bonds and/or thiol modifications to modulate oxidation state, an N-ethylmaleimide- and 5-iodoacetamidofluorescein-based dual-labeling strategy was employed in conjunction with 2D-gel electrophoresis and LC-MS/MS. Initial experiments with root protein extracts from *B. juncea* identified several new proteins that were differentially expressed and/or oxidized in response to exposure to the glutamate-cysteine ligase inhibitor buthionine sulfoxide or H₂O₂. A clear lack of overlap between the proteins altered by each condition was also noted. To assess oxidative changes to the plant thiol-redox proteome under agriculturally relevant conditions, soybean plants were field grown under ambient and elevated tropospheric ozone concentrations. Investigation into changes in protein expression and oxidation state again yielded numerous novel protein responses. Intriguingly, many of the largest changes were observed in pathways involved in core carbon metabolism, a sharp contrast to the changes in redox-centric pathways seen following acute ozone exposure. This observation, in conjunction with a

comparison of protein responses across several different ozone concentrations, led to the conclusion that ozone exposure is governed by a threshold effect: a concentration at which the plants transition from an active redox response toward maintenance of core processes and metabolism.

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For everyone who believed I could.

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LIST OF ABBREVIATIONS

<u>Abbreviation</u>	<u>Full Name</u>
redox	reduction-oxidation
ROS	reactive oxygen species
NAD(P)H	nicotinamide adenine dinucleotide (phosphate)
ATP	adenosine triphosphate
γ EC	γ -glutamylcysteine
GSH	glutathione
GSSG	oxidized glutathione
hGSH	homoglutathione
GCL	glutamate cysteine ligase
GS	glutathione synthetase
hGS	homoglutathione synthetase
SAR	systemic acquired resistance
NEM	N-ethylmaleimide
IAM	iodoacetamide
IAA	iodoacetic acid
DTT	dithiothreitol
TCEP	tris[2-carboxyethyl]phosphine
IAF	5'-iodoacetamidofluorescein
HPLC	high performance liquid chromatography

FPLC	fast protein liquid chromatography
MS	mass spectrometry
SDS-PAGE	sodium dodecyl sulfate polyacrylamide gel electrophoresis
MALDI	matrix-assisted laser desorption/ionization
BSO	buthionine sulfoximide
NMVOCS	non-methane volatile organic chemicals
ppb	parts per billion
FACE	free air concentration enrichment

CHAPTER 1

INTRODUCTION

In chemistry, the term “oxidation” describes a means by which an ion, atom, or molecule loses total possession of one or more of its electrons due to an increase in its overall oxidation state [IUPAC, 1997]. As a concept, the notion of oxidation arose out of the Phlogiston theory of the 17th century, which sought to explain the alchemical processes behind fire and rust, among other transformations. Following the 18th century work of Antoine Lavoisier, who identified the roles of oxygen and hydrogen in many chemical reactions, the tenets of modern oxidation theory began to gain traction among scientists and thinkers [Lavoisier, 1789]. In current parlance, the concept of oxidation exists in a duality with that of reduction, which (as the exact opposite of oxidation) describes the gain of total electron possession by an ion, atom or molecule. However, following Lavoisier’s investigation of oxidation, more than 100 years passed before scientists in the field of electrochemistry drafted the ionic theory of dissociation, which defined the paired, yet opposite, nature of oxidation and reduction reactions [Jensen, 2007].

While redox (reduction-oxidation) reactions are traditionally associated with chemistry, they also perform a number of critical functions in biology, particularly in plants. For example, the successful production of glucose via photosynthesis relies on redox reactions for the oxidation of water to molecular oxygen, the reduction of NADP^+ to NADPH, and the reversible reduction and oxidation of the protein ferredoxin [Schurmann and Buchanen, 2008]. As the process complimentary to photosynthesis, cellular respiration in both plants and animals likewise relies on many of the same reactions, only driven in the reverse direction (i.e., the reduction of molecular oxygen to

water). Many other reactions utilized by plants, including those associated with xenobiotic metabolism, light-sensing, and herbivory signaling, also employ redox chemistry.

Although their reversible nature is a boon in terms of adaptability, the widespread adoption of redox reactions by biological systems has resulted in a number of difficulties, both large scale and small. From a macro-scale perspective, most of the earth's habitable spaces exist in an atmosphere composed of approximately 78% N₂, 21% O₂, and 1% argon gas. While both N₂ and argon are largely inert under the prevailing atmospheric conditions, O₂ is more reactive. In the presence of heat and light, O₂ can degrade and recombine into a number of free radical species, among them peroxide, superoxide, and ozone, all of which are effective oxidizing agents [Halliwell, 2006]. In the absence of either preventive or responsive mechanisms, these radicals, or reactive oxygen species (ROS), can rapidly permeate cell membranes, abolish redox reaction equilibrium through large scale generation of oxidized reactants/products, and irreversibly inactivate redox-sensitive proteins. It is estimated that approximately 1-2% of the O₂ that enters a plant will either consist of or be converted to ROS [Bhattacharjee, 2005]. On the micro-scale, many cellular processes also produce local concentrations of hydroxide and peroxide as reaction byproducts or for use in signaling pathways; without available containment and decontamination procedures, affected cells will likewise suffer the effects of widespread oxidative disruption and damage.

Fortunately, cells have evolved a number of different systems by which they control the relative levels of oxidized and reduced species present at any given time;

these processes are collectively referred to as redox homeostasis. In plants, three major molecules - NADPH, ascorbate, and glutathione - along with several secondary molecules, such as NADH, flavins, and quinones, are responsible for maintaining an appropriate redox state (Figure 1). In solution these compounds exist as redox couples with a fluctuating ratio of reduced to oxidized molecules. When local cellular conditions grow too oxidized, these compounds will react with the excess ROS in order to drive the system back towards homeostasis. In order to then restore the compound's redox equilibrium, additional reduced molecules will need to be synthesized or the oxidized molecules will be reduced via a compound-specific regeneration cycle.

Although all of the major redox couples in plants are capable of buffering against oxidative conditions, each compound also fulfills a more specialized redox-related role. NAD(P)H provides essential reducing power for numerous enzymatic reactions including those associated with photosynthesis; among the major redox couples it has the largest (most negative) midpoint reducing potential: -320mV at pH 7.0 [Noctor, 2006]. Ascorbate, as the smallest of the three major redox couples, can traverse the plasma membrane to assist in detoxification of the apoplasm. It also can be stored in an unconjugated form in the vacuole at relatively high concentrations for future use. Finally, glutathione, the only major redox player to contain a thiol group, acts as a sulfur sink and cysteine storage molecule. By virtue of its thiol group, glutathione is also highly effective at maintaining protein thiol-redox state through the modification of disulfide bonds, as discussed below.

PART I - HOMOGLUTATHIONE SYNTHETASE AND GLUTATHIONE BIOSYNTHESIS

Glutathione (GSH) is composed of glutamate, cysteine, and glycine, with a γ -linkage between glutamate and cysteine rendering the peptide immune to degradation by all but one class of protease, the γ -glutamyl transpeptidases (Figure 2). As described above, glutathione can be found in both a reduced (GSH) and an oxidized (GSSG) form, with the reduced form favored over the oxidized form by up to 200-fold [Noctor, 2006; Masip, et. al, 2006]. Due to multiple polar groups, glutathione is exceedingly soluble, and can be found at a foliar concentration of ~1 to 5mM, with local concentrations of 7 to 20mM reported [Mullineaux and Rausch, 2005]. These traits, in combination with a relatively high reduction potential (-240 mV at pH 7.0), make glutathione both effective and highly adaptable as a modulator of redox homeostasis [Rouhier, 2008]. Not only can glutathione spontaneously detoxify reactive oxygen and nitrogen species, but it also can directly protect proteins against irreversible oxidation through glutathionylation of critical residues [Gallogly and Mieyal, 2007]. Additional roles for glutathione include detoxification of peroxides through the ascorbate-glutathione cycle, conversion of toxic aldehydes like formaldehyde and methylglyoxyl to less harmful variants, and sequestration of toxic heavy metals such as cadmium [Potters et al., 2002; Dixon et. al, 1998; Rauser, 1995; Skipsey et. al, 2000].

In the event of protein glutathionylation or the formation of an undesirable disulfide bond, the disulfide-oxidoreductase glutaredoxin is capable, with the assistance

of two GSH molecules, of reducing the residue moiety via either a monothiol or a dithiol mechanism [Rouhier et. al, 2008]. This activity results in the formation of one GSSG molecule, which then must be reduced to maintain the buffering capacity of glutathione. Another oxidoreductase, glutathione reductase, utilizes the greater reducing potential of NADPH to convert GSSG back to GSH, producing NADP⁺ in the process [Gill and Tuteja, 2010]. Finally, the regeneration cycle is completed when NADP re-enters the chloroplast stroma, and is reduced to NADPH by ferredoxin-NADP⁺ reductase as part of the photosynthetic Z-scheme.

If additional or replacement reducing power is required by the plant cell, more glutathione may be synthesized in two ATP-dependent steps. In the first reaction, the enzyme glutamate cysteine ligase (GCL; also known as γ -glutamylcysteine synthetase; EC 6.3.2.2) catalyses the formation of γ -glutamylcysteine from glutamate and cysteine. In the second reaction the enzyme glutathione synthetase (GS; EC 6.3.2.3) adds glycine to γ -glutamylcysteine to produce the complete tripeptide. In *Arabidopsis thaliana*, GCL contains a chloroplast-localization sequence and is expressed solely in the plastid, while GS is primarily cytosolic [Meyer and Hell, 2005]. Expression analysis indicates that of the two enzymes, GCL functions as the rate-limiting step and is the target of multiple regulatory controls [Foyer et al., 1995; Arisi et al., 1997].

Because cellular glutathione levels can impact so many different facets of a plant's stress response, GCL activity remains at all times tightly regulated via at least three major mechanisms. 1.) At the substrate level, GCL activity is limited by the availability of glutamate and cysteine, with the latter being the more limiting of the two.

Because cysteine is derived from serine and acetyl-CoA, its use as a substrate for GCL drains from carbon reserves that could be instead spent to generate proteins and/or ATP [Youssefian et al., 2001]. Accordingly, in order to minimize the perturbations of other pathways, only those resources that are absolutely needed are drawn off in order to synthesize glutathione. 2.) At the level of redox control, GCL is regulated by a unique mechanism that was only recently deduced. It has been known for nearly twenty years that mammalian GCL is heterodimer composed of a larger (MW 70,000 kDa), catalytic subunit and a small (MW 30,000 kDa) regulatory subunit. Under reducing conditions and in the absence of the regulatory subunit, catalytic activity is only a fraction of that found when GCL is present as a holoenzyme [Huang et al, 1993a; Huang et al., 1993b, Chen et al., 2005]. This data in conjunction with more recent mutagenesis studies led to the conclusion that formation of a reversible disulfide bond between the catalytic subunit and the regulatory subunit prompts GCL activation [Fraser et al., 2003]. But while the human GCL has a regulatory subunit, no comparable protein subunit has been found to be encoded by a plant genome. Instead, a combination of mutagenic, kinetic, and crystallographic studies have shown that plant GCL is a homodimer whose activation and subsequent dimerization under oxidative conditions is controlled by a pair (one on each monomer) of disulfide bonds [Hicks et al., 2007] (Figure 3). Additionally, access of substrates to the active site is controlled by a second disulfide bond; only under sufficiently oxidative conditions does the β -hairpin flap over the active site entrance swing back to allow uninhibited access [Hothorn et al, 2006]. 3.) Finally, at the level of expression, transcription of the gene encoding GCL has been shown to increase when

plants are subjected to known sources or signals of oxidative stress, including jasmonic acid and the heavy metals copper and cadmium [Xiang and Oliver, 1998]. Although the human GCL promoter has been well characterized, little comparable work has been done in plants [Soltaninassab et al., 2000]. The Arabidopsis GCL promoter does contain several possible G-box elements similar to those found in the jasmonic acid sensing portion of the *Pin2* promoter, but no further studies have been done to verify their role [Xiang and Oliver, 1998]. With additional evidence also suggesting that GCL activity is regulated by light intensity and/or phosphorylation, (the latter being documented for the human GCL variant), it is clear that the enzyme requires still a great deal of further investigation [Ogawa et al., 2004; Sun et al., 1996].

While there remain many unanswered questions with regards to GCL, the second step in the glutathione synthesis pathway presents an entirely different set of unknowns. In most plants, γ -glutamylcysteine is converted to glutathione through the activity of GS. Like GCL, eukaryotic GS is catalytically active as a dimer; however, in *Escherichia coli*, and other prokaryotes, GS functions as a tetramer. To date, three x-ray crystal structures of the GS from *Saccharomyces cerevisiae*, *E. coli*, and *Homo sapiens*, as well as a "loopless" *E. coli* variant structure, have been solved [Yamaguchi et. al, 1993; Kato et. al, 1994; Polekhina et. al, 1999; Gogos and Shapiro, 2002]. These structures indicate that GS falls within the large ATP-grasp superfamily of protein structures [Galperin and Koonin, 1997]. Members of this protein family are defined by the presence of two sets of two anti-parallel β -sheets connected by a series of loops. This motif, known as an ATP-grasp or palmate-grasp for its provision of an ATP-binding pocket, comprises the

majority of the active site in proteins of this family, and also provides the family with its name. In addition to the shared motif, members of the ATP-grasp superfamily also rely on a similar mode of action: that is, the ATP-dependent ligation of a carboxyl group carbon from one reactant with the amino (or imino) group nitrogen of a second reactant via the formation of an acyl-phosphate intermediate [Meister and Anderson, 1983; Ogita and Knowles, 1988; Meister 1989; Fan et. al, 1995]. For the GS-catalyzed reaction, γ -glutamylcysteine provides the carboxyl group, while glycine provides the amino group. With regard to enzyme kinetics, the reaction likely proceeds via a random terreactant mechanism, with slight preference given to the order of substrate binding [Jez and Cahoon, 2004]. As an interesting side note, although members of the ATP-grasp superfamily are structurally and mechanistically similar, their amino acid sequences are quite divergent and range only from 10%-20% identical across the family [Galperin and Koonin, 1997]. Other examples of ATP-grasp family members include D-Ala:D-Ala ligase, biotin carboxylase, and carbamoyl phosphate synthase [Fan et. al, 1994; Artymiuk et. al, 1996, Thoden et. al, 1997].

Among the branches of life, synthesis of glutathione as a storage agent and redox buffer is extremely well conserved. Virtually all eukaryotes produce glutathione, as do many bacteria. Only among the archaea is glutathione synthesis sharply limited, with just halobacteria known to produce it. Despite its time-tested effectiveness, some eukaryotic parasites (such as those of the genus *Trypanosoma*), do not produce glutathione at all, while many plant species produce it in addition to one or more glutathione homologs [Muller et al., 2003]. For example, in many leguminous plants homoglutathione – in which

the terminal glycine has been replaced by β -alanine, is found with or in place of glutathione in a tissue-dependent fashion [Moran et al, 2000]. In members of the Poaceae, except maize, a tripeptide called hydroxymethylglutathione, which contains a terminal serine instead of glycine, is produced [Klapheck et al, 1994]. In maize, yet another glutathione-like peptide is synthesized, this one with a terminal glutamate [Meuwly and Rauser, 1992]. Of these alternate forms of glutathione, the most is known about homoglutathione. Except for broadbean and lupine, every legume that has been thus far investigated produces homoglutathione in addition to glutathione [Moran et al., 2000]. In some tissues, such as alfalfa leaves, homoglutathione completely replaces glutathione as the dominant thiol compound, while in others they may be found more or less one-for-one. By contrast, in cowpea leaves glutathione is the dominant tripeptide and homoglutathione is almost completely absent [Matamoros et al., 1999]. In soybean, both homoglutathione and glutathione are present, with leaves and seeds containing 50- to 200-fold and 135-fold more homoglutathione than glutathione, respectively. [Klapheck, 1988; Matamoros et al., 1999].

For synthesis of both glutathione and homoglutathione, the first reaction - synthesis of γ -glutamylcysteine - is shared. However, while glutathione synthesis requires the activity of GS to add a terminal glycine to the tripeptide, homoglutathione synthesis relies on the homologous enzyme homoglutathione synthetase (hGS) for introduction of β -alanine (Figure 4). Because the genomes of numerous legumes, soybean included, show evidence for two rounds of genome duplication, it has been proposed that hGS arose from GS by divergent evolution after the first duplication event

[Shoemaker et al.,2006; Van et al., 2008; Gill et al., 2009, Frendo et. al, 2001]. In the soybean genome, as an example, there are two copies each of the genes encoding GS and hGS, with each pair sharing 87% and 93% sequence identity, respectively. The relative expression patterns of the various copies remain unknown.

Between legumous GS and hGS genes, the sequence identity is ~60-70% depending on the species assessed [Frendo et. al, 2001]. Given such a high degree of identity, it is curious that hGS has managed to evolve a unique, if parallel, function to GS. From a redox perspective, the benefit of synthesizing homoglutatione instead of, or in addition to, glutathione is unclear. Both molecules preserve the redox-reactive cysteine residue that allows conversion from GSH to GSSG; in fact the only obvious difference is that homoglutatione is effectively one carbon bond-length longer than glutathione courtesy of β -alanine. But while the difference in substrate size does not shed light on a defined role for homoglutatione, it does hint at structural differences that have arisen between hGS and GS. Based on the available GS structures, γ -glutamylcysteine occupies a pocket at one end of the cleft formed by the ATP-grasp motif, while ATP and glycine occupy the opposing end [Yamaguchi et. al, 1993; Polekhina et. al, 1999; Gogos and Shapiro, 2002]. The carboxyl tail of glycine contacts two Ala residues (Ala 462 and Ala 463 in the human GS), that are part of a larger alanine-rich loop domain. Based on a comparison of the yeast apoenzyme and ATP/ γ -glutamylcysteine-containing GS structures, it is evident that this domain is capable of movement during the overall reaction cycle [Gogos and Shapiro, 2002]. However, kinetic studies indicate that AtGS will not accept β -alanine as a substrate, which thus implies that any domain movement

that occurs is simply not enough to accommodate the longer β -alanine molecule in the glycine binding pocket [Galant et. al, 2009, Jez and Cahoon, 2004].

In addition to the previously mentioned alanine-rich loop, a second domain of the GS enzyme appears mobile based upon the yeast crystal structure [Gogos and Shapiro, 2002]. This domain, termed the “lid domain” according to the nomenclature associated with the human GS structure, is composed of residues 355-417 of the yeast enzyme and makes up one corner of the roughly triangular-shaped overall GS structure. Within the lid domain is a subdomain, known as the glycine-rich loop, that appears to make extensive contacts with the bound ATP moiety. Based on the apoenzyme and reactant-bound yeast GS structures, it appears that this domain’s purpose is to swing inward and lock ATP in place following its binding within the ATP-grasp cleft. Investigation into the kinetic mechanism of AtGS has indicated that the GS enzymes favor a semi-specific binding order for their three substrates: namely, either ATP or γ -glutamylcysteine first, and glycine (or an appropriate homolog) last [Jez and Cahoon, 2004]. Accordingly, the yeast structure bound with γ -glutamylcysteine and an ATP homolog represents the third stage in the reaction mechanism, with the apoenzyme and enzyme bound with either ATP or γ -glutamylcysteine representing the first and second stages, respectively. Among the yeast, *E. coli*, and human GS structures, no structure for the γ -glutamylcysteine-bound variant of the second stage exists; thus it is unclear if any movement of the lid domain is prompted by the binding of γ -glutamylcysteine alone. Furthermore, while cumulatively the available crystal structures provide a visual representation of the reaction mechanism, snapshots of more than two stages are not available for a single enzyme, leading to

difficulties in the comparison of domains of different sizes and of different numbering schemes across the various GS structures.

In order to fill in the gap in the GS-type reaction mechanism as well as provide a series of snapshots from a single enzyme, I solved three crystal structures of the homogluthathione synthetase from *Glycine max*. The three structures are of the apoenzyme, the enzyme with γ -glutamylcysteine bound, and the enzyme with ADP and homogluthathione bound, and represent respectively the first, second, and fifth stages of the reaction mechanism. This series of structures, along with their accompanying mutagenesis and kinetics data, identify the structural elements that are responsible for the differences in substrate specificity between GS and hGS. For more information, please refer to Chapter 2.

PART II - THE THIOL-REDOX PROTEOME - BUTHIONINE SULFOXIMINE (BSO), H_2O_2 , AND *BRASSICA JUNCEA* (INDIAN MUSTARD)

The example of GCL from Part 1 illustrates that in some cases, the formation of disulfide bonds is desirable as a means of controlling enzyme activity. In plants, this type of redox regulation is relatively rare, not because few proteins utilize it, but rather because only a handful of redox-sensing candidates have been appropriately characterized. The majority of known redox-regulated proteins are controlled by one of a half-dozen or so different potential-sensing pairs, among them thioredoxin/ferredoxin, glutathione/glutaredoxin, and NADP/thioredoxin [Buchanan and Balmer, 2005]. The

protein components of these redox pairs - namely ferredoxin, thioredoxin, and glutaredoxin - all utilize a reduction mechanism that necessitates direct contact between them and the target protein that they are reducing. Accordingly, screens to identify potential targets of these regulatory proteins have typically been able to employ affinity chromatography followed by mass spectrometry or N-terminal sequencing; this method only pulls out proteins that strongly interact with the “bait”, with no regard for the specific means of redox regulation (disulfide bond, glutathionylation, etc.) utilized by the individual protein [Balmer et al., 2003; Motohashi et al., 2001; Yano et al., 2001]. Currently, the total number of identified plant redox-regulated proteins identified via affinity chromatography stands at several hundred, with the largest subsets shown to interact with thioredoxin and glutaredoxin, respectively [Buchanan and Balmer, 2005, Hisabori et al., 2005, Rouhier et al., 2005, Wormuth et al., 2007]. To date, very few proteins known to use disulfide bonds to regulate their activity have been identified via this methodology. Besides GCL, one example is NPR1, which was originally studied because of its essential role in plant systemic acquired resistance (SAR). Under normal conditions, NPR1 exists as an inactive oligomer; however, reduction of intermolecular disulfide bonds between the subunits following SAR initiation allows the protein to achieve its active monomeric form [Mou et al., 2003]. Another example is OxyR, a peroxide-sensing transcription factor from *E. coli* and *Salmonella typhimurium* that is activated through the formation of an intramolecular disulfide bond and deactivated via the activity of glutaredoxin [Zheng et al., 1998, Christman et al., 1989].

While affinity-based methodologies are well established for isolating certain subsets of the redox-regulated proteome, other more inclusive methodologies exist as well. One alternative is a two-dimensional (2D)-gel electrophoresis approach that allows all proteins within a given sample to be separated and fixed. Most 2D-gels rely on isoelectric point for separation in the first dimension, and denatured molecular weight in the second dimension, though native molecular weight may also be used [O'Farrell, 1975]. In order to utilize 2D separation for the detection of redox-sensitive proteins, the protein mixture under scrutiny must first be treated so as to distinguish redox-labile proteins from the remainder of the proteins in the sample. In a complex protein sample, many proteins will have solvent-exposed cysteine residues. While some of these residues may form disulfide bonds or contain thiol modifications such as glutathionylation, others will be present as free thiol groups that will not and will never be modified naturally. Because this latter class of cysteine residues is not redox-labile, it is important that they be chemically blocked to prevent reactions during the subsequent detection of modified and disulfide-bound cysteines. Thus, a soluble protein extract must first be treated with a thiol alkylating agent such as *N*-ethylmaleimide (NEM), iodoacetamide (IAM) or iodoacetic acid (IAA). Although all of these compounds form an adduct that is almost impossible to reverse, NEM is perhaps the best choice based on faster reaction speed and activity under a wider pH range relative to the other two compounds [Rogers et al., 2006].

Once the non-labile thiols in a complex protein mixture have been blocked, the next step is to detect the redox-reactive cysteine residues. Because these thiols are either

involved in disulfide bonds or are blocked by secondary modifications, the protein mixture must first be treated with a reducing agent such as dithiothreitol (DTT) or tris[2-carboxyethyl]phosphine (TCEP) so as to reduce the thiol groups. Because the previous reaction of free thiols with NEM was done via alkylation as opposed to thiol-disulfide exchange, reducing agents will have no effect on those adducts. With reduction of the previously oxidized thiols complete, they become available for chemical modification via one of several fluorescent compounds. The choice of which fluorescent compound to use depends upon the emission wavelength required, the pH of the protein mixture to be labeled, and the overall experimental design. Popular thiol-reactive dyes include those of the bromobimane family (most commonly monobromobimane), iodoacetamide-fluorescein conjugates (typically 5'-iodoacetamidofluorescein - IAF), and the cyanine dyes (Cy2, Cy3, and Cy5) [Timms, 2005; Fahey and Newton, 1987; Baty et al., 2002; Chen et al., 2008].

When labeling of the complex protein mixture is complete, the protein sample may be loaded onto a 2D gel and separated as previously described. Because the protein spots will not be visible to the naked eye, the resulting gel(s) must be imaged at the appropriate wavelength for whichever fluorescent dye was selected in order to see the spots that contain redox-reactive proteins. For comparison of the relative numbers of redox-reactive proteins to total proteins across a given pI/mW range, the gels may then be further stained with a total protein dye and imaged again. There are many sensitive dyes, including SYPRO Ruby, SYPRO Tangerine, and FOCUS FASTsilver, available; however, care must be taken to ensure that the dye chosen does not bind covalently (as

this may interfere with downstream mass spectrometry applications) and that its excitation and emission wavelengths do not overlap with those of the chosen thiol-reactive dye.

Depending on the complexity of the protein sample under scrutiny, several different methods for identifying individual spots from the 2D gels may be available. If a protein is particularly well characterized, whole protein extraction and verification of retention time via HPLC may be all that is required. In most cases however, the protein identification will not be so easy, and a mass spectrometry-based approach will be necessary. For this method, the protein will need to be digested with a predictable protease such as trypsin, and the resulting peptides separated via reverse-phase liquid chromatography. Those peptides will then further fragmented by MS/MS, and the resulting spectra will be identified via comparison against a known database (NCBI, SWISSPROT, etc.) of protein sequences.

As stated previously, the 2D gel-based approach, although more time consuming, presents an advantage over related affinity chromatography methodologies because it allows for the detection of proteins that do not react strongly with a “doxin” and/or contain one or more redox-labile disulfide bonds. In plants as well as other eukaryotes, a number of efforts have already been undertaken in order to identify novel redox-reactive proteins using 2D-SDS-PAGE. In Alvarez et al. [2009a], the authors dissected shoot tissue from *Arabidopsis thaliana* seedlings, and labelled with iodoacetamide and mBBBr before separating the protein in two dimensions. Fifty resulting proteins were then identified via nano-LC-MS/MS as redox-reactive; five proteins were further identified as

new members of the thiol-redox proteome. In an effort to tease out the relationship between redox-regulation and dormancy control, another group isolated protein from hormone-treated wheat seeds, labeled with mBBr and iodoacetamide (the opposite order from above), and performed nano-LC-MS/MS and MALDI MS after 2D-gel separation. Their experiments resulted in the identification of 79 unique redox-modifiable proteins with possible roles in seed dormancy [Bykova et al., 2011]. Across the literature, there exist other example of studies seeking to identify redox-labile plant proteins under a given set of conditions [Zhou et. al, 2011; Tanou et al, 2010; Alvarez et al., 2009b]; Maeda et al., 2005; Rinalducci et al., 2008].

Importantly, compared to the sheer volume of redox-related studies in animal or microbial systems, the investigation of redox-reactive proteins in plants is still in its infancy. The majority of studies in plants to date have been furthermore very narrow in focus, concentrating solely on the effect of a specific compound or growth condition on the relative abundance or redox state of proteins. While such investigations are necessary to elucidate the mode of action of the relevant plant-response networks, they do not address what is happening between the stimuli and the protein that is changing. Namely, how do specific proteins respond to the ROS that serve as antagonizers and/or signaling molecules in response to a set change in conditions?

To begin to address this question, I have undertaken a series of experiments designed to address how different sources of ROS differentially antagonize a redox-responsive system. Using *Brassica juncea* (Indian mustard), I treated the roots of potted plants with buthionine sulfoximine (BSO), hydrogen peroxide, or water (as a control).

Because BSO, as an inhibitor of GCL, is an indirect source of endogenous ROS [Griffith and Meister, 1979], and hydrogen peroxide is a direct source of exogenous ROS, I anticipated that they would affect the expression and redox state of different, though possibly overlapping, sets of proteins. Furthermore, because only a relatively small body of work on redox-responsive proteins has originated in field of plant biology, I hoped that this series of experiments would add to the available body of knowledge, particularly if they allowed for the identification of new targets of redox-regulation. A more detailed description of the methodology and results of these preliminary 2D-gel experiments is presented in Chapter 3.

PART III - THE THIOL-REDOX PROTEOME - OZONE AND *GLYCINE MAX* (SOYBEAN)

The use of *Arabidopsis thaliana* as a model system began as early as 1907, when Strasburger and his student Laibach suggested its value for studying chromosomes. In the mid-forties, the development of *Arabidopsis* as a platform for mutagenesis began, and by the 1970s the plant and its close relatives had been widely adopted by biology labs around the world [www.arabidopsis.org (TAIR); Redei, 1975]. *Arabidopsis* is an excellent model system for a number of reasons, including its fast growth rate, genome plasticity, and extensive family tree. While some members of the Brassicales are widely cultivated, *Arabidopsis* itself has no agricultural significance. And while ostensibly the plant and its cousins occur naturally throughout much of Europe and Asia, the modern

native habitat of *Arabidopsis* is very much a petri dish in a lab. Thus, when it comes time to study how a plant responds to and interacts with real world oxidative conditions and stimuli, *Arabidopsis* and its brethren are not necessarily the best choice for further investigation. But in order to select an appropriate plant, one must first understand the breadth and scale of the challenge that oxidative damage causes in a more natural environment.

As described briefly at the beginning of the introduction, one of the largest sources of naturally occurring ROS stems from the conversion of O_2 to various oxide radicals. While the availability of O_2 does not change significantly, the relative concentration of another ROS-producing molecule - ozone - is increasing. In its simplest state, ozone is composed of three charge-stable oxygen atoms; the chemistry that gives rise to it, however, is somewhat complicated and also depends upon where the ozone is being produced. In the stratosphere, ultraviolet energy in the form of a photon can split O_2 to yield monoatomic oxygen. Monoatomic oxygen is highly unstable, so it rapidly recombines with O_2 to yield O_3 . In the troposphere, ozone production begins when carbon monoxide reacts with hydroxide, yielding a proton and carbon dioxide. The proton then further reacts with O_2 to produce the peroxy radical HO_2 . HO_2 is also unstable, and will react quickly with any number of non-methane volatile organic chemicals (NMVOCs); their products will then react further with ultraviolet energy to produce monoatomic oxygen and subsequently ozone as previously described [Tang et al., 2011; Renaut et al., 2008] (Figure 5). Unlike stratospheric ozone production, the production of ozone in the troposphere is dependent upon the availability of NMVOCs;

these compounds may include nitrogen oxides, sulfur oxides, terpenes, and assorted aqueous solvents. Many NMVOCS are released into the atmosphere as part of the waste streams from various industrial processes; others are byproducts of the combustion of gasoline and diesel in vehicle engines. Unfortunately, as both global averaged industrial output and vehicle ownership, largely as a result of economic development in China and Southeast Asia, are rising and predicted to continue doing so, the available global tropospheric concentrations of NMVOCS, and in turn ozone, are likely to increase as well [Monks et al., 2009; van Aardenne et al., 2001; Fu et al., 2007; Meagher et al., 1998].

Historical data indicates that prior to the industrial revolution, tropospheric ozone concentrations in the northern hemisphere were quite low, averaging only 11 ppb (parts per billion) with deviation of 5 ppb depending on the season. Even as the pace of industrialization increased during the first half of the 19th century, ozone concentrations in the northern hemisphere remained modest at 15 ppb [Volz and Kley, 1988]. However, during the period between 1950 and 1980, ozone concentrations began to trend upward by approximately 0.35 ppb per year, and by the mid-1980s were increasing by up to 0.5 ppb per year [Tang et al., 2011; Fuhrer, 2009; Cooper et al., 2010; Hudman et al., 2008]. Perhaps by virtue of a balance in decreasing and increasing emissions between developed and developing countries respectively, the present day rate of increase is holding steady at approximately 1-2% of the ambient concentration per year [Morgan et al., 2006; Chameides et al., 1994]. Currently, the average annual ambient tropospheric ozone concentration ranges from 20 to 45 ppb over the mid-latitudes of the Northern

Hemisphere [Vingarzan, 2004; Booker et al., 2009]. As ozone synthesis is dependent upon available energy levels, ozone concentrations tend to follow a cyclical cycle throughout the year. Thus, during the summer months (June-August depending upon latitude), local ozone concentrations may peak at an average of 60-80 ppb before tapering off again in the fall [Fowler et al., 1999; Mauzerall et al., 2000]. Furthermore, due to the earth's natural light-dark cycle, ozone concentrations are also diurnally cyclical, with the highest concentrations coinciding with the brightest/hottest parts of the day [Fuhrer et al., 1997].

Because both historical trends and present-day atmospheric profiling data support a situation in which tropospheric ozone concentrations will continue to rise over the course of the next 50-100 years, many different environmental models have been put forth to help predict likely ozone concentrations and the areas that might be most affected. One model, taking into account biomass emissions and emissions legislation, predicts that India and southeast Asia, including southern China, will see large increases (9-11 ppb) in surface ozone concentrations between now and 2030 while concentrations over North America remain steady [van Dingenen et al., 2009]. Another prediction, which averages the results of 10 different modeled scenarios, indicates that the Middle East, India, and China will see summertime concentrations rise by 45-55 ppb, and that the southern/eastern United States and Mexico will see increases of 25-35 ppb by 2100 [Prather et al., 2003]. The results of many other modeled scenarios have been published, and while they tend to utilize different sets of baseline data and differing predictive criteria, the world regions that are highlighted as being under threat from rising tropospheric ozone concentrations - namely China, India, and the eastern United States -

remain largely consistent [Murazaki and Hess, 2006; Liao et al., 2006; Bell et al., 2007; Ebi and McGregor, 2008; Racherla and Adams, 2006; Nolte et al., 2008]. This is problematic for several reasons. From a human health perspective, high tropospheric ozone concentrations are dangerous not only because they can lead to tissue oxidation and irreversible damage, but also because ozone is one of the primary components of smog. Because ozone is denser than air, it can trap pollutants close to the earth's surface, leading to a variety of respiratory problems. Local geography and population density can further exacerbate the problem; images of a smog-filled Los Angeles, which sits in a natural depression, and Beijing, which is home to more than 12 million people, have been etched into the public consciousness in recent years. Since China and India together currently account for roughly 45% of the world's population (a percentage that continues to rise every year), increasing ozone concentrations in those regions could be particularly catastrophic for the health of large numbers of people in the future [CIA World Factbook, <https://www.cia.gov/library/publications/the-world-factbook>].

Another reason why the predicted increases in ozone concentrations over China, India, and the United States are problematic stems from the ability of these countries to produce large volumes of crops for home use and export. While crop production of course partially correlates with country size (the United States, China, and India are ranked 3rd, 4th, and 7th respectively by total land area), these three countries are nonetheless consistently the top producers of many different grains and legumes including rice, maize, millet, soybeans, and wheat [FAOSTAT]. For these crops and many others, ozone exposure, much as it does for humans, can cause widespread

oxidative damage to essential tissues, which in turn negatively affects crop yield. From a food security perspective, it is easy to predict how widespread devastation of harvest yield as a result of oxidative damage (coupled of course to ongoing disasters such as droughts and floods) could rapidly lead to food shortages and famine in highly populated countries without implementation of preventive measures.

How might a country, and the world at large preemptively avoid widespread crop loss through oxidative damage? The most obvious option: implementation of widespread limitations to further industrial emissions, is both politically difficult and slow to yield fruitful results. Many crops, particularly those grown in regions where summertime ozone averages top 60 ppb, are currently suffering from oxidative damage and producing reduced yields. The second option: the utilization of ozone-resistant crop strains, is arguably much cheaper and politically favorable; however, there is a problem. Because high tropospheric ozone concentrations are a relatively new phenomenon there are very few cultivars, natural or engineered, which demonstrate ozone-tolerance or resistance. While a few crop species - namely plums and strawberries - seem to be partially resistant to the effects of ozone exposure, most of the major grain and legume crops show moderate to severe sensitivity [Mills et al., 2007]. Among those major crops, soybean is the most sensitive to ozone. Across all assayed soybean cultivars, the relationship between seed yield and seasonal daytime ozone concentration is largely linear, with ozone concentration in ppb inversely proportional to seed yield in kg/hectare [Betzberger et al., 2010]. At concentrations as low as 40 ppb, soybean growth and seed yield begins to decrease; increasing the atmospheric concentration to 70 ppb accordingly

results in yield losses of 11-36% [Morgan et al., 2006; Emberson et al., 2009; Heck et al.; 1983; Heagle et al., 1998].

While numerous transcriptomic, metabolomic, and proteomic studies have been undertaken in both soybeans and other crop species to elucidate the mechanism by which ozone exposure negatively impacts yield, a clear answer has yet to emerge. One possibility is that irreversible oxidation of key proteins, particularly those involved in photosynthesis, forces the plant to shift resources destined for starch storage and/or cell division toward supplemental amino acid and protein synthesis. Another possibility is that affected plants utilize their resources to upregulate ROS scavenging pathways with the hope of maintaining the status quo. In either case, one would expect that exposure to similar ozone concentrations would upregulate similar response pathways in the various affected crop species. However, there has been very little consensus across crops as to the transcripts and proteins that are differentially expressed following ozone exposure [Ahsan et al., 2010; Agrawal et al., 2002; Feng et al., 2002; Bohler et al., 2007; Bagard et al., 2008; Cho et al., 2008; Sarkar et al., 2010; Torres et al., 2007; Tosti et al., 2006; Gadjev et al., 2006]. This suggests that either the response mechanisms vary significantly between different crops (not an unexpected conclusion given the evolutionary distance between monocots, dicots, etc.), or that plants are very sensitive to variations in the sets of exposure conditions utilized across the various experiments. The limited availability of evidence to support either hypothesis indicates that, in order to shed further light on ozone response pathways and jumpstart the development of ozone-resistant crop cultivars, further experimentation is necessary.

In order to identify novel proteins which are differentially regulated and/or differentially expressed in response to elevated tropospheric ozone concentrations, I have undertaken a series of proteomics-based experiments based upon the methodology developed and described in Chapter 3. Instead of utilizing *Arabidopsis* or *Brassica juncea* (as described previously) as my system of inquiry, I opted to conduct my investigation with soybeans, for several reasons. First, although *Brassica juncea* is a minor crop plant, neither it nor *Arabidopsis* are widely cultivated, meaning - in terms of sheer scale - neither will greatly contribute to food insecurity due to increasing ozone. Soybeans, on the other hand, are one of the most widely bred crop species, particularly in regions of the world most at-risk from rising ozone concentrations. In the United States, the majority of soybean cultivation occurs in the upper Midwest - specifically in Illinois, Indiana, and Iowa, as well as in Nebraska - which is within the region where ozone concentrations are predicted to increase the most in the coming decades (USDA-NASS; Fishman et al., 2010). In addition to its localization and practical utility, the physiological response of soybeans to chronic ozone exposure has already been well characterized. For *Arabidopsis*, the majority of experiments have utilized acute exposure regimens which, while damaging, do not impart the same long-term effects as naturally-occurring chronic exposure cycles [Chen et al., 2009]. In order to maximize the value of physiological data imparted from chronic exposure experiments with soybeans, it ideally should be paired with more in-depth analysis of protein oxidative responses. Accordingly, in Chapter 4, I present the results of redox proteomics experiments

comparing protein expression and oxidation profiles in soybean tissue grown in the field under ambient and elevated chronic ozone concentrations.

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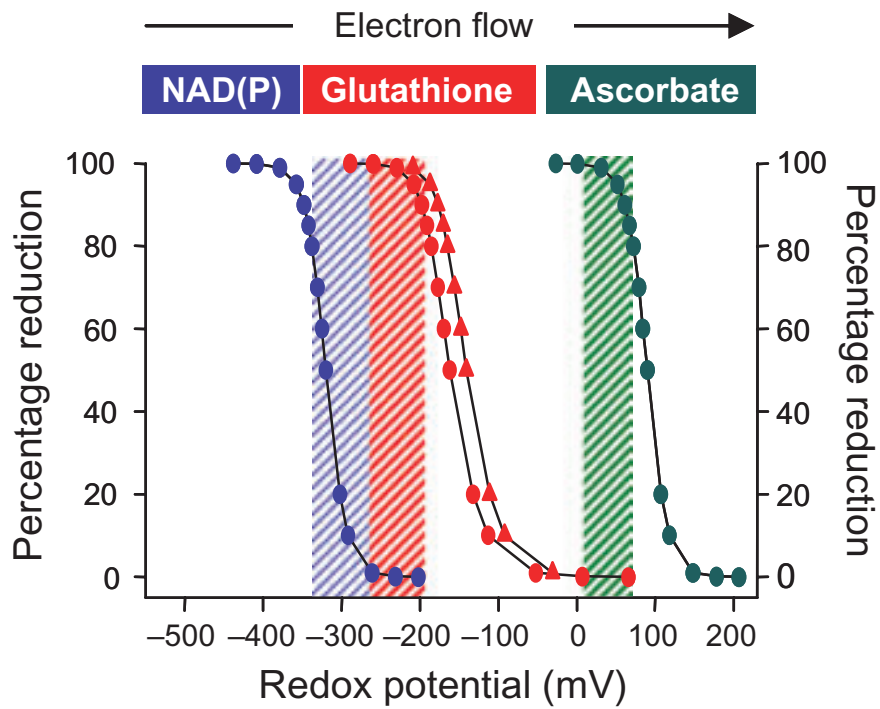


Figure 1. The three major redox couples found in plants. The relationship between percentage reduction and redox potential for each redox couple is shown.

From Noctor, 2006.

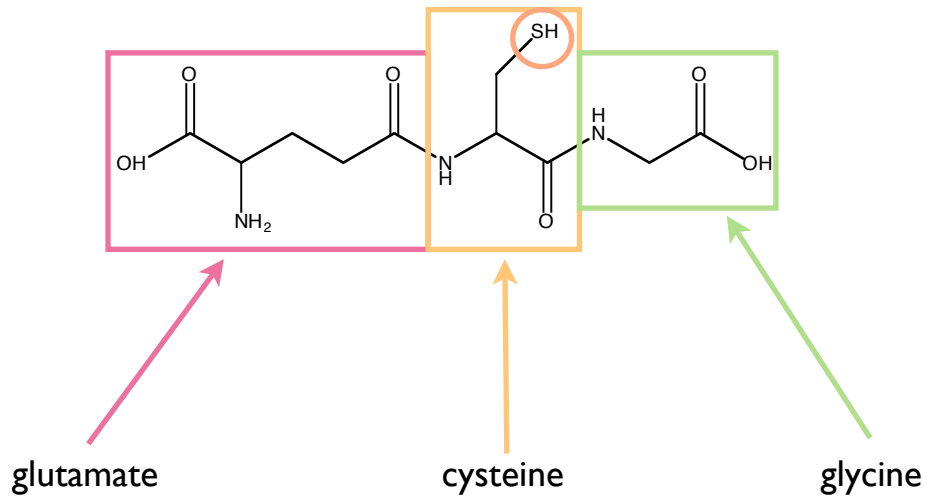


Figure 2. The tripeptide glutathione. Glutathione is composed of three amino acids: glutamate, cysteine, and glycine, as outlined. The critical sulfhydryl group that allows glutathione to act as an effective redox agent is circled.

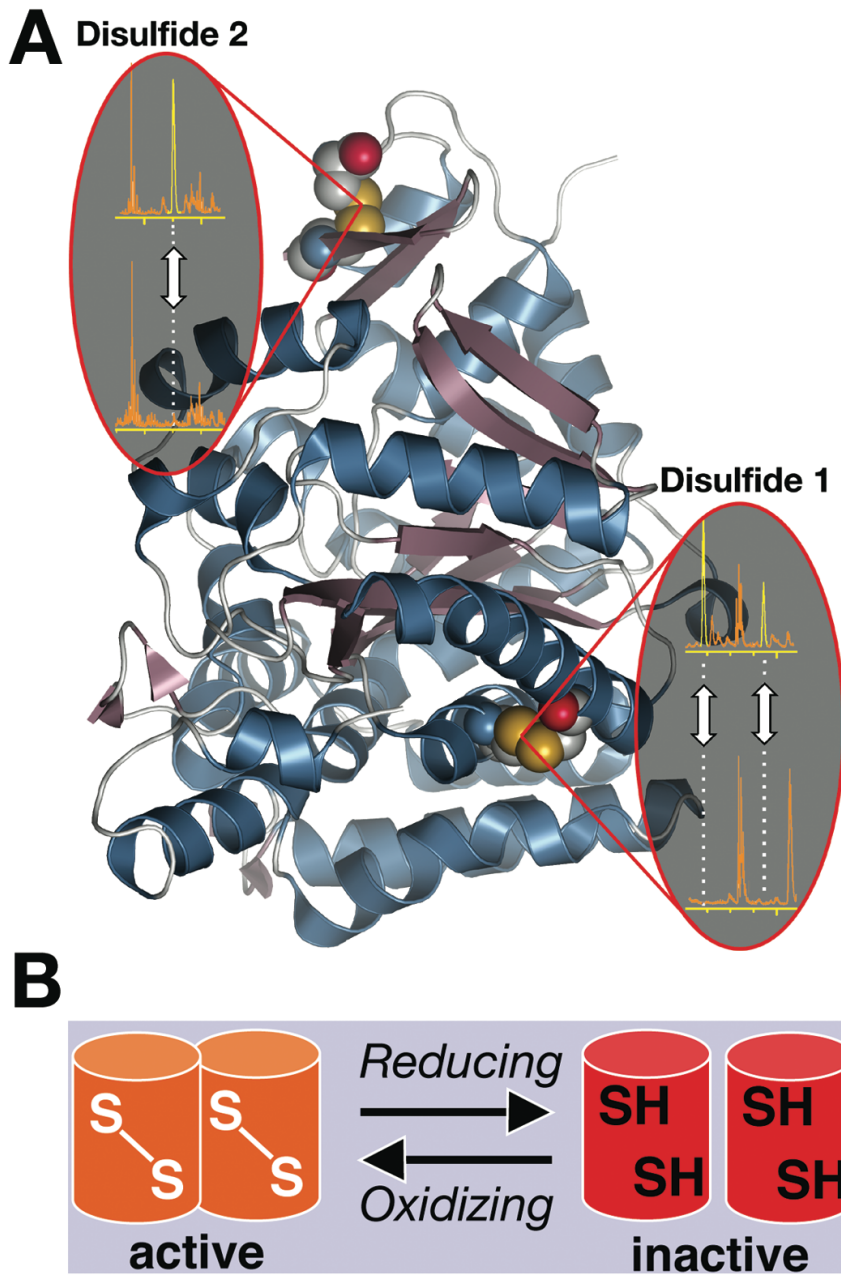


Figure 3. Redox regulation of plant GCL. A.) A monomer of the *Brassica juncea* (Indian Mustard) GCL enzyme with the locations and mass spectrometry profiles of the reduced versus oxidized disulfide bonds indicated. B.) Schematic detailing the transition from the inactive to active form of the GCL enzyme, and *vice versa*.

From Yi et al., 2010.

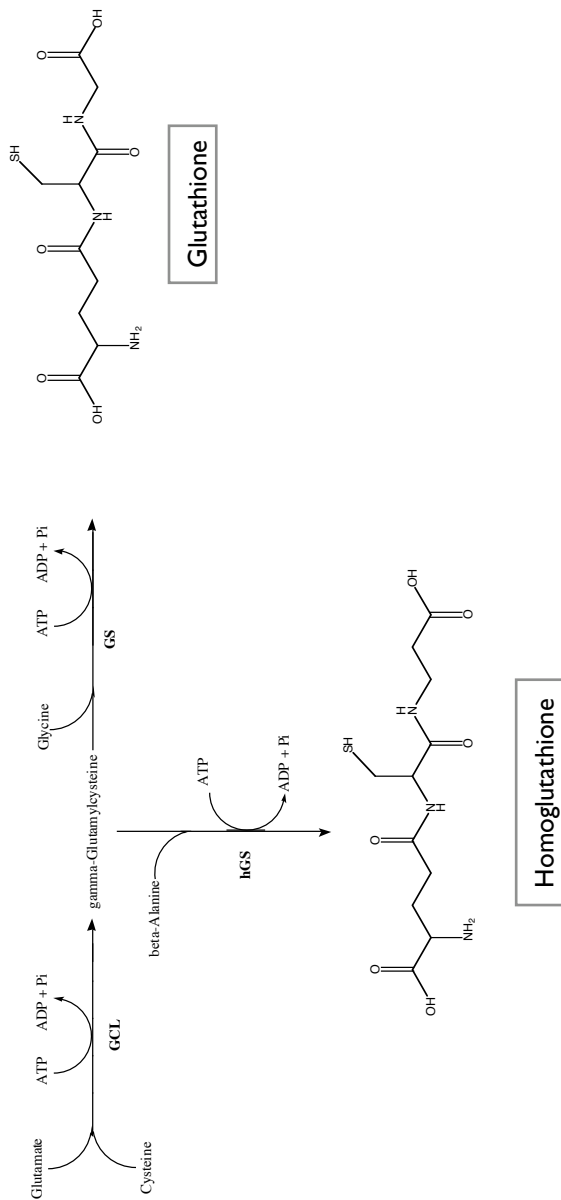


Figure 4. GSH and hGSH biosynthesis. hGSH and GSH are synthesized in two ATP-dependent steps. GCL activity is shared between the two pathways, while a committed enzyme (either GS or hGS) performs the second reaction.

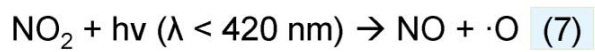
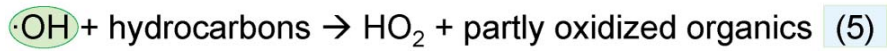
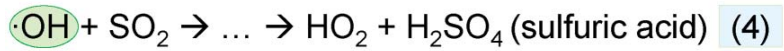
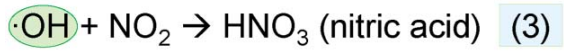
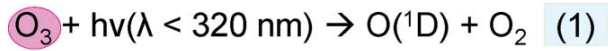


Figure 5. Key reactions for synthesizing ozone in the stratosphere and troposphere.

From Tang et al., 2011.

CHAPTER 2

STRUCTURAL BASIS FOR EVOLUTION OF PRODUCT DIVERSITY IN SOYBEAN GLUTATHIONE BIOSYNTHESIS

PREFACE

As described in the introduction, homoglutathione (hGSH) is not the only redox-labile glutathione homolog that is produced in plants. To date, three such tripeptides - hGSH (in legumes), hydroxymethylglutathione (in grasses), and gamma-glutamylcysteinylglutamate (in cadmium-stressed maize), have been isolated. Of these three, glutamylcysteinylglutamate - isolated in the early 1990s - is the most recent discovery; by the mid-90s both hGSH and hydroxymethylglutathione (hmGSH) had been under study for nearly a decade [Meuwly et al., 1993]. While the enzyme activity behind hGSH synthesis was rapidly identified, early hmGSH work focused on its interactions with alcohol dehydrogenase and reactive oxygen species (ROS) [Macnicol, 1987; Zopes et al., 1993; Martinez et al., 1996]. In 2002, carboxypeptidase Y was identified as catalyzing the synthesis of hmGSH *in vitro*; however, it remains unknown if this activity has any physiological significance [Okumura et al., 2003]. To date, the source of glutamylcysteinylglutamate remains unclear. Kinetic characterization has confirmed that the maize glutathione synthetase is not capable of using glutamic acid in place of glycine [Skipsey et al., 2005]. Thus, it is likely that unidentified enzyme is responsible for glutamylcysteinylglutamate synthesis. The chemical diversity of the GSH homologs suggests that the substrate specificity of the glutathione synthetase (GS)-related enzymes in these plants differs from the canonical GS, but the origin and exact role of these enzymes remains obscured.

Author Contributions: JMJ designed research; AG, KAJA, CZ, and REC performed research; AG, REC, and JMJ analyzed data; AG and JMJ wrote the paper.

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Structural Basis for Evolution of Product Diversity in Soybean Glutathione Biosynthesis^W

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The redox active peptide glutathione is ubiquitous in nature, but some plants also synthesize glutathione analogs in response to environmental stresses. To understand the evolution of chemical diversity in the closely related enzymes homogluthione synthetase (hGS) and glutathione synthetase (GS), we determined the structures of soybean (*Glycine max*) hGS in three states: apoenzyme, bound to γ -glutamylcysteine (γ EC), and with hGS, ADP, and a sulfate ion bound in the active site. Domain movements and rearrangement of active site loops change the structure from an open active site form (apoenzyme and γ EC complex) to a closed active site form (hGS-ADP-SO₄²⁻ complex). The structure of hGS shows that two amino acid differences in an active site loop provide extra space to accommodate the longer β -Ala moiety of hGS in comparison to the glycyl group of glutathione. Mutation of either Leu-487 or Pro-488 to an Ala improves catalytic efficiency using Gly, but a double mutation (L487A/P488A) is required to convert the substrate preference of hGS from β -Ala to Gly. These structures, combined with site-directed mutagenesis, reveal the molecular changes that define the substrate preference of hGS, explain the product diversity within evolutionarily related GS-like enzymes, and reinforce the critical role of active site loops in the adaptation and diversification of enzyme function.

INTRODUCTION

The tripeptide glutathione (GSH) is found in nearly all eukaryotes and prokaryotes and functions as a key component in an array of redox-linked cellular systems (Meister, 1995). In plants, GSH maintains cellular redox homeostasis, detoxifies harmful xenobiotics and heavy metals, and can regulate enzyme activity through glutathionylation (May et al., 1998; Noctor and Foyer, 1998; Rouhier et al., 2008). Although GSH is the predominant thiol-containing tripeptide found in plants, various plant species produce glutathione homologs in which the terminal Gly is substituted with a different amino acid (Figure 1A). For example, legumes make GSH, in addition to producing homogluthione (hGS), in which β -Ala replaces Gly, in a tissue-specific manner (Klapheck et al., 1995; Matamoros et al., 1999). Synthesis of hGS maintains redox balance in legume nodules (Moran et al., 2000) and is critical for rhizobia-legume nodulation in roots (Matamoros et al., 2003; Frendo et al., 2005; Loscos et al., 2008). Similarly, many grasses synthesize GSH and hydroxymethylglutathione, with Ser instead of Gly, and exposure to cadmium activates the production of γ -glutamylcysteinylglutamate in maize (*Zea mays*; Rauser et al., 1986; Klapheck et al., 1994; Meuwly et al., 1995). The molecular details of how these peptides

are generated and the biological functions of GSH analogs in plants are poorly understood, but these specialized peptides likely provide for specific responses to various environmental stresses.

Although the biosynthetic routes for the Ser- and Glu-containing peptides are unclear, the two-step pathways leading to GSH and hGS are similar and better understood at the metabolic level. In the first reaction of the pathway, Glu-Cys ligase catalyzes the formation of γ -glutamylcysteine (γ EC) from Glu and Cys (Jez et al., 2004; Hicks et al., 2007). The second step in the synthesis of either GSH or hGS depends on the specificity of the synthetase for the terminal substrate. In nearly all organisms, glutathione synthetase (GS) catalyzes the addition of Gly to γ EC (Meister, 1995; Jez and Cahoon, 2004; Herrera et al., 2007). In legumes, homogluthione synthetase (hGS) uses β -Ala instead of Gly to form hGS (Matamoros et al., 1999; Frendo et al., 2001; Iturbe-Ormaetxe et al., 2002). Although GS and hGS share similar reaction mechanisms based on biochemical and structural studies, the molecular basis for the difference in substrate specificity is unclear due to no available structural data for any plant GS or hGS.

Based on sequence similarity, both GS and hGS are members of the ATP-grasp enzyme superfamily (Galperin and Koonin, 1997). All ATP-grasp family members catalyze the ATP-dependent ligation of the carboxyl group carbon of one substrate to the amino- or imino-nitrogen of another substrate. For example, hGS catalyzes the transfer of the γ -phosphate group of ATP to the C-terminal carboxylate of γ EC to yield an acylphosphate intermediate (Figure 1B). Subsequent nucleophilic attack on this intermediate by β -Ala leads to formation of hGS with release of ADP and inorganic phosphate (Figure 1B). The structurally characterized tetrameric GS from *Escherichia coli* (Yamaguchi

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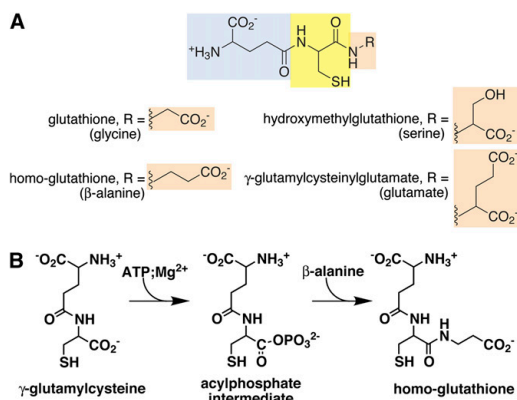


Figure 1. Diversity in GSH Biosynthesis.

(A) Chemical structures of GSH and related peptides from plants. Colors indicate the γ -glutamyl (blue), cysteinyl (yellow), and variable amino acid (peach) residues of each molecule.

(B) Overall reaction catalyzed by hGS synthetase. This scheme shows the formation of the acylphosphate intermediate resulting from phosphorylation of γ -glutamylcysteine and the subsequent addition of β -Ala to yield hGS. Note that the release of ADP and P_i from the reaction is not shown.

et al., 1993) and the dimeric GS from eukaryotes, such as humans, yeast, and plants (Polekhina et al., 1999; Gogos and Shapiro, 2002; Jez and Cahoon, 2004), are unrelated in sequence but share the common ATP-grasp fold. The GS from human, yeast, and *Arabidopsis thaliana* and the hGS from soybean (*Glycine max*) are related, with ~40% sequence identity.

The evolution of GS-related enzymes in plants led to greater product diversity; however, the molecular basis for this adaptation is unknown. To understand the structural evolution of hGS from GS, we determined the x-ray crystal structures of soybean hGS at three separate points during its reaction sequence: (1) the apoenzyme in an open active site conformation, (2) an open form with γ EC bound, and (3) a closed form with hGS, ADP, and a sulfate ion bound in the active site. These structures, combined with site-directed mutagenesis, reveal the structural features that define the substrate preference of hGS, explain the product diversity within evolutionarily related GS-like enzymes, and reinforce the critical role of active site loops in the adaptation and diversification of enzyme function.

RESULTS

Protein Expression and Kinetic Analysis of hGS

Soybean hGS was overexpressed in *E. coli* as a His-tagged fusion protein and purified using Ni²⁺-affinity and size-exclusion chromatographies. Analysis of the protein by SDS-PAGE

showed a monomeric molecular mass of 50 kD, which agrees with the predicted mass based on amino acid sequence (see Supplemental Figure 1 online). The protein eluted from the gel filtration column as a 102-kD species corresponding to a dimer (see Supplemental Figure 1 online). Other eukaryotic GS also are dimeric (Polekhina et al., 1999; Gogos and Shapiro, 2002; Jez and Cahoon, 2004). Purified recombinant hGS had a specific activity of 1.2 $\mu\text{mol min}^{-1} \text{mg protein}^{-1}$ and required Mg²⁺ for activity. Steady state kinetic parameters of hGS for γ EC, ATP, and β -Ala were determined (Table 1). In comparison to the GS from *Arabidopsis* (Jez and Cahoon, 2004; Herrera et al., 2007), hGS displayed a turnover rate (V/E_0) fivefold lower but with comparable K_m values for both ATP and γ EC. In contrast with GS, which shows no activity if Gly is substituted with β -Ala, Ser, or Glu (Jez and Cahoon, 2004), hGS exhibited a 700-fold preference for β -Ala over Gly as the terminal substrate. Estimates of the turnover rate and K_m values of hGS with Gly should be considered as approximate because higher concentrations of Gly, and higher amounts of protein were required to observe activity. hGS did not accept either Ser or Glu as a substrate.

Overall Structure of hGS

Soybean hGS crystallized under similar conditions in either the absence or presence of ligands (Table 2). The protein adopts either a closed active site form (bound with hGS and ADP) or an open active site form (apoenzyme and γ EC bound). In the closed form, the unit cell contained two crystallographically independent molecules with the physiological dimer formed by crystallographic symmetry. For each open-form structure, the asymmetric unit contains two monomers that represent the physiologic dimer.

The overall structure of the closed-form hGS homodimer is shown in Figure 2A. The core structure of each monomer is a triangular α/β -fold that is ~60 Å × 60 Å in length and width, in which binding of hGS and ADP (Figure 2B) defines features of the active site. A smaller lid domain (residues 366 to 427) formed by an antiparallel β -sheet, two α -helices, and a Gly-rich loop (residues 390 to 398) undergoes major conformational changes

Table 1. Comparison of Kinetic Parameters for *Arabidopsis* GS and Soybean hGS

	GS ^a		
	V/E_0 (s ⁻¹)	K_m (μM)	k_{cat}/K_m (M ⁻¹ s ⁻¹)
γ EC	12.2 ± 0.3	39 ± 5	312,800
ATP	12.1 ± 0.3	57 ± 10	212,300
Gly	12.6 ± 0.5	1,510 ± 88	8,340
β Ala	–	–	–
	hGS		
γ EC	2.5 ± 0.1	44 ± 6	56,820
ATP	1.7 ± 0.1	23 ± 4	73,910
Gly	<0.1	>100 mM	1
β Ala	2.4 ± 0.1	3,390 ± 100	708

Values are expressed as a mean ± SE for $n = 3$.

^aKinetic parameters for *Arabidopsis* GS are from Jez and Cahoon (2004) and are provided here for comparison to soybean hGS.

Table 2. Crystallographic Statistics

Crystal	Open	Open + γ EC	Closed + hGSH + ADP
Space Group	P2 ₁	P2 ₁	P3 ₂
Cell dimensions	$a = 64.96 \text{ \AA}$, $b = 80.55 \text{ \AA}$, $c = 90.00 \text{ \AA}$; $\alpha = \gamma = 90.0^\circ$, $\beta = 96.9^\circ$	$a = 64.88 \text{ \AA}$, $b = 80.95 \text{ \AA}$, $c = 89.12 \text{ \AA}$; $\alpha = \gamma = 90.0^\circ$, $\beta = 95.6^\circ$	$a = b = 115.7 \text{ \AA}$, $c = 101.8 \text{ \AA}$; $\alpha = \beta = 90.0^\circ$, $\gamma = 120^\circ$
Data Collection			
Wavelength (\AA)	0.979	0.979	0.979
Resolution range (\AA) (highest shell resolution)	28.4–2.0 (2.05–2.0)	19.8–2.1 (2.16–2.1)	29.3–1.9 (1.95–1.9)
Reflections (total/unique)	132,097/59,099	191,449/51,043	445,529/116,705
Completeness (highest shell)	94.8% (86.8%)	96.5% (92.3%)	97.2% (93.1%)
$\langle I/\sigma \rangle$ (highest shell)	13.7 (3.4)	11.4 (2.4)	17.4 (2.9)
R_{sym}^a (highest shell)	6.9% (37.9%)	10.8% (46.7%)	5.1% (45.8%)
Model and Refinement			
$R_{\text{cryst}}^b/R_{\text{free}}^c$	19.8/26.9	20.7/28.8	19.7/25.0
No. of protein atoms	7108	6992	7424
No. of water molecules	472	290	741
No. of ligand atoms	–	48	152
r.m.s. deviation, bond lengths (\AA)	0.021	0.028	0.047
r.m.s. deviation, bond angles ($^\circ$)	2.03	2.55	3.46
Average B-factor (\AA^2)	26.0	29.3	36.1
Stereochemistry: most favored, allowed, generously allowed	88.4, 9.3, 2.3%	86.2, 12.2, 1.6%	91.1, 8.2, 0.7%

^a $R_{\text{sym}} = \sum |I_h - \langle I_h \rangle| / \sum I_h$, where $\langle I_h \rangle$ is the average intensity over symmetry.

^b $R_{\text{cryst}} = \sum |F_o - \langle F_c \rangle| / \sum F_o$, where summation is over the data used for refinement.

^c R_{free} is defined the same as R_{cryst} but was calculated using 5% of data excluded from refinement.

between the closed and open active site structures (Figures 3B and 3C). In addition, a second loop (residues 479 to 491) forms part of the active site. For consistency with the human and yeast GS structures (Polekhina et al., 1999; Gogos and Shapiro, 2002), this second loop is referred to as the Ala-rich loop, even though the corresponding Ala residues are replaced by a Leu and a Pro in hGS. Dimerization of hGS occurs through a pseudo-twofold axis between two helices ($\alpha 2$ and $\alpha 9$) and an antiparallel β -sheet ($\beta 1$ and $\beta 2$) of each monomer. The closed-form structure of soybean hGS is similar to those of other ATP-grasp family proteins, such as human GS (root mean square [r.m.s.] deviation of 1.6 \AA^2 for 459 C α atoms) (Figure 2C) and yeast GS (r.m.s. deviation of 2.2 \AA^2 for 448 C α atoms). The hGS structure is also related to the synthetase domains of the bifunctional glutathionylspermidine synthetase/amidase from *E. coli* (r.m.s. deviation of 3.8 \AA^2 ; Pai et al., 2006) and trypanothione synthetase/amidase from *Leishmania* (r.m.s. deviation of 4.0 \AA^2 ; Fyfe et al., 2008).

Domain Movements: Open and Closed Active Site Forms

The apoenzyme and γ EC-bound structures are nearly identical, with an r.m.s. deviation of 0.5 \AA^2 . Disordered regions include most of the Gly-rich loop (residues 391 to 396) and other portions of the lid domain (residues 410 to 420) in each monomer of the dimer. For both open form structures, the Ala-rich loop is ordered in one monomer but disordered (residues 480 to 489) in the second monomer. In the open form, both the lid domain and Ala-rich loop are positioned away from the active site to reveal the binding sites for γ EC and ATP and allow for substrate binding (Figures 3A and 3B).

In the closed form, the lid domain, including the Gly-rich loop, and the Ala-rich loop undergo major rearrangements compared with the open form (Figures 3A and 3C). With the exception of residues 410 to 416, the lid domain becomes ordered, with residues in the Gly-rich loop providing multiple interactions with the nucleotide. Likewise, the Ala-rich loop shifts to position residues for contact with both ADP and hGSH. As noted for the bacterial and eukaryotic GS (Polekhina et al., 1999; Gogos and Shapiro, 2002; Jez and Cahoon, 2004), nucleotide binding triggers movement of the lid domain and Ala-rich loop through multiple protein–ligand interactions. Based on the proposed reaction mechanism for GS (Herrera et al., 2007), enclosure of the active site likely prevents hydrolysis of the reactive acyl-phosphate intermediate (Figure 1B).

γ -Glutamylcysteine Binding Site in the Open Form

In the reactions catalyzed by hGS and GS, γ EC is a common substrate, and its binding site is highly conserved in both sequence and structure between hGS and the GS from human, yeast, and *Arabidopsis*. In the γ EC binding site of hGS, Ser-176, Arg-295, Glu-241, and Gln-238 interact with the glutamyl portion of the molecule (Figure 4A). Of these, the charge–charge interaction between the Arg and the carboxylate group is critical for γ EC binding in GS (Herrera et al., 2007), suggesting an analogous role for this interaction in hGS. Similar to interactions observed in the structure of yeast GS complexed with γ EC and an ATP analog (Gogos and Shapiro, 2002), Tyr-298 forms a hydrogen bond to the carbonyl of the glutamyl group and there is a bidentate charge–charge interaction between Arg-153 and the

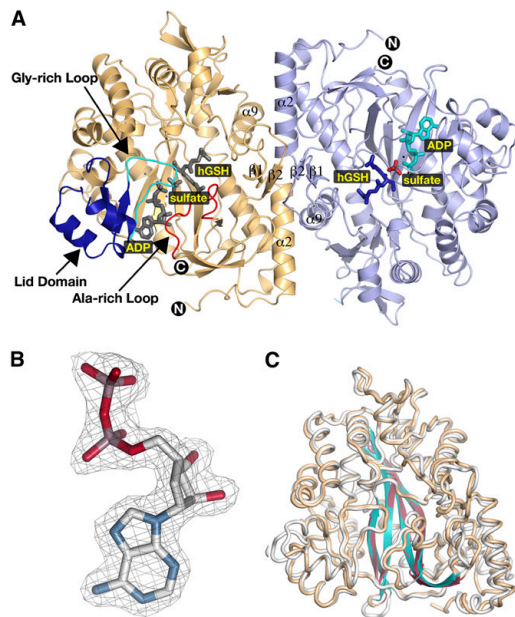


Figure 2. Structure of hGS.

(A) Ribbon diagram of the hGS dimer. Each monomer is colored either gold or blue. Secondary structure elements that form the dimer interface are labeled. The locations of the lid domain (dark blue), Gly-rich loop (cyan), and the Ala-rich loop (red) are highlighted in the gold monomer. The positions of bound ADP (cyan), sulfate (red), and hGSH (dark blue) are highlighted in the blue monomer with corresponding ligands colored gray in the gold monomer. The N- and C-terminal residues of each monomer observed in the electron density maps are indicated.

(B) Sample electron density. The $2F_o - F_c$ omit map (1.0σ) for ADP bound in the active site of the closed form.

(C) Structural overlay of human GS (tan) and soybean hGS (white). The ATP-grasp structural motifs in GS and hGS are colored magenta and cyan, respectively.

carboxylate of the cysteinyl moiety (Figure 4A). Nearly all of these interactions are conserved when hGSH is bound in the site.

Active Site and Ligand Binding in the Closed Form

To define the active site, hGS was cocrystallized in the presence of reaction products ADP and hGSH (Figure 4B). In addition to the reaction products, a sulfate ion and three magnesium ions were identified in the active site of the closed form structure. Clear tetragonal density for the sulfate, which mimics binding of the inorganic phosphate product, was observed (Polekhina et al., 1999). Based on the positional similarity with the yeast and human GS structures, coordination, and strong electron density (4σ), three atoms were modeled as Mg^{2+} .

As with the γ EC binding site, the residues forming the nucleotide binding site between the lid domain and Ala-rich loop of

hGS (Figures 3B and 4B) are structurally conserved with those in the structures of human and yeast GS (Polekhina et al., 1999; Gogos and Shapiro, 2002). The adenosine ring forms main-chain contacts with Ile-427 and Gln-425 and a hydrogen bond with Lys-388. The ribose hydroxyl groups interact with Lys-477 and Glu-450, respectively. A series of polar interactions occur between the diphosphate tail and Lys-334, Asn-397, and two Mg^{2+} ions. The α - and β -phosphate groups of the nucleotide and Glu-169 coordinate one Mg^{2+} with a second ion bound by the β -phosphate group, the sulfate, Glu-169, Asn-171, and Glu-392. Based on mechanistic studies of *Arabidopsis* GS, the magnesium ions and their coordinating residues play critical roles in stabilizing charges during catalysis (Herrera et al., 2007). The functional role of the third Mg^{2+} is unclear, as it does not interact with any of the bound ligands. This ion is coordinated by interactions with Glu-392 and main-chain contacts with Met-170 and Gly-332 that appear to help orient residues coordinated to the other Mg^{2+} ions.

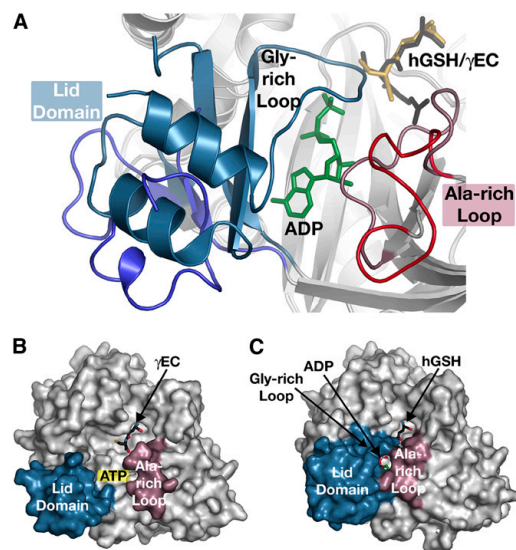


Figure 3. Domain and Loop Movements in hGS.

(A) Ribbon diagram comparing the open and closed active site forms. The active site regions of the γ EC bound open form and the closed form are aligned. Stick drawings show γ EC (gold) in the open form and ADP (green) and hGSH (black) in the closed form. The positions of the lid domain, including the Gly-rich loop, and the Ala-rich loop in the closed form are shown in blue and rose, respectively. The locations of the lid domain and Ala-rich loop in the open form are shown in lighter blue and red, respectively. In the open form, the Gly-rich loop region is disordered.

(B) Surface rendering of the open form bound with γ EC. The lid domain (blue) and Ala-rich loop (rose) leave the nucleotide binding site open.

(C) Surface rendering of the closed form bound with ADP and hGSH. The lid domain (blue) and Ala-rich loop (rose) enclose the active site.

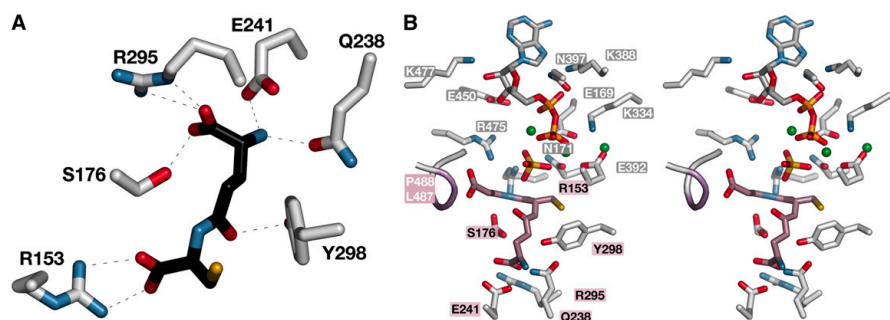


Figure 4. Substrate Binding Sites in the Open and Closed Forms of hGS.

(A) The γ EC binding site. Side chains of residues that form hydrogen bonds (dotted lines) with the bound ligand are shown. Atom types are indicated by colors: red = oxygen; blue = nitrogen; yellow = sulfur.

(B) The active site of hGS. The positions of hGSH (rose), ADP (gray), sulfate (yellow), and magnesium ions (green) and the side chains of interacting residues are shown. A portion of the Ala-rich loop is shown as a tube with the residues that vary between hGS and GS colored. For clarity, main chain and water-mediated contacts are not shown. Residues in the nucleotide binding site, the tripeptide binding site, and the Ala-rich loop are labeled white on gray, black on rose, and white on rose, respectively. Alternate conformers for Ser-176 are shown. Atom types are indicated by colors: red = oxygen; blue = nitrogen; yellow = sulfur; orange = phosphorus.

Within the peptide binding site, all the interactions of the glutamyl portion of hGSH are identical to those observed in the γ EC complex with minor differences in interactions with the cysteinyl group (Figure 4). Ser-176 is observed in alternate conformations. The side chains of Tyr-298 and Arg-153 are repositioned in the closed form complex. Tyr-298 rotates away from the tripeptide, and Arg-153 now interacts with the cysteinyl carbonyl group and the sulfate. The Arg is essential for catalyzing formation of the acylphosphate intermediate in the first part of the catalytic mechanism and in guiding nucleophilic attack in the second half of the reaction to yield the tripeptide product (Herrera et al., 2007). The carboxylate of the β -Ala moiety of hGSH forms a hydrogen bond with the backbone amide of Val-486 and an ionic interaction with the guanido group of Arg-475. Additional van der Waals contacts between the β -Ala-derived portion of hGSH are made with Leu-487 and Pro-488 in the Ala-rich loop. Interestingly, these two residues differ in hGS compared with GS.

Determinants of Substrate Specificity and Product Diversity

In the active site of hGS, Leu-487 and Pro-488 are the only residues that differ from the characterized eukaryotic GS sequences (Figure 5). In GS, these residues are sequential Ala residues, which help give the Ala-rich loop its name. Structural comparison of hGS and human GS shows that the Ala-rich loop in hGS is shifted ~ 3 Å away from the corresponding position of the loop in the GS structure to accommodate the larger β -Ala moiety (Figure 5).

To test the functional significance of Leu-487 and Pro-488 in determining the specificity of hGS for β -Ala over Gly, we generated Ala substitutions at each position (L487A and P488A) and the corresponding double mutant (L487A/P488A). Each mutant protein was expressed, purified, and assayed to determine steady state kinetic parameters for β -Ala and Gly as substrates

(Table 3). Wild-type hGS displays a specificity ratio 708:1 in preference of β -Ala. Each point mutation altered substrate preference to different degrees. Although the L487A mutant shows a 3.4-fold reduction in catalytic efficiency with β -Ala and a 46-fold improvement using Gly as a substrate, this enzyme still prefers the hGS substrate by nearly fivefold. The P488A mutation yields an enzyme with almost equal preference for either substrate, resulting from a 274-fold increase in efficiency with Gly and a minor 2.3-fold reduction in k_{cat}/K_m with β -Ala. The L487A/P488A mutant retains activity with β -Ala at a 10-fold reduction compared with the hGS, but this mutant is as effective with Gly as the parent enzyme is with β -Ala. The combination of substitutions in the double L487A/P488A mutant converts hGS into a GS with a 950-fold increase in k_{cat}/K_m with Gly.

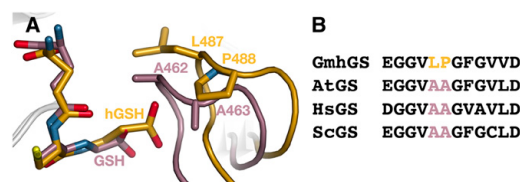


Figure 5. Comparison of the Ala-Rich Loops in hGS and GS.

(A) The overlaid structures of soybean hGS (gold) and human GS (rose) show the similarity of hGSH (gold) and GSH (rose) binding in each structure. The Ala-rich loop of each structure is shown with the side chains of the two amino acid differences between the structures shown.

(B) Sequence comparison of the Ala-rich loops of soybean hGS (GmhGS), *Arabidopsis* GS (AtGS), human GS (HsGS), and yeast GS (ScGS). The amino acid differences are highlighted using the color scheme from **(A)**.

Table 3. Substrate Specificity of Wild-type and Mutant Soybean hGS

	β-Ala		
	V/E_t (s^{-1})	K_m (mM)	k_{cat}/K_m ($M^{-1} s^{-1}$)
hGS	2.4 ± 0.1	3.4 ± 0.1	708
L487A	0.8 ± 0.1	3.8 ± 0.6	211
P488A	2.5 ± 0.2	8.0 ± 1.4	313
LP/AA ^a	1.9 ± 0.2	24.8 ± 2.3	77
		Gly	
hGS	<0.1	>100	1
L487A	0.3 ± 0.1	6.5 ± 0.5	46
P488A	1.4 ± 0.1	5.1 ± 0.4	274
LP/AA ^a	2.1 ± 0.3	2.2 ± 0.1	950

Values are expressed as a mean \pm SE for $n = 3$.

^aDenotes the corresponding double mutant (L487A/P488A).

DISCUSSION

Functional diversity across enzyme families with shared three-dimensional structures and reaction chemistry is a hallmark in the evolution of metabolic pathways. Nearly all eukaryotes and prokaryotes synthesize the multifunctional peptide GSH (Meister, 1995); however, some plants also synthesize GSH analogs with substitutions of the terminal Gly (Figure 1A) (Rausser et al., 1986; Klapheck et al., 1994; Klapheck et al., 1995; Meuwly et al., 1995; Matamoros et al., 1999). In particular, many legumes produce hGSH for root nodulation (Matamoros et al., 2003; Frendo et al., 2005; Loscos et al., 2008). As hGS likely evolved from GS, we examined the structural basis for adaptation of product diversity in hGS. Crystallographic analysis of soybean hGS provides insight on structural changes during the catalytic cycle of both hGS and GS and, combined with site-directed mutagenesis, defines active site differences that govern substrate preference. This work reinforces the critical role of flexible loops in the adaptation and diversification of enzyme function.

Catalysis in hGS and GS requires the orchestration of binding multiple substrates and the rearrangement of active site features, including the lid domain, Gly-rich loop, and Ala-rich loop. Together with studies of the kinetic and chemical mechanisms of GS (Jez and Cahoon, 2004; Herrera et al., 2007), crystal structures of hGS (Figures 2 and 3) and GS (Polekhina et al., 1999; Gogos and Shapiro, 2002) now provide views of the progression through the catalytic cycle from apoenzyme (hGS and yeast GS) to first substrate complex (hGS- γ EC complex) to second substrate complex (yeast GS in complex with γ EC and an ATP analog) to product complex (hGS and human GS). Kinetic analysis of *Arabidopsis* GS indicates a mechanism in which γ EC is the preferred first substrate followed by ATP (Jez and Cahoon, 2004). Within the active site, hGS shares common structural and chemical features with GS. In both enzymes, the γ EC binding site is structurally static, whereas the ATP and β -Ala/Gly binding sites are dynamic. The structure of the hGS- γ EC complex in the open active site conformation (Figures 3B and 4A) provides direct evidence for formation of this complex in agreement with the predicted mechanism for GS and hGS. Binding of ATP, which makes extensive contacts with

residues in the lid domain, Gly-rich loop, and Ala-rich loop (Figures 3C and 4B), likely triggers transformation to the closed active site structure (Gogos and Shapiro, 2002; Gunasekaran et al., 2003). The closed active site conformation protects the reactive acylphosphate reaction intermediate from hydrolysis (Figure 1B) and orders the Ala-rich loop to form the binding site for either Gly or β -Ala (Figure 4B). Functionally, these conformational changes provide a cooperative linkage through the reaction cycle as binding of one substrate enhances binding of the next substrate, as suggested by the interaction factors in the kinetic mechanism (Jez and Cahoon, 2004). Structural/functional analysis of hGS and GS also suggests that the dynamic nature of the active site is important for catalysis and substrate recognition.

The structural conservation between the active sites of hGS and GS implies a shared reaction mechanism (Herrera et al., 2007). In the first half of the hGS reaction, formation of the acylphosphate intermediate occurs by transfer of the γ -phosphate of ATP to γ -glutamylcysteine. For this step, the Mg^{2+} ions in the active site orient the phosphate group and Arg-153 likely stabilizes the transition state. In the second half of the reaction, nucleophilic attack of the β -Ala amino group on the acylphosphate intermediate releases phosphate and yields hGSH. Positioning of Arg-153 and the Mg^{2+} bound by Glu-169 and Asn-171 would stabilize the transition state with the Ala-rich loop and Arg-475 orienting β -Ala for attack on the reaction intermediate to yield hGSH.

The major difference between hGS and GS is substrate specificity for β -Ala and Gly, respectively. In each enzyme, residues in the Ala-rich loop contact the terminal residue of the tripeptide product (Figure 5). A Leu and Pro in the hGS from soybean and other legumes replaces the invariant double Ala sequence of the eukaryotic GS (Moran et al., 2000; Frendo et al., 2001; Iturbe-Ormaetxe et al., 2002; Skipsey et al., 2005). Structurally, the Ala-rich loop of hGS shifts relative to the same loop in GS to allow space for binding of the larger hGSH product and β -Ala substrate (Figure 5A). Site-directed mutagenesis of Leu-487 and Pro-488 demonstrates that changes at both positions are necessary to convert hGS ($k_{cat}/K_m^{\beta\text{-ala}} = 708 M^{-1} s^{-1}$) to a GS with comparable catalytic efficiency ($k_{cat}/K_m^{gly} = 950 M^{-1} s^{-1}$) (Table 3). Interestingly, the L487A/P488A mutant retains limited activity with β -Ala ($k_{cat}/K_m^{\beta\text{-ala}} = 77 M^{-1} s^{-1}$). This suggests that additional changes in the Ala-rich loop, or more subtle allosteric mutations, may be required to completely shift substrate preference and product specificity. The mobility of active site features in both hGS and GS (i.e., the lid domain and Ala-rich loop) likely plays a role in determining the rate of catalysis and for allowing evolutionary changes in these enzymes.

In both hGS and GS, structuring of the lid domain and Ala-rich loop appears linked to binding of ATP and the terminal substrate (i.e., β -Ala or Gly). Although the rate constants for each step in the catalytic cycle of either enzyme are unknown, the crystal structures of these enzymes suggest that dynamic active site structures may limit catalysis and explain the different turnover rates of GS ($k_{cat} \sim 12 s^{-1}$) and hGS ($k_{cat} \sim 2 s^{-1}$) (Table 1) (Gunasekaran et al., 2003; Tokuriki and Tawfik, 2009). Based on these results, it is possible that the nucleophilic attack of the terminal substrate is a limiting step in the reaction mechanism. Presumably, GS is a highly evolved enzyme in eukaryotes

because of the central role that glutathione plays in regulating intracellular redox state (Meister, 1995). By contrast, hGS likely evolved by gene duplication and subsequent mutation (Tokuriki and Tawfik, 2009), and additional sequence changes in the lid domain and/or Ala-rich loop may be needed to optimize interactions with substrates and the movement of active site features.

Active site loops are central in the evolution of enzyme functionality (Todd et al., 1999; Penning and Jez, 2001; Gunasekaran et al., 2003; Tokuriki and Tawfik, 2009). The flexible and mutable nature of loops allows for the sampling of the new sequences and localized structures that generate shifts in substrate specificity or new catalytic activity. Frendo et al. (2001) originally proposed that legumes evolved hGS from gene duplication of GS after the divergence of the order Fabales, which includes the legumes, from other flowering plants. Our results suggest a molecular mechanism underpinning the evolution of hGS from GS. Although hGS retains the γ EC and ATP binding sites and maintains the positioning of catalytically essential Arg residues (Arg-153 and Arg-475) and key Mg^{2+} ions, two changes in the Ala-rich loop are sufficient to alter substrate specificity.

While this work helps illuminate the molecular basis for hGS evolution from an ancestral GS, many questions remain as to the role hGS and other GS analogs in plants. Although the interplay between genomes, protein function, and a plant's environment shapes the evolution of new metabolism, it is unclear why legumes required evolution of hGS and hGS production in nodules. Aside from the shared localization of hGS in nodules (Moran et al., 2000; Frendo et al., 2001; Iturbe-Ormaetxe et al., 2002; Skipsy et al., 2005), there appears to be no correlation between the presence of hGS in a legume species and the position of that species in the legume phylogeny (Wojciechowski et al., 2004). Nonetheless, given the conservation of hGS in the legumes examined so far, it seems likely that environmental factors, such as nodulation and/or habitat, contributed to the diversification of GSH metabolism. In addition, as suggested by the presence of Ser- and Glu-containing GSH analogs in other plants (Rauser et al., 1986; Klapheck et al., 1994; Meuwly et al., 1995), the adaptation of GSH biosynthesis for production of specialized tripeptides in response to environmental stresses may be more widespread. Continued genomic and biochemical explorations of legumes, and other plants, promise new insights on how these plants evolved more specialized environmental response systems.

METHODS

Materials

All oligonucleotides were synthesized by Integrated DNA Technologies. Ni^{2+} -nitrilotriacetic acid (NTA) was from Qiagen. Benzamidine-sepharose and the HiLoad 26/60 Superdex-200 FPLC column were purchased from GE/Amersham Health Sciences. The QuikChange site-directed mutagenesis kit was from Stratagene. hGS was from Bachem. All other reagents were of ACS grade or better and were purchased from Sigma-Aldrich.

Protein Expression, Purification, and Mutagenesis

Soybean (*Glycine max*) hGS was PCR-amplified from a soybean seed cDNA library using 5'-dTTCATGGCATGGCTCAACCTTTGACC-

ACC-3' as the forward primer (the *NcoI* site is underlined, and the start codon is in bold) and 5'-dTTCGCGGCCGCTCAAGTTAGGTATACAGTATCTACCAC-3' as the reverse primer (the *NotI* site is underlined, and the stop codon is in bold). The resulting PCR product was digested with *NcoI* and *NotI* and then subcloned into pHIS8 (Jez et al., 2000) for expression of an N-terminally octahistidine-tagged protein. Automated nucleotide sequencing confirmed the fidelity of the bacterial expression construct (Washington University Sequencing Facility).

Transformed *Escherichia coli* BL21(DE3) cells were grown at 37°C in Terrific broth containing 50 μ g mL^{-1} kanamycin until $A_{600} \sim 0.8$. After induction with 1 mM isopropyl 1-thio- β -D-galactopyranoside, the cultures were grown at 20°C for 4 to 8 h. Cells were pelleted by centrifugation (10,000g; 10 min) and resuspended in 50 mM Tris-HCl, pH 8.0, 500 mM NaCl, 20 mM imidazole, 5 mM $MgCl_2$, 10% (v/v) glycerol, and 1% (v/v) Tween 20. Sonication was used to lyse cells. Following centrifugation (45,000g; 45 min), the supernatant was passed through a Ni^{2+} -NTA column. The column was then washed with the same buffer minus Tween 20. His-tagged protein was eluted with 50 mM Tris-HCl, pH 8.0, 500 mM NaCl, 250 mM imidazole, 5 mM $MgCl_2$, and 10% (v/v) glycerol. Incubation with thrombin (1/1000th the amount of hGS by weight) during overnight dialysis at 4°C against wash buffer removed the His tag. Dialyzed protein was reloaded on a mixed benzamidine-sepharose/ Ni^{2+} -NTA column. The flow-through of this step was loaded onto a HiLoad 26/60 Superdex-200 FPLC column equilibrated with 25 mM HEPES, pH 7.5, 5 mM $MgCl_2$, and 100 mM NaCl. Fractions containing purified protein were pooled, concentrated to 10 to 12 mg mL^{-1} , and stored at $-80^\circ C$. Protein concentration was determined by the Bradford method (Protein Assay; Bio-Rad) with BSA as standard.

Site-directed mutants of hGS (L487A, P488A, and L487A/P488A) were generated using oligonucleotides containing the desired mutations (see Supplemental Table 1 online) and the QuikChange PCR method with the pHIS8-hGS vector as template. Introduction of the desired mutation was confirmed by sequencing of the constructs. Expression and purification of each mutant protein was performed as described for the wild-type protein.

Enzyme Assays

The activity of hGS was determined spectrophotometrically at 25°C by measuring the rate of formation of ADP using a coupled assay with pyruvate kinase and lactate dehydrogenase. A standard reaction mixture (0.5 mL) contained 100 mM HEPES, pH 7.5, 150 mM NaCl, 20 mM $MgCl_2$, 2.5 mM γ EC, 10 mM β -Ala (or Gly), 2.5 mM disodium ATP, 2 mM sodium phosphoenolpyruvate, 0.2 mM NADH, 5 units of type III rabbit muscle pyruvate kinase, and 10 units of type II rabbit muscle lactate dehydrogenase. The rate of decrease in A_{340} ($\Sigma = 6270 M^{-1} cm^{-1}$) was observed using a Beckman DU800 UV/Vis spectrophotometer. Steady state kinetic parameters were determined by initial velocity experiments in which concentrations for two substrates were fixed at saturating levels and the third substrate concentration varied (0.2 to 10 times the K_m value). Untransformed data was fit to the Michaelis-Menten equation, $v = k_{cat}[S]/(K_m + [S])$, using Kaleidagraph (Synergy Software).

Protein Crystallization and Structure Determination

Crystals of hGS were obtained by the vapor diffusion method in 4- μ L hanging drops of a 1:1 mixture of protein and crystallization buffer (20% PEG3000, 0.1 M MOPSO, pH 7, and 0.2 M $MgSO_4$) at 4°C over a 0.5-mL reservoir. For cocrystallization with ligands, either 5 mM γ EC or 2.5 mM ADP and 5 mM hGS was added to the protein before crystallization. All crystals were stabilized in cryoprotectant (crystallization solution plus ligands with 15% [v/v] glycerol) before flash freezing in liquid nitrogen. Data collection (100K) was performed at the Stanford Synchrotron

Radiation Facility (SSRL) on monochromatic beamline 9-1. Diffraction data was integrated and reduced using XDS (Kabsch, 1993) and scaled with XSCALE (Kabsch, 1993). The structure of closed-form hGS in complex with ADP and hGSH was solved by molecular replacement performed with PHASER (McCoy et al., 2007) using a homology model of the soybean enzyme generated with SWISS-MODEL (Kopp and Schwede, 2003) from the structure of human GS (PDB: 2HGS; Polekhina et al., 1999). Model building was performed in O (Jones et al., 1993), and all refinements were performed with REFMAC (Murshudov et al., 1997). Waters were added using ARP (Lamzin and Wilson, 1993). Quality of the model was evaluated using PROCHECK (Laskowski et al., 1993). Structures of the open form hGS and open form hGS in complex with γ EC were solved by molecular replacement using the final closed form hGS structure. Modeling building, refinement, and assessment were performed as above. Crystal parameters, data collection statistics, and refinement statistics for the three structures are summarized in Table 2. Atomic coordinates and structure factors have been deposited in the Protein Data Bank (www.rcsb.org). All structural figures were generated with PyMol (http://www.pymol.org).

Accession Numbers

Sequence data from this article can be found in the GenBank/EMBL database under the following accession numbers: soybean hGS (accession CAB91078), human GS (PDB: 2HGS; accession NP_000169), *Saccharomyces cerevisiae* GS (accession CAA74136), and *Arabidopsis thaliana* GS (J22359). Coordinates and structure factors for the soybean hGS apoenzyme (PDB: 3KAJ), γ -glutamylcysteine complex (PDB: 3KAK), and the hGS+ADP complex (PDB: 3KAL) have been deposited in the RCSB Protein Data Bank.

Supplemental Data

The following materials are available in the online version of this article.

Supplemental Figure 1. Protein Expression and Purification Analysis.

Supplemental Table 1. Oligonucleotide Primers Used for Site-Directed Mutagenesis.

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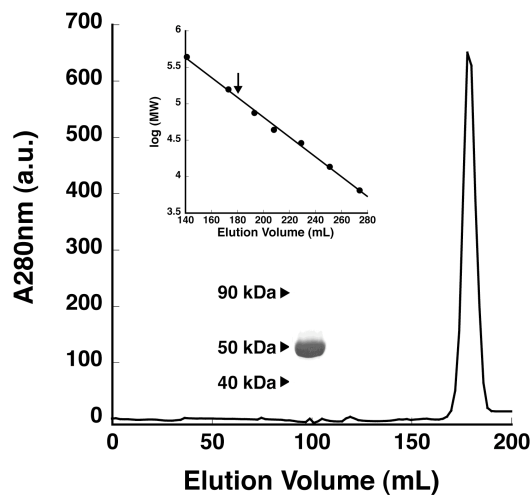
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Supplemental Figure 1. Protein expression and purification analysis. Size-exclusion chromatography of GmhGS. Purified hGS was chromatographed on a Superdex-200 26/60 FPLC column with 25 mM Hepes (pH 7.5), 5 mM MgCl₂, and 100 mM NaCl. The inset graph shows the molecular weight calibration of the column. The following standards were used: ferritin (440 kDa), aldolase (158 kDa), conalbumin (75 kDa), ovalbumin (44 kDa), carbonic anhydrase (24 kDa), ribonuclease A (13.7 kDa), and aprotinin (6.5 kDa). The arrow represents the elution volume of soybean hGS. The inset SDS-PAGE shows the purified protein stained with Coomassie Blue. Arrows correspond to molecular weight markers as indicated.

Supplemental Table 1. Oligonucleotide primers used for site-directed mutagenesis.

L487A	5'-dCTTATGAAGGAGGAGTT GCGC CTGGTTTTGGAGTGGTAG-3'
P488A	5'-dCTTATGAAGGAGGAGTTTT GGCT GGTTTTGGAGTGGTAG-3'
L487A/P488A	5'-dCTTATGAAGGAGGAGTT GCGCT GGTTTTGGAGTGGTAG-3'

For PCR-based mutagenesis, complementary sense and antisense primers were used. The table only shows the sense sequence. Codons encoding the mutations are in bold type.

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CHAPTER 3

REDOX-REGULATORY MECHANISMS INDUCED BY OXIDATIVE STRESS IN

BRASSICA JUNCEA ROOTS MONITORED BY 2-DE PROTEOMICS

PREFACE

In this chapter, I describe the application of an NEM and IAF-based thiol labeling strategy to protein extracts from *B. juncea* roots that had been treated with either 1mM H₂O₂ or 50µm BSO. The framework for this series of experiments arose from a 2008 NSF grant application submitted by Dr. Joseph Jez and Dr. Leslie Hicks; that grant was in turn based upon earlier experiments which determined that GCL utilizes intramolecular disulfide bonds as a means of redox regulation [Jez et al., 2004; Hicks et al., 2007]. Because at the time only a handful of proteins (only one of which was from plants) which utilized thiol-based regulatory switches had been identified, the grant, among other things, proposed the use of a 2D-SDS-PAGE/LC-MS/MS methodology for identifying additional candidate proteins. The methodology itself was not entirely new - variants had previously been used for identifying thiol-containing proteins in mammals, yeast, and bacteria [Yang et al., 2007; Le Moan et al., 2006; Dosanjh et al., 2005]. However, in plants to date use of the technique had been much more limited, and primarily focused on the identification of novel thioredoxin targets [Yano et al., 2002; Lee et al., 2004; Yano and Kuroda, 2005]. Instead of using thioredoxin to reduce protein extracts, in our approach we opted to use the general reductant DTT. The advantage of this choice was that we would be able to identify target proteins that are reduced by other “doxins” besides thioredoxin, or proteins for which the physiological reductant is unknown.

Because dataset briefs published in the journal Proteomics are limited to ~2500 words, the body of this chapter contains only an abbreviated description of the protein

extraction and labeling methodology employed. Thus, I have included a more detailed version at the end of the chapter.

Author Contributions: JMJ and LMH designed research; AG performed research; SA, AG, JMJ, and LMH analyzed data; AG, JMJ, LHM, and SA wrote the paper.

Preface References

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DATASET BRIEF

Redox-regulatory mechanisms induced by oxidative stress in *Brassica juncea* roots monitored by 2-DE proteomics

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ROS, including hydrogen peroxide (H₂O₂), can serve as cellular signaling molecules following oxidative stress. Analysis of the redox state of proteins in *Brassica juncea* roots by 2-DE proteomics following treatment with either exogenous H₂O₂ or buthionine sulfoximine, which depletes glutathione to cause accumulation of endogenous H₂O₂, led to the identification of different sets of proteins. These data suggest that exogenous and endogenous oxidative stresses trigger specialized responses.

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**Keywords:**

2-DE / Buthionine sulfoximine / Hydrogen peroxide / Oxidative stress / Plant proteomics

ROS produced endogenously in response to environmental changes serve as signaling molecules in communications within and between cells [1]. Among ROS, hydrogen peroxide (H₂O₂) causes reversible and irreversible redox modifications to proteins during oxidative stress [2, 3]. Although many H₂O₂-induced protein modifications result in irreversible oxidative damage, reversible modification of cysteines (i.e., oxidation of thiols to disulfide bonds, glutathionylation, or S-nitrosylation) is an important mechanism for regulating protein function. To balance between deleterious effects and oxidative signaling, intracellular H₂O₂ levels are controlled by mechanisms, such as the glutathione-ascorbate system, that maintain concentrations of key reducing molecules [1]. As a consequence, H₂O₂ has long been used to elicit oxidative stress responses to study redox mechanisms and provide insight into the molecular physiology of adaptive responses.

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Abbreviations: BSO, buthionine sulfoximine; DHAR, dehydroascorbate reductase; H₂O₂, hydrogen peroxide; IAF, 5-iodoacetamidofluorescein; TPI, triose phosphate isomerase; TRXh, H-type thioredoxin

In this study, we examine the changes in the redox proteome of *Brassica juncea* (Indian mustard) roots using specific labeling of cysteines by 5-iodoacetamidofluorescein (IAF) in response to exogenous and endogenous H₂O₂-induced oxidative stresses. Application of H₂O₂ to plant roots provides an exogenous stress and application of buthionine sulfoximine (BSO), which depletes glutathione, produces an accumulation of endogenous H₂O₂ [4]. Largely different sets of proteins regulated by H₂O₂ were identified for each treatment at the redox and abundance levels. Interestingly, proteins involved in similar biological processes, such as the brassinosteroid signaling pathway, were differentially regulated by each H₂O₂ source in *B. juncea* roots.

Wild-type *B. juncea* seeds were germinated in a growth chamber at 22°C, 200 μmol/m² light intensity, 50% relative humidity, during a 16-h light/8-h dark cycle. After 3 wk, seedlings were transplanted to 3.8 L pots in the greenhouse (same light/dark cycle). After 6 wk, plants were treated with 2 L of distilled water, 1 mM H₂O₂, or 50 μM BSO, positioned to allow rapid draining, and after 1 h treated again with 1 L of solution. Following draining (2 h), roots were washed to remove soil, flash-frozen in liquid nitrogen, and stored at –80°C. Concentrations of H₂O₂ and BSO were chosen

*These authors have contributed equally to this study.

Colour Online: See the article online to view Fig. 1 in colour.

based on a previous experiment, showing that these compounds alter oxidation state of a redox-sensitive protein in planta [5].

For each treatment, three biological replicate samples from three different plants were obtained for processing. Root tissue (~800 mg FW) was ground and suspended in extraction buffer (100 mM Tris-HCl, pH 8.0; 100 mM *N*-ethylmaleimide (NEM); 1% CHAPS; 1% protease inhibitor cocktail (Sigma, St. Louis, USA) to 200 mg/mL for protein extraction and alkylation of free sulfhydryl groups (Fig. 1, step 1). Samples were centrifuged and the soluble protein fraction was removed, precipitated with methanol 3 ×, resuspended in 150 μL of reduction buffer (50 mM Tris-HCl, pH 8.0, 7 M urea, 2 M thiourea, 50 mM DTT), and incubated for 15 min (25°C) to reduce disulfide bonds (Fig. 1, step 2). Proteins were next precipitated with methanol 3 ×, resuspended in 150 μL of labeling buffer (40 mM HEPES, pH 7.5; 50 mM NaCl; 200 μM IAF), and incubated for 10 min (25°C) for the labeling reaction (Fig. 1, step 3). Proteins were precipitated with methanol 3 × and resuspended in destreak rehydration buffer (GE Healthcare, Waukesha, WI, USA). Protein concentrations were determined by CBX protein assay (G-Biosciences, St. Louis, USA).

Extracted protein (200 μg) was loaded onto pH strips 4-7 (Bio-Rad, Hercules, CA, USA) and 2-DE performed as described previously [6]. Gels were imaged with a Typhoon 9410 (GE Healthcare) to detect IAF-labeled proteins ($\lambda_{ex} = 488$ nm and $\lambda_{em} = 526$ nm). Gels were then stained with Sypro Ruby and imaged to detect total proteins ($\lambda_{ex} = 457$ nm and $\lambda_{em} = 610$ nm). Image analysis, including gel alignment, spot averaging and normalization, and multivariate statistics, employed SameSpots software (Nonlinear Dynamics, Durham, NC, USA) to determine which protein spots changed in protein abundance and oxidation in response to H₂O₂ and BSO treatments relative to controls. Means and standard deviations were calculated from three replicates and compared between control and treatments using ANOVA. Spots with a *p*-value of <0.05 were picked for protein identification via trypsin digestion and LC-MS/MS as described previously [6]. The peptide tandem mass spectra were processed using Analyst QS v1.1 (AB Sciex,

Foster City, CA, USA) and searched against the NCBI nr database (July 2010, 11 368 323 sequences) using an in-house version of MASCOT v2.20 (Matrix Science, Boston, MA, USA) with the following parameters: tryptic peptides with ≤1 missed cleavage site; precursor and MS/MS fragment ion mass tolerances of 0.8 and 0.8 Da, respectively; variable carbamidomethylation and fluoresceination of cysteine; and variable oxidation of methionine. The data were filtered using Scaffold 3 (Proteome Software, Portland, OR, USA). Positive identification criteria were ≥2 peptide sequences, protein probability of 99.9%, and peptide probability of 80%.

A total of 59 and 50 spots showed significant changes (*p* < 0.05) in redox-state after H₂O₂ and BSO treatments, respectively, and 27 and 40 spots differed significantly in total protein abundance for the H₂O₂ and BSO treatments, respectively (Table 1 and Supporting Information Table 1). For the four comparisons (Table 1), the *q*-values ranged from 12 to 37% and from 55 to 61% of the spots were confidently identified by LC-MS/MS using the criteria described. Of the 103 spots confidently identified, only the 52 spots containing a single protein were used for further analysis of redox and abundance changes. These proteins were categorized according to their biological process (Supporting Information Table 2). The 29 proteins that change in redox state after H₂O₂ and BSO treatments are most represented in amino acid biosynthesis, redox homeostasis, and glycolysis (Fig. 2). The two main biological processes in which the 23 proteins change in abundance are redox homeostasis and defense response (Supporting Information Table 2 and Supporting Information Fig. 1). Images from IAF labeling and Sypro staining were overlaid to identify possible co-regulation of redox and protein abundance changes (Supporting Information Fig. 2). A significant number of the protein spots do not overlap and none of the protein spots showing changes in abundance overlapped with the ones identified as redox regulated. Thus, specificity of the post-translational redox change is largely independent of changes in total protein abundance. Only the redox changes will be discussed further.

Multiple proteins, such as dehydroascorbate reductase (DHAR), glutathione-S-transferases (GST), and H-type thioredoxins (TRXh), involved in redox homeostasis were identified as changed in oxidation state following each treatment. In response to H₂O₂ application, DHAR, which is essential for the glutathione-ascorbate cycle, showed decreased IAF spot intensity, indicating greater reduction of the enzyme compared with the control (Fig. 2). Spinach DHAR contains a thiol group required for reduction of oxidized glutathione [7]. Thus, a change in the redox state of DHAR may increase the regeneration of ascorbate from dehydroascorbate and enhance detoxification of H₂O₂. Two GST isoforms showed increased oxidation in response to BSO and H₂O₂ and one isoform was more reduced only following BSO treatment (Fig. 2). GSTs catalyze the conjugation of reduced glutathione to sulfhydryl groups of

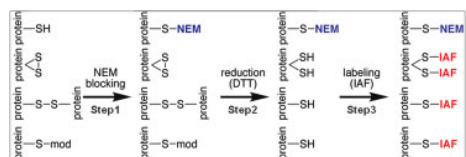


Figure 1. Redox Proteome Labeling Approach. Proteins with free thiols (–SH), disulfide bonds (–S–S–), or modified cysteines (–S–mod) are incubated with *N*-ethylmaleimide to block free sulfhydryl groups. Oxidized thiols are reduced with DTT. The resulting free thiols are fluorescently labeled with IAF and the proteins separated by 2-DE and identified by LC-MS/MS.

Table 1. Total number of spots differentially expressed and oxidized ($p < 0.05$) in response to H₂O₂ and BSO

	H ₂ O ₂ /Sypro	H ₂ O ₂ /IAF	BSO/Sypro	BSO/IAF	Total
Total number of spots detected	235	243	288	250	–
Number of spots differentially expressed or oxidized	27 (37%)	59 (12%)	40 (29%)	50 (16%)	176
Number of spots identified as one protein ID	11	17	12	12	52
Number of spots identified with multiple proteins IDs	4	19	12	16	51
Total number of spots identified	15 (55%)	36 (61%)	24 (60%)	28 (56%)	103

The q -values for each experiment for the number of spots significantly different are indicated in parentheses.

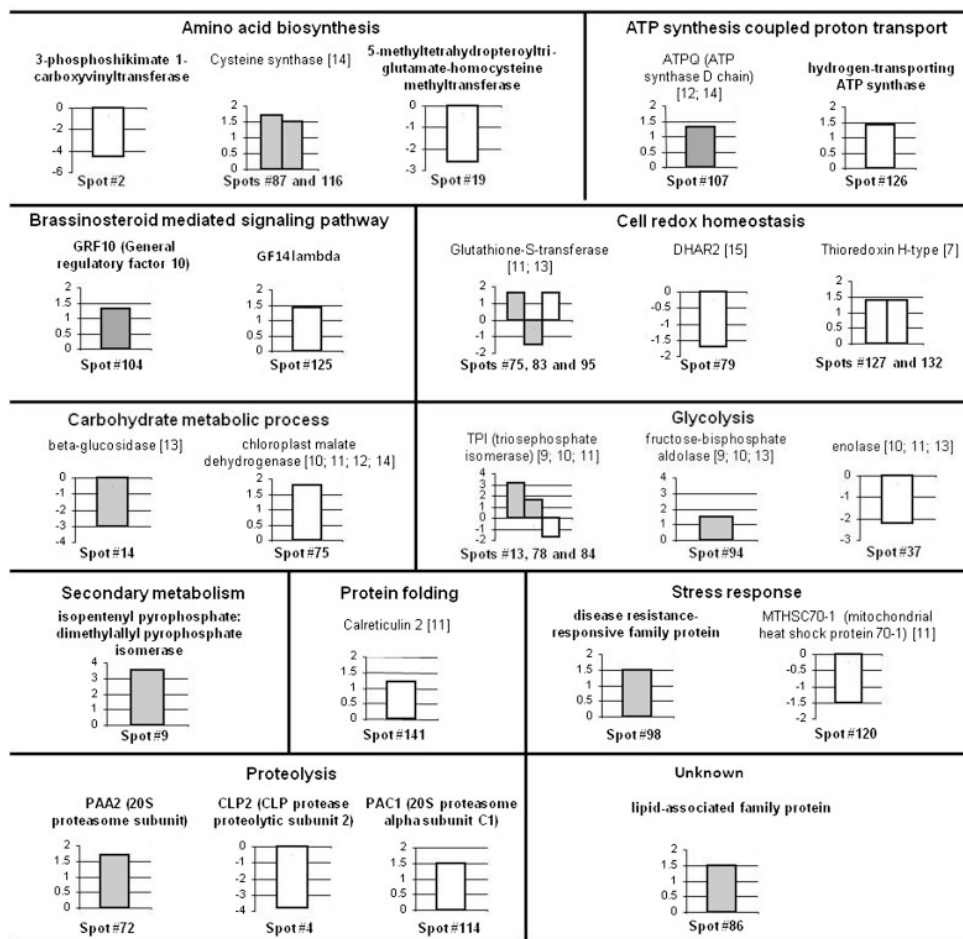


Figure 2. Differential Protein Redox Changes. Proteins with a change in redox state were grouped by biological function with fold change in oxidation shown in the bar graph. White and grey bars correspond to H₂O₂ and BSO treatments, respectively. Proteins described for the first time as redox-altered proteins are indicated in bold. For the proteins previously described as either disulfide-bonded proteins or S-thiolated protein, the reference numbers are indicated.

proteins and small molecules; in the case where BSO inhibits the biosynthesis of glutathione, GST activity also likely decreases in the absence of substrate. The increase in oxidized state of GST is mainly due to the oxidative conditions from H₂O₂ accumulation. Two TRXh isoforms also showed greater oxidation following H₂O₂ application, indicating increased disulfide formation and/or thiol modification (Fig. 2). TRXhs reduce disulfide bonds in a range of proteins to provide a mechanism for regulating redox imbalance [8]. In poplar, mitochondrial TRXh2 contains a glutathionylation site that modifies the redox potential of TRXh2 to decrease its activity [7], but in pea TRXh isoforms can differentially effect redox imbalance regulation [9]. Increased oxidation of TRXh in *B. juncea* may result from elevated demand to modulate H₂O₂ effects. Overall, exogenous H₂O₂ yields more changes on cellular antioxidant mechanisms such as the glutathione–ascorbate cycle and the thioredoxin system than BSO treatment.

Proteins in glycolysis, stress response, carbohydrate metabolism, and proteolysis also exhibited redox changes (Fig. 2). As relatively little information (as compared with mammalian systems) on redox regulation of proteins in plants is available, it is difficult to define the redox effect on the biological process according to the treatment since the reduction/oxidation of a particular protein can cause activation and/or repression of the protein activity [3]. Several proteins identified in this study as redox-sensitive are known targets of thioredoxins and/or glutathionylation (Fig. 2). For example, triose phosphate isomerase (TPI), a glycolytic enzyme, was first identified as a target for glutathionylation in *Arabidopsis* [10]. TPI requires glutathionylation for maintaining activity and oxidized glutathione inhibits the enzyme. TPI is also regulated by thioredoxin in the endosperm during germination of cereal grains [11] and *Medicago truncatula* seeds [12]. Here, several isoforms of TPI were identified as changing in redox state following BSO and H₂O₂ treatment. Two isoforms showed increased oxidation after BSO treatment, whereas one isoform showed greater reduction only following H₂O₂ treatment. Decreased glutathione levels after BSO treatment increases the oxidation state of cells and may trigger the specific oxidation of TPI to maintain energy production through the thioredoxin system. Although a specific modification may result from a treatment, we were not able to determine if the modification was either formation of a disulfide bridge or glutathionylation. BSO treatment also increased the oxidation of a second glycolytic enzyme, fructose-biphosphate aldolase. The previous studies demonstrate that this enzyme is glutathionylated in *Arabidopsis* [10] and is a thioredoxin target during germination of wheat grains [11]. On the contrary, enolase showed increased reduction in response to H₂O₂ (Fig. 2). Enolase is also a thioredoxin target and is redox regulated during germination of wheat grains and *M. truncatula* seeds [11, 12]. Additional proteins identified from carbohydrate metabolism and ATP-coupled proton transport are known to be redox regulated (Fig. 2) [13–16].

Potential new disulfide-containing proteins in amino acid synthesis and proteolytic processing were also identified in this study, including 3-phosphoshikimate 1-carboxyvinyl-transferase, cobalamin-independent methionine synthase, the PAA2 20S proteasome subunit, the CLP protease proteolytic subunit 2, and 20S proteasome α -subunit C1. More interestingly, two 14-3-3 proteins involved in brassinosteroid signaling, general regulatory factor 10 (GRF10 or GF ϵ) and GF14 λ , were identified as increased in oxidation state in response to BSO and H₂O₂ treatments, respectively (Fig. 2) [17]. Protein phosphorylation mediates the interaction of 14-3-3 proteins with target proteins. Redox modification of 14-3-3 proteins may change protein conformation, thus impairing protein–protein interaction and inactivating signaling pathways. Brassinosteroids are plant hormones involved in a range of cellular and physiological processes including plant growth and tolerance to a variety of abiotic and biotic stresses [18, 19]. Brassinosteroids induce H₂O₂ in cucumber leaves and increase oxidative tolerance [20]. In this study, the application of H₂O₂ and the induction of endogenous H₂O₂ may have different effects on 14-3-3 proteins and possibly alter brassinosteroid signaling involved in the induction of oxidative stress tolerance.

In conclusion, several new oxidative stress redox-regulated proteins were identified using a specialized 2-DE proteomics approach. These results showed that specific redox and protein induction occurred when H₂O₂ was applied directly, including changes of specific protein isoforms, and that different mechanisms can be induced if redox regulation mechanisms, such as the glutathione–ascorbate cycle, are blocked to increase endogenous H₂O₂ levels. By resolving different protein isoforms either from the same gene family or from differential post-translational modifications, 2-DE proteomics has proven its utility to decipher the complexity of redox regulation mechanisms in plants. This approach is directly applicable to examine biologically relevant stress situations on agronomic crops, and could significantly impact the understanding of redox regulation both generally and specifically to facilitate crop improvement efforts.

Supporting data are accessible in the PRIDE database, login review33615, password hTXrNqWY, direct link <http://www.ebi.ac.uk/pride/login.do>.

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The authors have declared no conflict of interest.

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PROTEOMICS

**Supporting Information
for Proteomics**

DOI 10.1002/pmic.201000450

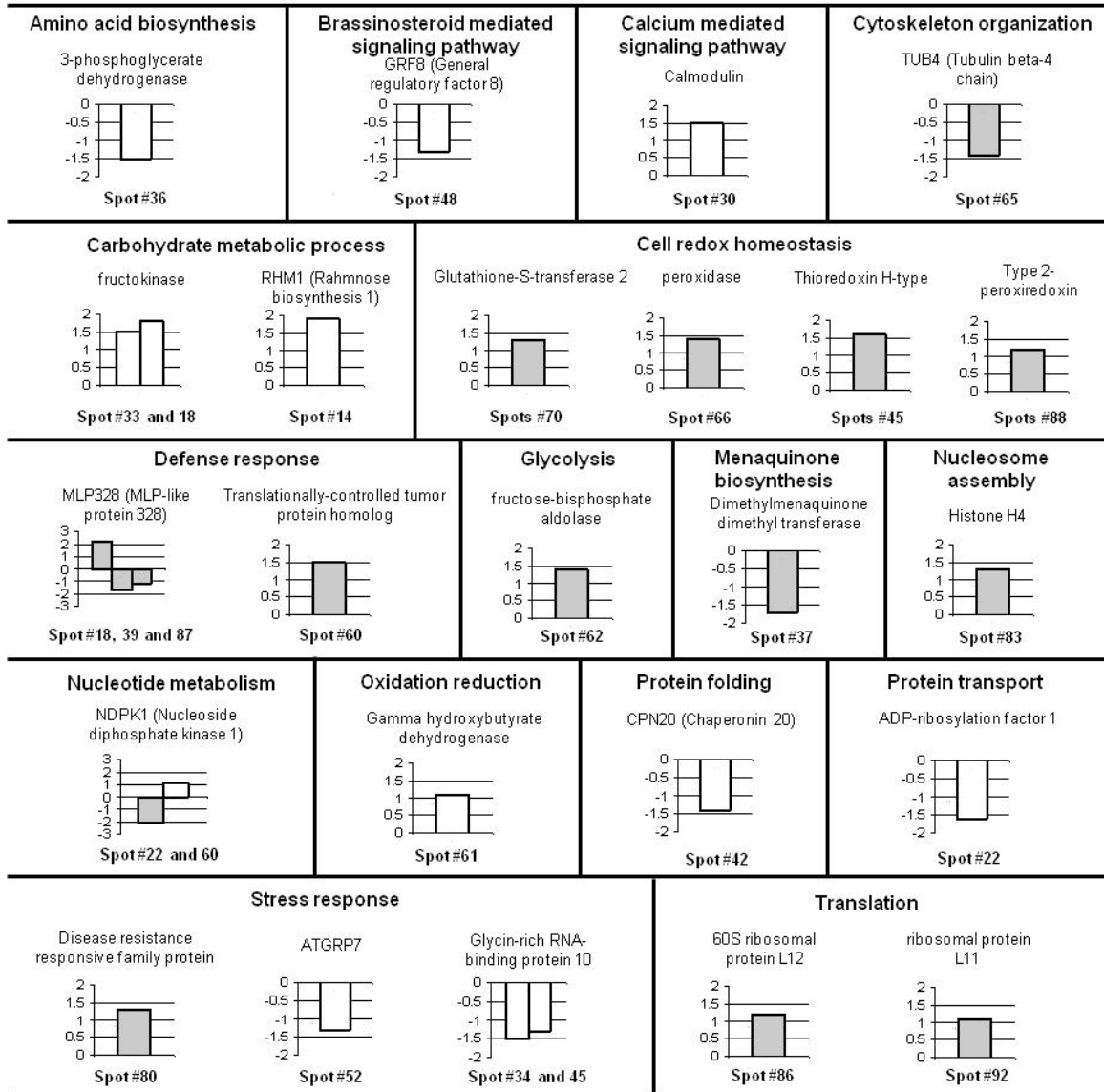
Sophie Alvarez, Ashley Galant, Joseph M. Jez and Leslie M. Hicks

**Redox-regulatory mechanisms induced by oxidative stress in *Brassica juncea*
roots monitored by 2-DE proteomics**

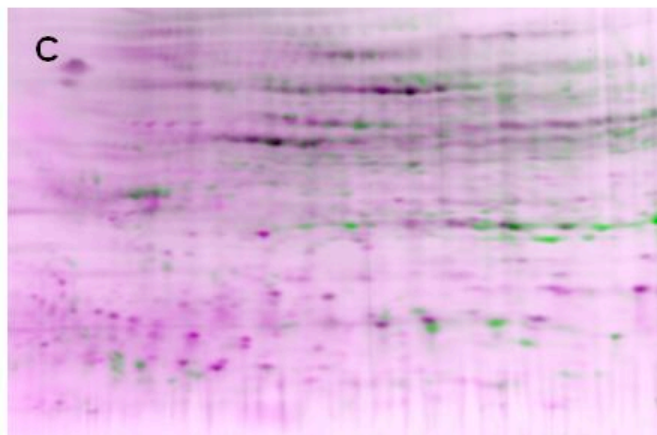
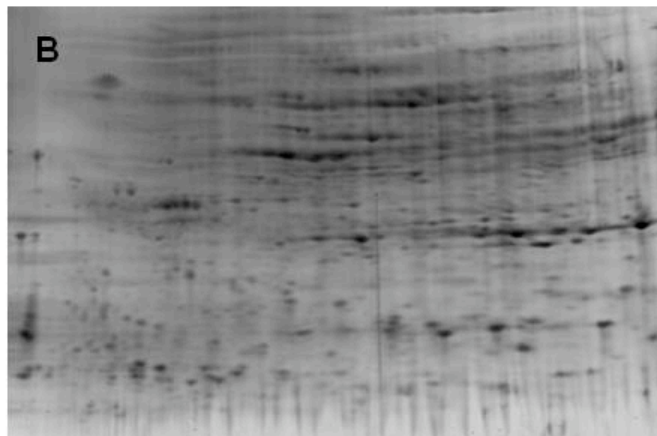
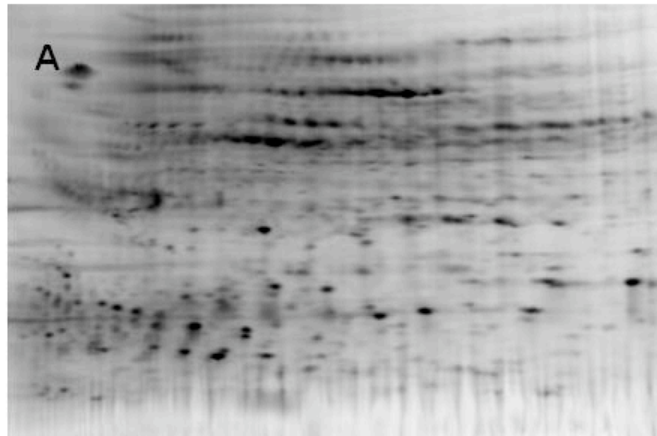
Supplemental Figure 1. Differential Protein Expression Changes. Proteins with a change in abundance were grouped by biological function with fold change in expression shown in the bar graph. White and grey bars correspond to H₂O₂ and BSO treatments, respectively.

Supplemental Figure 2. Example of 2D-gel images after IAF-labeling (A) and Sypro staining (B). The two pictures were overlapped using SameSpots (C) with color pink and green representing IAF and Sypro, respectively.

Supplemental Figure 1



Supplemental Figure 2



IAF	H2O2	114	g i 15235288	PAC1 (20S proteasome alpha subunit)	AAYGANNAAQASLIK_D0VALGEK TTFSPFGR	60	52	45	203	35.1	310.4	13.2	1.5	0.012
IAF	H2O2	116	g i 11131564	Cysteine synthase	KPSMADQK KPSMADQK	69	62	59	928.1	56.3	1407.2	192.1	1.5	0.009
IAF	H2O2	116	g i 110741046	hypothetical protein	KAAASGRVAK_SALMNLEER	47	42	42						
IAF	H2O2	118	g i 116484871	hypothetical protein ARALYDRAFT_902771	AVGNVSIQALGK_TYDLNEK	53	65	61			966.6	125.6	1.5	0.016
IAF	H2O2	118	g i 1110740927	eukaryotic protein synthesis initiation factor 4A	YDANFKEINDDQSK VMSSTSEIWAQ	47	55	42	634.2	83.6				
IAF	H2O2	118	g i 15224101	cystecid aminopeptidase	MLFDICK_VLITDILAR QSDSTWATELV_FVDSQVPLVDSQK SAPVMSKMLELRA	87	51	59						
IAF	H2O2	120	g i 130691626	MITHC70 (mitochondrial heat shock protein 70-1); ATP-binding, cytosolic protein binding (clathrinose Italian)	AVYVAVPANDQR_GELGVTRK KCLVDQR SSQGLSDENNR_TTPSVWANGK	49	53	50	3331.6	139.7	2255.9	321.3	-1.5	0.009
IAF	H2O2	123	g i 24421231	ascorbate peroxidase	EGLLOVSVK SYPLVSEDDK	77	41	68	413.1	68.1	608.1	50.4	1.5	0.021
IAF	H2O2	123	g i 15224983	PA22 (20S PROTEASOME SUBUNIT PA22)	AVVEVQVR_LFOVEVFK KPSMADQK KAAEDTMAWK	30	59	68						
IAF	H2O2	125	g i 13447108	GF14 lambda	AADIAADMAPTFPR_DSTLMOLLR KAAEDTMAWK LLEELFSAASESK_YEEMWVNEK	59	54	56	402.5	81.3	578.7	50.4	1.4	0.04
IAF	H2O2	126	g i 113720386	hydroxymethyltransferase ATP synthase	LASASTDLEK_LVNFVLPVAELESK	83	109	48	1145.3	46.1	1645.7	146.8	1.4	0.003
IAF	H2O2	127	g i 11135129	Thioredoxin H-type	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	75	70	56	925.2	148.6	1325.8	185.8	1.4	0.043
IAF	H2O2	128	g i 15224351	OAS2 (O-ACETYL SERINE (THIOL) LYASE B)	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	72	70	70						
IAF	H2O2	129	g i 15241111	CYS2D2 (CYSTEINE SYNTHASE D2)	RLKGMLEK_LVVVFRGGR	46	50	42						
IAF	H2O2	129	g i 297794887	molybdenum cofactor sulfuryase family protein	RESLVAV_LAVVELEPK	46	55	66	242.5	18.9	337.7	60.1	1.4	0.006
IAF	H2O2	129	g i 22331875	OAS2 (O-ACETYL SERINE (THIOL) LYASE ISOFORM C)	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	72	70	70						
IAF	H2O2	129	g i 181176557	annexin-like protein	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	47	51	44						
IAF	H2O2	129	g i 1022973	fructokinase	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	72	70	70						
IAF	H2O2	132	g i 11135129	Thioredoxin H-type	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	50	64	44	1650.3	86.6	2236.3	158.7	1.4	0.004
IAF	H2O2	141	g i 297849222	cathartin 2	FVLSAFFFESNK_GQISEVFR LAVLDFEYK_LLESDVDQK	65	57	49	7732.6	347.8	9263.7	356.4	1.2	0.006
Sypro	BSO	9	g i 169652657	hypothetical protein.CCG1_06970	AVLDFEYK_LLESDVDQK	44	49	44						
Sypro	BSO	9	g i 15232888	6-phosphogluconate dehydrogenase family protein	FLSLQVDR_LPVNVOAQR KPSMADQK_VIIMLEK	75	75	75	157.3	44.7	57.2	20.5	-2.8	0.017
Sypro	BSO	9	g i 15237422	PROTEIN SYNTHASE I	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	56	60	60						
Sypro	BSO	17	g i 18339899	lipid-associated family protein	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	49	53	53	104.5	13.6	48.5	9.5	-2.2	0.004
Sypro	BSO	18	g i 18379240	MUP328 (MUP-LIKE PROTEIN.328)	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	44	62	62	357.1	71.4	763	146.7	2.1	0.01
Sypro	BSO	22	g i 19570344	nucleoside diphosphate kinase 1	IGAINPAASEFGR KIGATNPAASEFGR	93	54	47	400	75	198.9	43.9	-2	0.013
Sypro	BSO	29	g i 18424620	TUB2 (Tubulin beta-2)	AVLDFEYK_LLESDVDQK LAVLDFEYK_LLESDVDQK	61	61	61						
Sypro	BSO	29	g i 15219345	ATM24 (METACASPASE 4)	AVLDFEYK_LLESDVDQK LAVLDFEYK_LLESDVDQK	51	59	59	982.4	176.1	530.7	27.7	-1.9	0.006
Sypro	BSO	29	g i 32967699	S-adenosyl-L-homocystein hydrolase	AVLDFEYK_LLESDVDQK LAVLDFEYK_LLESDVDQK	64	64	64						
Sypro	BSO	29	g i 11214474	elongation factor-1 alpha	AVLDFEYK_LLESDVDQK LAVLDFEYK_LLESDVDQK	53	63	63						
Sypro	BSO	37	g i 15232963	dimethylmenaquinone methyltransferase family protein	AVLDFEYK_LLESDVDQK LAVLDFEYK_LLESDVDQK	48	48	50	306.2	36.1	175.4	49.2	-1.7	0.28
Sypro	BSO	43	g i 15233272	TPI (TRIOSEPHOSPHATE ISOMERASE)	AVLDFEYK_LLESDVDQK LAVLDFEYK_LLESDVDQK	61	46	60	396.3	58.5	644.2	131.5	1.6	0.027
Sypro	BSO	43	g i 115434516	Os01g0147900	AVLDFEYK_LLESDVDQK LAVLDFEYK_LLESDVDQK	65	61	43						
Sypro	BSO	46	g i 11135129	Thioredoxin H-type	RAPVRLAK_VYGVYELLEQR VDDVLEATWAK	45	77	77	477.3	21.9	768.3	205.1	1.6	3.80E-02
Sypro	BSO	47	g i 224981577	14-3-3	DAESTLWAK KDAENILWAK KDAENILWAK	66	66	81						
Sypro	BSO	47	g i 13447104	GF14 omega	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	54	65	61	906.7	106	584.5	82.7	-1.6	0.014
Sypro	BSO	47	g i 18395103	GRF10 (GENERAL REGULATORY FACTOR 10)	AVVEVQVR_LFOVEVFK KPSMADQK LLEELFSAASESK	65	65	65						

Sypro	BSO	47	g j 297811739	CFE1 (GENERAL REGULATORY FACTOR 5)	DSTLMQLLR EENVYLAK SADZADLPTFRPR	NILSVAVK	49	65	54												
Sypro	BSO	50	g j 14423528	putative fucokinase	GGDFECHM AGILR		110	71	53												
Sypro	BSO	50	g j 1429207	annexin	XSGDFRMP VYR		67	60	67												
Sypro	BSO	50	g j 14422255	enoyl-facyl-carrier protein reductase	IFSGGGSMSAK VLDGSLMEK		52	48	67												
Sypro	BSO	60	g j 20140684	Transcriptionally-controlled tumor protein, 102kDa	LGEPVYDK VVDVDFR			43	47												
Sypro	BSO	63	g j 11034734	myrosinase	GEFRLEVVK GYAEIDAFGR	IGPVAITR	58	59	58												
Sypro	BSO	63	g j 217074426	unknown	YAGMLEYDGEK YENDWQVK		83	59	44												
Sypro	BSO	66	g j 18424620	TUB2	AMLDPEPTMDSLR LAWLPPFR	PFQALNSDLR VSEQTFMR	66	51	63												
Sypro	BSO	66	g j 15241472	TUB4 (tubulin beta-4 chain)	FDYFNASGSK MAMTFEYPSK		66	51	66												
Sypro	BSO	66	g j 15241472	TUB4 (tubulin beta-4 chain)	FDYFNASGSK MAMTFEYPSK		66	50	57												
Sypro	BSO	66	g j 166198115	peroxidase precursor	VAEYDDEYDTPK IAASLR		89	76	63												
Sypro	BSO	67	g j 297811069	inorganic pyrophosphatase family protein	AMNEVWDEDTPK IHETNESWAK	IVMSLDDPK	51	66	70												
Sypro	BSO	67	g j 297841835	predicted protein	IEAVFEPK IEAVFEPK		70	50	51												
Sypro	BSO	70	g j 1790095	glutathione S-transferase 2	LATVDVYFVAR FALDNLK	VKAEEDLKK FVAKDSSEYTK	83	46	39												
Sypro	BSO	72	g j 4828472	type 2 peroxidoxin	VFEDNLGR VPLNAQTR		70	76	46												
Sypro	BSO	72	g j 15232963	dimethylmenaquinone methyltransferase family protein	ONVDLSTYK YAPTTSPLEK			59	47												
Sypro	BSO	72	g j 205364142	glutathione peroxidase				54	54												
Sypro	BSO	77	g j 15237947	UTP-glucose-1-phosphate uridylyltransferase, putative / UDP-glucose pyrophosphorylase, putative / UGPase, putative [Arabidopsis thaliana], g j 2585448 sp j P2751.1 UGP1_AR ATH RecName: Full=UTP-glucose-1-phosphate uridylyltransferase 1; Alternative names: UGPase 1; pyrophosphorylase 1; Short-UGPase 1; Short-UDGP 1; g j 13430864 gb AAK25954.1 AF36024 4_1 putative UDP-glucose pyrophosphorylase [Arabidopsis thaliana]; g j 15237947 gb AAK32829.1 AF36181 6_1 AT5G17310 MKP11_16 [Arabidopsis thaliana]; g j 1017076 dbj BAB10518.1 UDP-glucose pyrophosphorylase [Arabidopsis thaliana]; g j 4532836 gb AAK4100.1 putative UDP-glucose pyrophosphorylase [Arabidopsis thaliana]	FDWAKVAVPR SFDNLVQR VADEYVYPSK	LVEADALK SIPSVELDLSK	79	51	69												
Sypro	BSO	77	g j 17939849	mitochondrial F1 ATP synthase beta subunit	ENNSFGQLDGK FOANSEVALLGR	IMNVLEPIER		48	48												
Sypro	BSO	77	g j 11678286	unknown	FMGGSAVLGAAK TEANNDAEGR	YGLTLLVYER	56	49	68												
Sypro	BSO	77	g j 297827139	hypothetical protein	FMGGSAVLGAAK YONAFK	GLTDSGQYNR	98	69	47												
Sypro	BSO	79	g j 15237947	UTP-glucose-1-phosphate uridylyltransferase, putative	FDWAKVAVPR SFDNLVQR	LVEADALK SIPSVELDLSK	80	56	62												
Sypro	BSO	79	g j 11547253	Os07g0574800	LISQMSLTSLR LISQMSLTSLR	SLDERITYTLNAR	54	54	42												
Sypro	BSO	79	g j 11678286	unknown	FMGGSAVLGAAK FMGGSAVLGAAK	TEANNDAEGR	61	69	61												
Sypro	BSO	79	g j 297827139	hypothetical protein	AGAVSGLYKAVGNWNIIPALIK TOLNIPKEINDDGSK		61	61	61												
Sypro	BSO	82	g j 13447108	GF14 lambda	AAEDTQWYK LLEENLFSAAASEK		61	67	61												
Sypro	BSO	82	g j 13447104	GF14 omega	DAENTLSAYK EEFYWAK	KDAENWLSAYK	47	59	78												

Sypro	BSO	82	g 1115446909	O50290503000	DSTLIMOLLR TVREELTYVEER DSTLIMOLLR EFTVYMAK SADIDANLAPLPHRR VADDOGDELVEER VHEEMFENIEK VLAEEK	59 51 58 57 51 125 51 56 55		1652.3	79.6	1297	194.3	-1.3	0.048	
Sypro	BSO	82	g 118411901	GRF2(GENERAL REGULATORY FACTOR 2)	DSTLIMOLLR EFTVYMAK SADIDANLAPLPHRR VADDOGDELVEER VHEEMFENIEK VLAEEK	59 49 69								
Sypro	BSO	82	g 1224981569	1433-1	DSTLIMOLLR EADOSLEAK VADDOGDELVEER VLAEEK	59 49 69								
Sypro	BSO	83	g 110171851	histone H4	DNIGGKPKNAK IFLENNIR TVAMDVYALK	53 49		323.6	9.7	411.6	47.5	1.3	0.025	
Sypro	BSO	86	g 15228098	60S ribosomal protein L12 (RPL12A)	EPILGAPK VTGVEVGAASSLAFK	44		1093.04	93.2	1324.8	100.8	1.2	0.041	
Sypro	BSO	87	g 18379240	MLP328 (MLP-LIKE PROTEIN 328)	ITWYKAA VADDOGDELVEER VLAEEK	42 42		3471.6	135.4	2895.3	283	-1.2	0.038	
Sypro	BSO	88	g 14828472	Type 2 peroxiredoxin	HALGUDLAK FVADDSGVEYTK	50 65		733.3	14.3	875.9	40.8	1.2	0.004	
IAF	BSO	9	g 17110985	isopenitryl pyridoxal phosphate-dimethylol	DRELEIRNALGVR GTLGEAVDMK	46 48		172.8	80.1	602.9	223.5	3.5	0.023	
IAF	BSO	11	g 14609471	stromal ascorbate peroxidase	FDSNYTEK VIELEK	75 43		682.9	334.8	206.6	96.5	3.1	0.037	
IAF	BSO	11	g 15240663	adenylate kinase, putative	VLFGRSSGSK VUNFADDSVLEER	42 56								
IAF	BSO	13	g 1297816638	TP1 (TRIOSEPHOSPHATE ISOMERASE)	NUSADWAATR VAVAJAGSK	52 62		129	62.5	404.6	113.9	3.1	0.037	
IAF	BSO	14	g 11732572	beta-glucosidase	FGYDYDK NUNTDNFR	60 41		772.6	203	255.7	198.4	-3	0.046	
IAF	BSO	30	g 261343270	O-acetylserine (thiol) lyase isoform A6	EDGYS2SGTGITGAK IGFMSIDNEK LVVMPFFGR	60 44 46		328	82.6	710.1	78.9	2.2	0.009	
IAF	BSO	30	g 14423528	putative fructokinase	AFGGARNAWASR LLVTLGSK	68 121								
IAF	BSO	46	g 15224983	PA22 (20S PROTEASOME SUBUNIT PA22)	FVEEVGVVR VGLGLATGADR	54 71		104.9	9	106.1	47.2	1.9	0.017	
IAF	BSO	46	g 15221463	coatomer protein epsilon subunit family protein / COPE family protein	ASAENFR LLMYLSPEIK	83 63								
IAF	BSO	51	g 11135129	Thioredoxin H-type	FAPFVLEAK VVDRELVAK	65 66		52.3	9.5	94.3	11.3	1.8	0.008	
IAF	BSO	51	g 110171851	histone H4	EGSINDVAADTK NUSADWAATR	52 43		1311.8	146.9	729.2	19.2	-1.8	0.001	
IAF	BSO	52	g 297816638	predicated protein	ALMESFVGGK VLDREGR	81								
IAF	BSO	61	g 297816638	predicated protein	EGSINDVAADTK VLDREGR	81								
IAF	BSO	61	g 11135129	Thioredoxin H-type	EGSINDVAADTK VLDREGR	81								
IAF	BSO	61	g 110171851	histone H4	EGSINDVAADTK VLDREGR	81								
IAF	BSO	61	g 297828984	hypothetical protein	VGMAAEEAK VADDOGDELVEER VLAEEK	77 70 61		420.8	86.1	729.3	41.1	1.7	0.011	
IAF	BSO	65	g 15224983	PA22 (20S PROTEASOME SUBUNIT PA22)	EGSINDVAADTK VLDREGR	81								
IAF	BSO	65	g 15233288	PA21 (20S proteasome alpha subunit C1)	EGSINDVAADTK VLDREGR	81								
IAF	BSO	65	g 2421231	ascorbate peroxidase	EGSINDVAADTK VLDREGR	81								
IAF	BSO	66	g 15242459	mHS70-2 (HEAT SHOCK PROTEIN 70)	KAVYVAPNDVQR EEDVAADTK VLDREGR	92 50								
IAF	BSO	66	g 2654210	heat shock 70 protein	KAVYVAPNDVQR EEDVAADTK VLDREGR	92 50		609	7.1	361.5	70	-1.7	0.009	
IAF	BSO	66	g 2483122	X-type proton ATPase catalytic subunit A	QAVNFINIEGK ITPSVAARNPK LGDLYR	50 68 61 71								
IAF	BSO	72	g 15224983	PA22 (20S PROTEASOME SUBUNIT PA22)	EGSINDVAADTK VLDREGR	81								
IAF	BSO	75	g 15224982	GSTT10	EGSINDVAADTK VLDREGR	81		1172.5	171.4	1931.4	377.2	1.6	0.024	
IAF	BSO	78	g 297816638	TP1 (TRIOSEPHOSPHATE ISOMERASE)	NUSADWAATR VAVAJAGSK	47 48		182.4	16.4	286.6	43.6	1.6	0.008	
IAF	BSO	81	g 297816638	predicated protein	FVGSVANK NUSADWAATR	52 62		171.7	23.2	106.1	6.8	-1.6	0.006	
IAF	BSO	81	g 15219257	PAB1 (PROTEASOME SUBUNIT PAB1)	EGFEIGSSK LYGEPYVTVQR	68 54		1033.8	222.3	688.1	44	-1.5	0.033	
IAF	BSO	83	g 87294807	glutathione S-transferase	RONSISAEISPK ONDFPSK	40 55		282.8	63.6	431.5	54	1.5	0.043	
IAF	BSO	86	g 18339899	lipid-associated family protein	FAPFVLEAK VVDRELVAK	41 49								
IAF	BSO	93	g 11135129	Thioredoxin H-type	FAPFVLEAK VVDRELVAK	41 49		663.9	27.7	986.9	24.5	1.5	0.001	
IAF	BSO	93	g 110171851	histone H4	EGSINDVAADTK VLDREGR	77 70								
IAF	BSO	94	g 15231715	fructose-bisphosphate aldohase, putative	EGSINDVAADTK VLDREGR	81 86		1476.2	204.3	2179.8	357.2	1.5	0.04	

IAF	BSO	97	g 15235213	callicofin-GAA 3-O-methyltransferase, putative	IPVLELNK LAVDYNR	45	57	2185.4	281.9	1495.4	150.3	-1.5	0.017
IAF	BSO	97	g 15224983	PAA2 (20S PROTEASOME SUBUNIT PAA2)	ATGITSQVR ATEIVQVR LQSLATGADQSR	88 59	46						
IAF	BSO	98	g 15222633	disease resistance-responsive family protein / dirigent family protein	DLNVGSGTDFPNSR GVPEFDITREQK	52	77	436.2	37.6	636.6	107.3	1.5	0.032
IAF	BSO	101	g 15239146	ATNADP-ME2 (NADP-MALIC ENZYME 2)	GVVVYDQSR YVAVMLQER	61 48							
IAF	BSO	101	g 15229559	HSP90 (Heat shock protein 90)	GVVQVQVAK SVAELSTSDYDEK	68 69		3239.5	112.9	2277.9	342.1	-1.4	0.018
IAF	BSO	101	g 1414103	myrosinase, thioglucoside glucosyltransferase	GVVGTDPGR IPVMTIR	47	59						
IAF	BSO	101	g 15010652	A1g09760F21M12_17	GVVQVQVAK LQDLQLIK	55	53	494.4	42.2	666.8	95.6	1.3	0.039
IAF	BSO	104	g 16395103	GRF10 (GENERAL REGULATORY FACTOR 10)	ILSDECK YLAFFSSGAKR	40	55	980.3	100.7	765.7	6.4	-1.3	0.013
IAF	BSO	106	g 15241266	MAB1 (MAB1-BOU)	LAEGRAEVNLR VAPYSAQAR	78	55	115.9	7.5	148.6	9.1	1.3	0.009
IAF	BSO	106	g 14423528	putative fructokinase	APGARAVAVNSR LLVTLDEK TALAELVLR	102	60						
IAF	BSO	107	g 15231176	ATPO (ATP SYNTHASE D CHAIN, MITOCHONDRIAL)	AFREVTLQTK IPALVLEK	57	59						
IAF	BSO	108	g 15236385	SBP1 (selenium-binding protein 1)	AVVQVFEQSGIDNK YVLRSLSR	47	63						
IAF	BSO	108	g 118420132	vacuolar ATPase subunit B	AVVQVFEQSGIDNK GVPRKLVLSLR	75	50	1557.8	159.5	1217.4	56.6	-1.3	0.02
IAF	BSO	108	g 29779542	hypothetical protein	AVVQVFEQSGIDNK TSSVAGDVALEK	47	64						
IAF	BSO	113	g 297824877	ARFALYDRAFT 492381	AVVQVFEQSGIDNK AVVQVFEQSGIDNK	49	51						
IAF	BSO	113	g 297816638	glutathione S-transferase predicted protein	AVVQVFEQSGIDNK VLAALYER	46	52	1106.3	60.1	962.9	22.9	-1.2	0.012
IAF	BSO	114	g 132285914	putative dehydroascorbate reductase	AVVQVFEQSGIDNK VLAALYER	69	64						
IAF	BSO	114	g 15233272	ATCTMC (CYTOSOLIC TRIOSE PHOSPHATE ISOMERASE)	AVVQVFEQSGIDNK VSAVLSIAPK	75	59	1606.4	76.2	1399.1	77.9	-1.1	0.03
IAF	BSO	114	g 15219257	PAB1 (PROTEASOME SUBUNIT PAB1)	AVVQVFEQSGIDNK EAGSTNDVAQTK VNSADVAATR EGFEGEISK KLP SILVDEASVCK	66 59 46	58						

* fold change indicates the direction of the change of the protein expression or oxidation in the treated sample.
Spot highlighted in yellow are spots for which more than one protein were identified.

Supplemental Table 2. List of the 52 proteins identified as single protein classified according to their biological process.

Strain	Treatment	Spot #	Protein accession #	Protein name	Biological pathway	Average of normalized volume in Control	Standard deviation	Fold Change*	P-value
IAF	H2O2	2	gi11417171	3-oxopropionyl-CoA:2-oxoglutarate:5-oxopropionyl-CoA ligase (cytoplasmic)	Amno acid biosynthesis	3081	127	24.3	5.5E-012
IAF	H2O2	19	gi15226066	5-methylcrotonyl-CoA:pyruvate carboxylase	Amno acid biosynthesis	7597	167	25.6	4.98E-06
IAF	H2O2	87	gi111131584	Cysteine synthase	Amno acid biosynthesis	3672	37.2	100.3	1.7E-009
IAF	H2O2	126	gi11523176	ATPQ (ATP SYNTHASE D CHAIN, MITOCHONDRIAL)	ATP synthesis coupled proton transport	1159	115.9	148.6	9.1E-009
IAF	H2O2	126	gi11523176	hydrogen-fueled ATP synthase	ATP synthesis coupled proton transport	1153	115.3	146.7	1.4E-009
IAF	BSD	104	gi18395103	GRF10 (GENERAL REGULATORY FACTOR 10)	Branched-chain amino acid metabolism	4944	48.2	96.6	1.3E-039
IAF	H2O2	125	gi13447108	GF 14 lambda	Branched-chain amino acid metabolism	4025	61.3	57.7	1.4E-046
IAF	H2O2	14	gi11732572	beta-glucosidase	Branched-chain amino acid metabolism	7726	203	255.7	199.4
IAF	H2O2	75	gi120760724	chloroplast malate dehydrogenase	Carbohydrate metabolic process	2716	42.3	47.4	1.6E-024
IAF	BSD	83	gi18224807	glutathione S-transferase	Cell redox homeostasis	10358	222.3	868.1	37.2E-004
IAF	H2O2	79	gi15222163	DHA22	Cell redox homeostasis	1826	182.6	114.2	23.9
IAF	H2O2	95	gi1207624977	glutathione S-transferase	Cell redox homeostasis	2029	17.2	332.5	40.4
IAF	H2O2	127	gi11135129	Thioredoxin H-type	Cell redox homeostasis	5252	146.6	136.8	1.4E-045
IAF	H2O2	132	gi111351293	Thioredoxin H-type	Cell redox homeostasis	1853	88.8	228.3	158.7
IAF	BSD	13	gi129761638	TPP (TRIOSEPHOSPHATE ISOMERASE)	Glycolysis	129	62.5	113.9	3.1E-037
IAF	BSD	78	gi129761638	TPP (TRIOSEPHOSPHATE ISOMERASE)	Glycolysis	1824	18.4	298.6	43.8
IAF	BSD	94	gi13237175	fructose-bisphosphate aldolase, putative	Glycolysis	14762	204.3	2179.8	387.2
IAF	H2O2	37	gi15221107	enolase, putative	Glycolysis	1995.5	344	902.9	71.4
IAF	H2O2	84	gi1207619638	TPP (TRIOSEPHOSPHATE ISOMERASE)	Glycolysis	7241	58.6	428.6	102.9
IAF	H2O2	141	gi1207649222	carefolin 2	Protein folding	7732.6	347.8	935.7	358.4
IAF	BSD	72	gi15224983	PAZ2 (20S PROTEASOME SUBUNIT PAZ2)	Proteolysis	4629	61.3	766.3	127.1
IAF	H2O2	4	gi18200712	CLP2 (CLP PROTEASE PROTEOLYTIC SUBUNIT 2)	Proteolysis	4656	27.2	122.1	64.7
IAF	H2O2	114	gi15232268	isopenicillin N synthase	Proteolysis	203	35.1	310.4	13.2
IAF	BSD	9	gi17110286	isopenicillin N synthase	Proteolysis	1728	80.1	802.9	223.5
IAF	BSD	98	gi15222633	disease resistance-responsive family protein / flagellin protein	Secondary metabolism	4382	37.6	636.6	107.3
IAF	H2O2	120	gi136061626	MHS2701 (mitochondrial heat shock protein 70-1) ATP-binding / unfolded protein bind	Stress response	3316	139.7	2265.9	321.3
IAF	BSD	96	gi13398999	lipid-associated family protein	Stress response	2828	63.6	1431.5	54
Sypro	H2O2	38	gi18394525	PGD1 (3-PHOSPHOGlycerate dehydrogenase)	unknown	1656	86.4	723.9	47.9
Sypro	H2O2	48	gi13038122	GRF6 (GENERAL REGULATORY FACTOR 6)	Amno acid biosynthesis	9501	32.9	723.6	59.1
Sypro	H2O2	30	gi15221264	clatrodin, putative	Branched-chain amino acid metabolism	9465	115.6	1461	112.4
Sypro	H2O2	33	gi14423528	putative fucosylase	Calcium-mediated signaling	10147	108.1	1392.2	314.7
Sypro	H2O2	16	gi14423528	putative fucosylase	Carbohydrate metabolic process	37.5	7.3	69.2	4.5
Sypro	H2O2	14	gi105623672	RHM1	Carbohydrate metabolic process	53.4	7.9	101.1	27
Sypro	H2O2	42	gi1207624877	glutathione S-transferase	Cell redox homeostasis	2016.5	170.4	1420.5	164.9
Sypro	BSD	70	gi13790095	glutathione S-transferase 2	Cell redox homeostasis	1416.7	272.9	1908.6	82.3
Sypro	BSD	66	gi186188116	peroxidase precursor	Cell redox homeostasis	1288.9	77	1778.7	317.8
Sypro	BSD	45	gi11135129	Thioredoxin H-type	Cell redox homeostasis	477.3	21.9	768.3	205.1
Sypro	BSD	88	gi14228472	Type 2 peroxidase	Cell redox homeostasis	733.3	14.3	875.9	40.8
Sypro	BSD	18	gi18379240	MLP28 (MLP-LIKE PROTEIN 28)	Defense response	357.1	71.4	763	140.7
Sypro	BSD	67	gi18379240	MLP28 (MLP-LIKE PROTEIN 28)	Defense response	3471.6	138.4	2966.3	293
Sypro	BSD	60	gi12014084	Transcriptionally-controlled locus protein homolog	Defense response	218.5	19.2	317.3	63
Sypro	BSD	63	gi115171851	Isotretinoin 14	Nucleosome assembly	523.8	9.7	411.6	47.5
Sypro	H2O2	60	gi13890069	NADPK1	Nucleotide metabolism	1838.9	88.9	2017	54.4
Sypro	H2O2	22	gi18570344	nucleoside diphosphate kinase 1	Nucleotide metabolism	400	79	199.8	43.9
Sypro	H2O2	61	gi1375068	gamma hydroxybutyrate dehydrogenase	Oxidation reduction	6246	22.4	680.9	165
Sypro	BSD	37	gi105673020	ADP-ribosylation factor 1	Protein transport	4614	45.6	280.3	65.9
Sypro	BSD	52	gi16222963	dimethylmenaquinone methyltransferase family protein	Secondary metabolism	3082	38.1	462	175.4
Sypro	H2O2	86	gi15228605	CR2 (COLD, CIRCADIAN RHYTHM, AND RNA BINDING 2)	Stress response	2871.1	357.9	2244.3	179.2
Sypro	BSD	17	gi18398939	lipid-associated family protein	unknown	1045	1093	1324.8	100.8
Sypro	BSD	17	gi18398939	lipid-associated family protein	unknown	1045	13.6	48.5	9.5

PROTEIN EXTRACTION AND LABELING PROTOCOL

Plant Growth

Wildtype *B. juncea* seeds were obtained from stocks maintained at the Donald Danforth Plant Science Center, and allowed to germinate in a growth chamber at 22°C, 200 mmol/m² light intensity, 50% relative humidity during a 16-hour light/8-hour dark cycle. Once their second set of true leaves began to emerge (typically two-three weeks after planting), the seedlings were transplanted to one gallon pots and moved to a greenhouse with the same light/dark cycle. The plants were grown normally until they began to flower, at which point they were separated into groups: control, H₂O₂-treated, and BSO-treated. H₂O₂-treated and BSO-treated plants were pot-watered with 2 L of 1 mM H₂O₂ and 2 L of 50 μM BSO respectively, and positioned to allow rapid draining. After 1 hour, an additional 1 L of 1 mM H₂O₂ or 1 L of 50 μM BSO was respectively applied, and the plants again were allowed to drain for an additional hour. Control plants were irrigated with distilled water, and were otherwise treated identically. At the end of 2 hours, the plant roots were rapidly washed to remove excess soil, flash frozen in liquid nitrogen, and stored at -80°C

Protein Labeling

For the H₂O₂, BSO, and control treatments, soluble protein extraction was completed in triplicate using tissue from three different plants, and all steps were carried out at 4°C unless otherwise indicated. Approximately 800 mg of root tissue, or 400 mg

of leaf tissue were ground to a fine powder using liquid nitrogen and a mortar and pestle. The tissue was suspended in extraction buffer [100 mM Tris-HCl, pH 8.0; 100 mM NEM; 1% CHAPS; 1% plant protease inhibitors (Sigma, P9599)] to 200 mg/mL, and sonicated for 3 x 15 seconds. In between each sonication the tissue was vortexed briefly and allowed to sit on ice for 30 seconds. The samples were then centrifuged for 16.1k x g for 15 minutes to precipitate the insoluble debris. Following centrifugation, the supernatant was mixed with 4 volumes of pre-chilled methanol and stored on ice. In most cases, protein precipitation was observed almost immediately. After 30 minutes, the samples were again centrifuged at 16.1k x g for 15 minutes, with the supernatant being discarded once the spin was complete. The protein pellet was washed twice more (2 x 30 minutes) with the same volume of methanol, and centrifuged at 16.1k x g for 5 minutes after each wash. During the washing steps, care was taken to periodically disrupt the pellet by hand (using a sterile pipette tip) or by vortexing to ensure complete removal of excess NEM. After the final spin, the pellet was air-dried for several minutes, and then resuspended in 150 μ L of reduction buffer [50 mM Tris-HCl, pH 8.0, 7 M urea, 2 M thiourea, 50 mM DTT (added in just prior to use)] per 200 mg of starting tissue. The suspension was incubated for 15 minutes at room temperature, and then re-precipitated with 4 volumes of pre-chilled methanol. As described above, the protein was washed 3 times (3 x 30 minutes) with the same volume of methanol and centrifuged after each wash (1 x 15 minutes; 2 x 5 minutes), with care taken to disrupt the pellet. After the final spin, the pellet was air-dried briefly, and resuspended in 150 μ L of labeling buffer [40 mM HEPES, pH 7.5; 50 mM NaCl; 200 μ M IAF (added in just prior to use)] per 200 mg of starting

tissue. Due to the light sensitivity of IAF, this and all subsequent steps were carried out under dim light. The suspension was incubated for 10 minutes at room temperature, and then re-precipitated with 4 volumes of pre-chilled methanol. Again, the protein was washed 3 times (3 x 30 minutes) with the same volume of methanol and centrifuged after each wash (1 x 15 minutes; 2 x 5 minutes), with care taken to disrupt the pellet. After the final spin, the pellet was air-dried briefly, and resuspended in a small (<60 μ L per 200 mg of starting tissue) volume of destreak buffer (GE Healthcare). The protein concentration was then determined by CB-X assay (G-Biosciences).

2D-SDS-PAGE

For 2D-SDS-PAGE, 200 μ g of extracted protein was resuspended to a total volume of 180 μ L in destreak buffer and absorbed into a pH 4-7 gel strip (Bio-rad) overnight. Isoelectric focusing in the first dimension was carried out at room temperature in a Proteon IEF cell using a four-step method: 1) 250 V(volts), linear increase, 30 minutes; 2) 500 V, linear increase, 1 hour; 3) 8000 V, linear increase, 2.5 hours; 4) 8000 V, rapid increase, 35,000 Vhours. Separation in the second dimension was by molecular weight, and achieved using a standard gel box run at 150 V until the dye front reached the end of the gel. First dimension gel strips were secured in place relative to the second dimension gel using 1 mL of agarose. Following the second dimension separation, gels were removed from their cassettes and imaged (Ex: 488 nm: Em: 520 nm) using a Typhoon 9410 variable mode imager (Amersham Biosciences) to detect IAF-labeled proteins. After imaging, gels were bathed for 30 minutes with 100 mL of fixing solution

[10% methanol 7% glacial acetic acid] using an orbital shaker. The fixing solution was then poured off, and gels were then bathed overnight in 50 mL of Sypro Ruby protein stain (Bio-rad). The next day, the protein stain was poured off, and the gels were again bathed for 30 minutes with 100 mL of fixing solution. Following a washing step with MilliQ water to remove any excess stain, the gels were imaged again (Ex: 457 nm; Em: 610 nm) to detect total protein. For all gel replicates within a given set or sets to be compared, the same laser intensity (400 V for IAF images and 650-800 V for Sypro images) was used.

Spot Analysis and Excision

Replicate gels images were aligned using Progenesis SameSpots (Nonlinear Dynamics). Further alignment of replicate control gel and replicate ozone treatment gel images was carried out for each for each pairwise comparison. In order to quantify expression and thiol composition differences between the control and treated samples, spot volume (as a function of intensity) was calculated and normalized for each spot in the aligned images. Those spots that different significantly in volume (ANOVA, $p < 0.05$) between the averaged control and ozone treatment gels were then marked for excision.

Excision of significant spots from their respective gels was performed using a Gelpix System (Genetix) under high humidity (>85%) to prevent gel distortion or tearing. Gel plugs were dehydrated with 200 μ L of acetonitrile (ACN) for 15 minutes at 900 rpm (revolutions per minute) using a room-temperature table-top shaker. The ACN was then removed, and, to remove the Sypro Ruby stain, the plugs were washed 5 x 15 minutes,

900 rpm, with 200 μL of 50mM NH_4HCO_3 , 50% ACN, with the liquid discarded after every wash. Following the last $\text{NH}_4\text{HCO}_3/\text{ACN}$ bath, the plugs further were washed for 5 minutes, 900 rpm, with 100 μL of ACN; when this step was complete the liquid was again discarded and the plugs were allowed to air-dry for several minutes. Once dry, the plugs were submerged in 20 μL of trypsin digestion buffer (50mM NH_4HCO_3 containing 6 ng/ μL trypsin) and rehydrated overnight at 37°C. The next morning, 30 μL of 1% formic acid, 2% ACN was added to the digests, which were then shaken for 30 minutes at 900 rpm. Following the wash, the supernatant was collected from each plug and transferred to a new tube. Again, the plugs were shaken for 30 minutes at 900 rpm, this time in 24 μL of 60% ACN. After this final wash, the supernatant was removed and added to that collected during the previous step, and the plugs were discarded. Using a SpeedVac the combined digest from each gel plug was lyophilized to dryness, then finally resuspended in 7 μL of 1% formic acid, 5% ACN. Identification of the proteins contained in each digest was carried out by nano-LC-MS/MS as previously described [Alvarez et. al, 2009].

Alvarez, S., Berla, B., Sheffield, J., Cahoon, R.E., Jez, J.M., Hicks, L.M. (2009). Comprehensive analysis of the *Brassica juncea* root proteome in response to cadmium exposure by complementary proteomic approaches. *Proteomics* 9: 2419-2431.

CHAPTER 4

FROM CLIMATE CHANGE TO PROTEINS: REDOX PROTEOMICS OF OZONE-INDUCED RESPONSES IN SOYBEAN

PREFACE

This chapter describes the application of the dual-labeling methodology developed in chapter 2 to protein extracts exposed to either ambient or elevated tropospheric ozone concentrations. This work was completed in collaboration with the laboratory of Dr. Lisa Ainsworth at the University of Illinois at Urbana-Champaign (UIUC), USDA-ARS; without their expertise and facilities, the experiments described herein would not have been possible.

The SoyFACE Facility

The specific facilities at UIUC are referred to as SoyFACE (Soybean Free Air Concentration Enrichment), and are one of only a handful of FACE sites worldwide. The majority of FACE sites focus on the effects of elevated CO₂ concentrations on plant growth (in these cases the C in FACE actually stands for CO₂ instead of concentration); only SoyFACE and AspenFACE at the Harshaw Experimental Forest in Wisconsin have investigated the effects of tropospheric ozone in addition to CO₂. Regardless of the specific antagonists and/or species under inquiry however, the basic technology behind all FACE-type experiments remains the same. At SoyFACE, soybean plots are surrounded by octagonal rings composed of micropore tubing. The rings are approximately 16 meters in diameter, and are separated from one another in all directions by 100 meters of untreated soybean plants to avoid gas cross contamination. In 2009, 16 rings were in active use, and contained various soybean cultivars exposed to target

concentrations of ozone ranging from ambient (~40 ppb) to 200 ppb. 4 of the 16 rings were not exposed to ozone, but rather to elevated CO₂ at a target concentration of 585 ppm. The ozone used for elevated concentrations is produced on site with dedicated ozone generators, and pumped from the generator housings directly to the various rings, where its rate and direction of diffusion can be directly controlled. Wind direction, wind speed, temperature, humidity and host of other factors are monitored in real time for each ring; these factors will determine the rate at which and direction from which ozone is released so as to maintain the target tropospheric concentration. As described in the thesis introduction, natural ozone concentrations are cyclical, with the highest concentrations observed during the daylight hours and the lowest concentrations at night. At SoyFACE, this natural cycle is mimicked by only running the ozone generators during a 9-hour daytime period, and allowing the rings to settle back to the ambient concentration at night.

Unlike plants grown in a growth chamber or in open-top pots, the plants at the SoyFACE facility are exposed to all of the elements - including rain, hail, extreme temperature fluctuations and insect infestations - that a normal soybean crop would experience. While these competing factors can make the final experimental statistics more difficult to deconvolute, they nonetheless provide a more accurate picture of how the sum of expected elements, including ozone exposure, affects crop yield and protein expression.

Life Stages of a Soybean Plant

As for *Arabidopsis*, the life of a soybean plant has been divided into a series of defined stages [TAIR; <http://www.arabidopsis.org/portals/education/growth.jsp>]. These stages are divided into two sets: the “V” stages which mark periods of vegetative growth, and the “R” stages which chronicle the emergence of the reproductive organs [<http://www.ag.ndsu.edu/pubs/plantsci/rowcrops/a1174/a1174w.htm#Growth>]. An outline of the stages is provided below:

Vegetative Stages

VE - seedling emergence

VC - cotyledons unfold

V1 - first trifoliolate unfolds

V2 - second trifoliolate unfolds

V3 - third trifoliolate unfolds

V4 - fourth trifoliolate unfolds

V5 - fifth trifoliolate unfolds

V6 - 6th trifoliolate unfolds; flowering will begin if it has not already

Reproductive Stages

R1 - the first flower opens

R2 - all flowers are open or have opened

R3 - the first pod develops

R4 - pod development extends to the top nodes of the plant

R5* - seed development begins

R6 - at least one full-size seed is present; Senescence of the lowest leaves begins

R7 - pod browning begins

R8 - 95% of pods are brown.

* Note that vegetative growth continues even after reproductive growth begins. Thus, a plant that is in stage R5 may also be simultaneously in stage V11 or higher.

Author contributions: AG, LMH, and JMJ designed research; AG and RPK performed research; RPK and EAA contributed new reagents/analytical tools; AG analyzed data; and AG, LMH, and JMJ wrote the paper.

Classification: Biological Studies - Plant Biology

**FROM CLIMATE CHANGE TO PROTEINS: REDOX PROTEOMICS OF
OZONE-INDUCED RESPONSES IN SOYBEAN**

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ABSTRACT

Ozone (O₃) is an important atmospheric pollutant with respect to agricultural losses. Although O₃ affects a range of crops, soybean yield is extremely sensitive to this environmental oxidative stress. To understand metabolic alterations in response to chronic O₃ exposure, changes in the total and redox proteomes of soybean plants grown in the field at the Soybean Free Air Concentration Enrichment (SoyFACE) facility under ambient (40 ppb), moderate (60 ppb), and high (115 ppb) O₃ levels were examined. The changes in the total and redox proteomes of soybean leaf tissue exposed to chronic high O₃ levels are more widespread and not the same as those resulting from short-term acute O₃ exposure. Compared to the ambient control, the 115 ppb O₃ leaf sample contained 35 proteins that increased up to 5-fold in expression level, 22 proteins that were up to 5-fold more oxidized without changes in expression levels, and 22 proteins that increased in total expression level and became 2- to 9-fold more oxidized. These changes occur in proteins across carbon metabolism, photosynthesis, amino acid metabolism, specialized metabolism of flavonoids and isoprenoids, signaling & homeostasis, antioxidant responses, protein degradation, and nucleic acid pathways. Our data directly demonstrates that O₃ exposure in plants changes the oxidation states of multiple proteins across metabolic pathways, and may provide a snapshot of metabolic adaptation to long-term field growth under chronic O₃ stress. Understanding how environmental O₃ affects redox-sensitive pathways will aid in the development of crops better adapted to global climate change.

INTRODUCTION

Global climate change and air pollution pose significant challenges to agriculture and food production worldwide (1). In the Northern hemisphere, tropospheric ozone (O_3) is a major pollutant that affects agriculture yields of multiple crops (2-4). Since the 19th century, ground O_3 levels have doubled with tropospheric concentrations in industrialized nations rising 0.5-2.5% per year, and major crop growing regions of the United States, India, and China facing more rapid changes of up to 10% per year (5-7). The current global mean O_3 level of ~60 ppb and higher localized concentrations are already above the established 40 ppb threshold for crop losses (2, 8-9). Climate models predict that mean surface O_3 concentrations may rise 20-25% globally by 2050 with levels in India and south Asia reaching comparable levels by 2020 (10-12). Understanding how crops respond to increasing O_3 pollution (and other environmental stresses) is essential for meeting the growing demands for sustainable food systems as the world faces increasing population, urbanization, and climate changes.

The negative effects of O_3 on crop yield are well documented from both short-term acute exposure studies and long-term chronic free-air concentration enrichment (FACE) experiments (2-4; 12-16). Among major food crops, soybean (*Glycine max*) is one of the most sensitive to atmospheric O_3 levels, which can vary between 50-120 ppb during summer days (17-18). At concentrations as low as 40 ppb, soybean growth and seed yield begin to decrease with even modest changes in O_3 levels significantly reducing crop production. For example, in FACE trials with soybean, a 13 ppb increase in O_3 from 56 to 69 ppb resulted in a 20% decrease in crop yield (13-14). Comparable reductions in

yield occur across multiple soybean varieties, suggesting that breeding for O₃ tolerance may be difficult. Economically, annual crop losses to O₃ damage at current tropospheric levels are estimated at \$2-4 billion in the US and \$3-5.5 billion in China, and will likely increase in the future (12).

As an environmental stress, O₃ acts as an oxidant in crop growth and results in visible necrotic damage, including chlorophyll loss and decreased seed yields in both mass and number (3, 14-21). At the molecular level, proteomic studies of rice, wheat, soy, and poplar exposed to acute, short-term O₃ stress in growth chambers reveal drastic reductions in the major leaf photosynthetic proteins and induction of defense/stress-related proteins (22-28). Although multiple physiological experiments indicate that acute and chronic ozone exposures do not induce the same damage mechanisms in plants (15-16), assessments of proteome changes have not examined crop plants grown in the field under chronic O₃ stress. Moreover, published studies do not probe the possible effect of O₃ on redox-sensitive proteins in plants, as these changes are not observable by standard proteomic methods (29-30).

Recently, we used a differential labeling and mass spectrometry-based approach (**Fig. S1**) to identify plant proteins that respond to changes in redox environment resulting from exogenous and endogenous oxidative stresses (30). Here we employ this method to assess the changes that occur in the total and redox proteomes of soybean in response to growth under chronic elevated O₃ levels in the field. Soybean plants were grown at the Soybean FACE (SoyFACE) facility (U. Illinois/USDA) under ambient (40 ppb), elevated (60 ppb), and high (115 ppb) atmospheric O₃ conditions. Soluble protein extracts from

root and leaf tissues were then isolated for analysis of changes in total and redox-proteomes using two-dimensional gel electrophoresis (2-DE), differential labeling, and nano-LC/MS/MS. The data presented here indicates that the changes in the total and redox proteomes of soybean leaf tissue resulting from chronic exposure to high O₃ levels are more widespread across metabolism than previously reported and are not necessarily the same as those resulting from short-term acute O₃ exposure. In addition, we provide the first direct demonstration that high O₃ exposure in leaf tissue alters the oxidation states of multiple proteins in different biochemical pathways. These changes may play a role in the metabolic adaptation to long-term field growth under chronic O₃ exposure.

RESULTS

Analysis of 2-DE spots in O₃-treated soybean

To identify O₃-responsive proteins in soybean, protein extracts of leaf and root tissue from plants grown at the SoyFACE facility under ambient, 60 ppb, and 115 ppb O₃ for were obtained (16). For each condition, protein extraction was performed in triplicate using tissue from three different plants. Extracted proteins were incubated with N-ethylmaleimide (NEM) to block free thiols, reduced with DTT, and then reacted with 5-iodoacetamidofluorescein (IAF) to label previously oxidized thiols (**Fig. S1**) (30). After 2-DE, gels were imaged for IAF signal, and then stained with SYPRO Ruby and imaged for total protein (**Fig. 1**). Three replicate gels for each of three independent samples were compared pairwise against the ambient gel images, and spots that significantly changed

in signal intensity identified. Across the 8 condition permutations (60 ppb or 115 ppb O₃; root or leaf tissue; and SYPRO or IAF), a total of 1455 significant spots were detected, of which 277 were differentially expressed and/or oxidized (**Table S1**). Spots were excised, trypsin digested, and analyzed by nano-LC/MS/MS. The resulting spectra were searched against the NCBIInr database using an in-house version of MASCOT (**Tables S2-S5**). From this search, 57 spots contained a single protein match and 83 spots were identified as containing two or more proteins (**Table 1**). The 115 ppb O₃ leaf sample had the largest numbers of identified proteins that changed in expression and/or oxidation state.

Identification of differentially expressed/oxidized proteins

In the identified spots, a total of 159 proteins were found to change in total expression and/or oxidation state (**Fig. 2A**). Of those proteins, 55, 27, 9, and 30 were unique to the 115 ppb O₃ leaf, 115 ppb O₃ root, 60 ppb O₃ leaf, and 60 ppb O₃ root samples, respectively. A further 38 proteins were found to change in multiple tissue-O₃ treatment combinations. Within each of the four tissue-ozone combinations varied numbers of proteins changed in total expression and/or oxidation state (**Figs. 2B-2E**). For example, the 115 ppb O₃ leaf sample contained a total of 79 unique proteins, of which 35 changed in total expression, 22 displayed altered oxidation state, and 22 changed in both expression and oxidation. For each protein identified, total expression level and/or oxidation state either increased or decreased in the treated tissue relative to the control (**Tables S2-S5**). While many proteins were localized to a single spot, some other proteins could be found in multiple spots, suggesting the presence of multiple isoforms. Where

these isoforms could not be distinguished from one another via the database search, the fold change (and associated spot number) is reported for the parent protein.

High O₃-induced changes the total and redox proteomes of soybean leaf

Comparison between the four tissue-O₃ treatment combinations revealed two distinct trends in the high O₃ leaf sample compared to the other samples - increased expression levels and oxidation of the largest number of proteins (**Fig. 3 and Fig. S2**). Although analysis of the 115 ppb O₃ root and 60 ppb O₃ leaf and root samples showed multiple proteins either increasing or decreasing in expression level, these changes were generally less than 2-fold different (**Fig. S2**). Moreover, both the numbers of protein changes and the fold changes in these samples were generally less than those observed in the high O₃ leaf sample. For example, in the 115 ppb O₃ root sample, the expression of 6 proteins increased and 10 proteins decreased, and in the 60 ppb O₃ leaf and root samples, fewer proteins increased in expression than decreased. Overall, the fold changes in total protein levels observed in the high O₃ root, elevated O₃ leaf and root samples were comparable to those described in previous studies of plant proteomes following acute O₃ exposure in growth chambers (22-28).

In stark contrast, the 115 ppb O₃ leaf sample contained 35 proteins with up to 5-fold increased expression compared to ambient samples, and only 2 proteins with ~1.5-fold decreased levels (**Fig. 3**). Even more striking was the shift in proteins that increased in oxidation in the high O₃ leaf sample compared to the other three samples. 22 proteins increased in total expression level and became 2- to 9-fold more oxidized. In addition, 22

other proteins became up to 5-fold more oxidized without significant changes in expression levels (**Fig. 3**). In comparison, the high O₃ root sample had 11 proteins that increased in oxidation and 8 that were more reduced (**Fig. S2A**). In the 60 ppb O₃ tissue samples, only a handful of proteins were either more oxidized or reduced than controls (**Figs. S2B & S2C**). The observed changes in the total and redox proteomes of leaf tissue exposed to high O₃ occurred across a range of metabolic pathways (**Fig. 4**), including redox systems, carbon metabolism, photosynthesis, signaling & homeostasis systems, amino acid metabolism, specialized metabolism of flavonoids and isoprenoids, protein degradation, and nucleic-acid systems, and are discussed in more detail later.

Analysis of enzymatic activities in leaf tissue exposed to high O₃-treatment

To better examine the activity changes in the 115 ppb O₃ leaf sample, targeted assays of the glycolytic/Calvin cycle enzymes phosphoglycerate kinase (PGK), fructose 1,6-bisphosphate aldolase (FBA), and glyceraldehyde-3-phosphate dehydrogenase (GAPDH), malate dehydrogenase (MDH) in the citric acid cycle, and glutamine synthetase (GS) in amino acid synthesis were performed. All of these enzymes showed increased protein expression and ~5-fold oxidation changes in the 115 ppb O₃ leaf sample. Except for PGK, each enzyme exhibited increased activity in the high O₃ leaf sample compared to the ambient control with the fold change in enzymatic activity correlated to increased expression level (**Table 1**).

In addition to these enzymes, the activity and total expression of ribulose 1,5-bisphosphate carboxylase oxygenase (RuBisCO) and phosphoenolpyruvate carboxylase

(PEPC) were examined because O₃ exposure can alter levels of these proteins in plants (2-3; 13-17; 19-21). For RuBisCO, both activity assays (**Table 1**) and Western blot analysis of the large subunit (**Fig. S3**) showed no significant difference between the high O₃ leaf sample and the ambient control. Likewise, expression of PEPC in the ambient and high O₃ leaf samples, as determined by Western blot, was not altered (not shown).

In the 115 ppb O₃ leaf sample, a glycosyl hydrolase/chitinase showed 4- and 9.4-fold increases in expression and oxidation state, respectively. Glycosyl hydrolases are involved in the degradation of various sugars, but are also mechanistically related to chitinases, which cleave glycosidic bonds and are typically involved in pathogen responses to insects or herbivory. Because many glycosyl hydrolases display varied activities, a fluorescence-based assay was used to evaluate exochitinase, endochitinase, and chitobiosidase activity in control and high O₃ leaf tissues, which were shown to increase 1.6-, 4.1-, and 11.1-fold, respectively (**Table 2**). It is unclear if these changes result from the identified protein or from aggregate changes in multiple glycosyl hydrolases.

DISCUSSION

Understanding the molecular mechanisms and metabolic consequences of how global climate changes, such as elevated tropospheric O₃ levels, impact crop plants is essential for efforts to maintain crop performance under increasing environmental stresses. Although earlier growth-chamber studies describe the effects of acute O₃ exposure on the proteomes of different crops, including soybean (22-28), these reports have neither

reported on the consequences of chronic O₃ exposure under field conditions nor examined the effect of O₃ on redox-sensitive proteins. Previous studies, in which soybean, rice, or wheat were exposed to constant 120 to 200 ppb O₃ for 3 to 5 days in growth chambers, typically identified 20 to 50 proteins that changed in expression level, either up or down (22-28). Analysis of the total and redox proteomes of leaf and root tissues from soybean plants grown in the field at SoyFACE under elevated (60 ppb) chronic daytime exposure to O₃ showed similar changes in both number of proteins and expression levels to earlier growth-chamber experiments (**Figs. 2 & S2**). In addition, leaf and root samples of soybean grown under elevated O₃ stress exhibited less than 2-fold differences in redox state for only a handful of proteins (**Fig. S2**). In contrast, the high O₃ soybean leaf sample analyzed here displayed striking increases in both expression levels and/or oxidation of multiple proteins across different metabolic pathways (**Figs. 3 & 4**). This suggests that there is a cross-over point between 60 and 115 ppb O₃ at which the expression levels and oxidation state of multiple proteins in leaf tissue dramatically shift, potentially as a metabolic adaptation to long-term field growth under chronic O₃ exposure.

Tropospheric O₃ negatively affects soybean growth and yield at concentrations greater than 40 ppb with decreased shoot and pod biomass, fewer pods produced, and premature leaf senescence (2-4; 12-18). O₃ enters leaves through the stomata and produces reactive oxygen species (ROS) that subsequently oxidize the plasma membrane and photosystem components leading to degradation of chlorophyll (2-3). Physiologically, long-term O₃ stress leads to reduced photosynthesis and mobilization of

reserve energy stores by converting leaf starch to sugars (28). Accordingly, sugar catabolism increases and previous studies identified several primary metabolic proteins, including MDH and phosphoglycerate mutase, as highly expressed under acute O₃ stress (23). The data presented here indicates that the changes in total and redox proteomes of soybean leaf tissue resulting from chronic exposure to high O₃ levels are more widespread across metabolism than previously reported and not necessarily the same as those resulting from acute O₃ exposure (**Fig. 4**).

Decreased photosynthetic efficiency, reduced RuBisCO activity, and elevated PEPC activity are classic markers for O₃ damage and senescence; however, these effects vary with the length, concentration, and type of exposure (2-3; 13-17; 19-21). Here soybean tissues were harvested at the R3 stage before significant changes in photosynthesis were observed (13; 16). The increased expression of chlorophyll a/b-binding protein, ferredoxin reductase, and a chlorophyllase-like protein observed in the 115 ppb O₃ leaf sample (**Fig. 4**) may help maintain photosynthesis at this growth stage before ozone-induced senescence occurs. Similarly, RuBisCO (large and small subunits), RuBisCO activase, a RuBisCO-associated protein, and RuBisCO-binding protein displayed elevated expression and/or oxidation in the 115 ppb O₃ leaf sample. Moreover, proteins related to iron homeostasis (ferredoxin reductase and ferritin) also change in soybean leaf under high O₃ stress. Because the spots containing RuBisCO included multiple proteins, activity assays and Western blot analysis were used to further examine possible changes in activity. Both methods showed no significant alteration in RuBisCO at this stage of soybean growth (**Table 1** and **Fig. S3**). Likewise, the effect of chronic 115 ppb O₃

exposure on PEPC levels in leaf tissue was analyzed by Western blot, which indicated no significant change in expression compared to ambient O₃ exposure at the time of harvest. These results suggest that major alterations in soybean photosynthesis are likely linked to senescence and occur later in the growing season for plants under chronic O₃ exposure.

Multiple proteins (i.e., PGK, GAPDH, FBA, ribose-5-phosphate isomerase, phosphoribulokinase, triosephosphate isomerase, MDH, and isocitrate dehydrogenase) in the reduction and regeneration phases of the Calvin cycle, glycolysis, and the TCA cycle increase in expression and/or oxidation state in the high O₃ leaf sample (**Fig 4**). In addition, the total activity levels of FBA, GAPDH, and MDH increased in the 115 ppb O₃ leaf sample (**Table 1**). This is consistent with earlier proteomic studies (22-28), but the analysis here indicates a wider range of protein changes and for the first time identifies redox-state alterations resulting from an environmental oxidative stress. All of these proteins are known to interact with thioredoxin, which is essential for maintaining the protein redox-state in plants (29). Moreover, phosphoribulokinase and GAPDH form a protein complex via thioredoxin-mediated redox changes in response to light intensity (31). O₃-related changes in cellular oxidation state may affect this interaction. Similarly, MDH is a critical regulatory point in the TCA cycle; however, the cytosolic form of the enzyme is redox-regulated and inactivated under oxidizing conditions (32). In the 115 ppb O₃ leaf sample, MDH had 2- and 5-fold higher expression and oxidation compared to controls with a 1.3-fold increase in total activity (**Table 1**). It is possible that these changes reflect the need to maintain MDH in the leaf to supply metabolites to the TCA cycle.

In addition to changes in core carbon metabolism, the starch and sugar mobilization pathways (phosphohexomutase, glucanase, and a glycosyl hydrolase/acid chitinase), the glycerate and glycolate pathways, and the biosynthesis of isoprenoids, carotenoids, and (iso)flavonoids display increased expression and/or oxidation state in the high O₃ leaf sample (**Fig. 4**). The changes in enzymes involved with the conversion of starch to sugar are consistent with a shift in energy demands of crops under O₃ stress (23). Of the 79 proteins identified in the 115 ppb O₃ leaf sample, a protein annotated as a glycosyl hydrolase/acid chitinase undergoes the greatest fold changes in both expression and oxidation state (**Fig. 3**). Although it is unclear if this protein functions in cell wall degradation, pathogen response, or sugar mobilization, the overall activity of glycosyl hydrolases in soybean leaf increase 1.6- to 11-fold in the high O₃ samples (**Table 1**). These increases may be connected to the mobilization of starch for energy production.

Proteins in three specialized metabolic pathways related to O₃ stress were also identified in the 115 ppb O₃ leaf sample. In the isoprenoid synthesis pathway, deoxyxylulose phosphate (DXP) oxidoreductase and isopentenyl diphosphate (IPP) isomerase are oxidized (**Fig. 4**); however, the effect of oxidation of these proteins remains to be determined. Interestingly, volatile isoprenoid emissions, including isoprene and monoterpenes, may act as an ozone protection mechanism in plants (33). Moreover, changes in carotenoid and (iso)flavonoid pathways (carotenoid-associated protein, chalcone isomerase, isoflavone reductase, and caffeoyl-CoA methyltransferase) suggest alterations in the synthesis of these compounds, which act as photoprotective compounds and anti-oxidants (34).

In leaves exposed to high O₃ levels, up-regulation and/or oxidation of proteins in amino acid biosynthesis and/or nitrogen homeostasis were also observed (**Fig. 4**). The cytosolic form of GS, a central player in nitrogen sensing, increased in expression and oxidation in the 115 ppb O₃ leaf sample, which also corresponded with increased total activity compared to controls (**Table 1**). Elevated expression of cytosolic GS is associated with leaf senescence and the recycling of ammonia during stress conditions (35). Aspartate-semialdehyde dehydrogenase is the primary control point for the biosynthesis of isoleucine, methionine, lysine, and threonine. Although redox-control has not been described for the plant enzyme, reversible oxidation of a catalytic cysteine in the bacterial homolog alters activity (36). Likewise, carbamoyl phosphate synthetase, which becomes more oxidized following high O₃ exposure in soybean, is also sensitive to changes in redox environment (37). Also related to nutrient metabolism, the observed expression and oxidation changes in 14-3-3 proteins may further modify the activities of enzymes across the carbon, nitrogen, and sulfur nutrient assimilation pathways and/or signal transduction systems linked to stress responses (38-39).

The proteomic analysis here supports studies demonstrating that redox-protection mechanisms play a critical role in plant responses to O₃ exposure, and for the first time directly demonstrates that O₃ exposure changes the oxidation states of multiple proteins in different metabolic pathways. High chronic O₃ exposure leads to an oxidative stress that activates redox protection mechanisms in plants and increases expression and/or oxidation of proteins in those systems, including ascorbate peroxidase, methionine sulfoxide reductase, and glutathione-S-transferases (GSTs) (**Figs. 3 and 4**). In plants, the

ascorbate-glutathione system is critical for maintaining redox homeostasis and for scavenging ROS produced by photosynthesis (40). As such, the increased expression and oxidation of ascorbate peroxidase, which is critical in this system, is directly linked to cellular responses to attenuate oxidative stress induced by high O₃ exposure in leaf tissue. This is also linked to increased mobilization of sugar stores, which can further enhance ascorbate synthesis (41). Thus, the up-regulation of glucose catabolism is linked to energy production and the generation of reducing equivalents for the detoxification of ROS. The nearly 3-fold increase in oxidation of methionine sulfoxide reductase, which targets oxidized methionine residues (42), suggests an important role for this protein in responding to O₃ stress. Moreover, the reaction mechanism for the fungal methionine sulfoxide reductase proceeds through the formation of a thioredoxin-mediated intramolecular disulfide bond (43); however, it is unclear if the activity of the plant enzyme is redox-responsive. Likewise, GST isoforms were detected as changing in expression and/or oxidation across several different tissue-type/ozone concentration combinations (**Tables S2-S5**). In plants, GSTs comprise a large family of enzymes that conjugate glutathione to either small molecules or proteins for xenobiotic detoxification and redox-modifications (44).

Although the role of O₃ as an oxidative environmental stress is well established (2-4, 7), the extent of redox-linked changes in crop plants in the field faced with chronic exposure to high O₃ concentrations has not been examined previously. Analysis of the soybean redox proteome in the 115 ppb O₃ leaf sample revealed 44 proteins with 2- to 9-fold higher oxidation than in control samples. This work is the first report that O₃-

exposure is directly linked to redox changes in plant proteins. Given that 2-DE methods were used, the observed changes in the redox proteome of soybean leaf are likely only a small fraction of the total number of proteins that change in oxidation state. Future targeted efforts using more sensitive isolation/detection strategies promise to reveal a greater extent of redox-linked changes resulting from O₃ stress. Because changes in redox-state of plant proteins can drastically alter activity in response to environmental and cellular stresses, further work is also required to examine how oxidative stresses modulate protein activity across plant metabolism. From a physiological perspective, a better understanding of how above and below ground metabolisms alter is also required. O₃ enters leaves via the stomata, but alterations in the expression and/or oxidation state of proteins in root tissues were observed, albeit not at the same intensity as in leaf tissue. Nevertheless, it is unclear if these result from systemic changes in metabolism or directly from O₃ exposure in roots. Ultimately, understanding how environmental ozone affects redox-sensitive pathways will aid in the development of crops better adapted to global climate change and provide information about how to target the engineering of ozone protection systems.

MATERIALS AND METHODS

Plant Growth and SoyFACE O₃ Treatment. Soybean (*G. max* (L.) Merr) were planted and exposed to ambient (40 ppb), elevated (60 ppb), and high (115 ppb) O₃, as described previously (16). All O₃ levels are seasonal 9-hour average concentrations. O₃ was not added at night or when leaves were wet.

Soluble Protein Extraction. Protein extraction was performed in triplicate using tissue from three different plants. All steps were carried out at 4 °C, unless otherwise indicated. Approximately 800 mg of root or 400 mg of leaf tissue were frozen in liquid nitrogen and ground to a fine powder using a mortar and pestle. The tissue was suspended in extraction buffer [100 mM Tris-HCl, pH 8.0; 100 mM NEM; 1% CHAPS; 1% plant protease inhibitors (Sigma)] to 200 mg mL⁻¹. For lysis, samples were sonicated (3 x 15 sec) with the tissue mixed and iced (30 sec) between sonications. Samples were then centrifuged (16,000 x g; 15 min) to precipitate insoluble debris. The resulting supernatant was mixed with 4 volumes of pre-chilled methanol and stored on ice for 30 min. Samples were again centrifuged (16,000 x g; 15 min) and the protein pellet harvested. The pellet was washed twice with methanol (2 x 30 min) and centrifuged (16,000 x g; 5 min) after each wash. During washing, care was taken to periodically disrupt the pellet to ensure complete removal of excess NEM. After the final spin, the pellet was air-dried and resuspended in 150 µL of reduction buffer (50 mM Tris-HCl, pH 8.0, 7 M urea, 2 M thiourea, 50 mM DTT) per 200 mg of starting tissue. The suspension

was incubated for 15 min at room temperature and then re-precipitated with 4 volumes of pre-chilled methanol. As described above, the protein was washed 3 times (3 x 30 min) with the same volume of methanol and centrifuged after each wash (1 x 15 min; 2 x 5 min). After the final spin, the pellet was air-dried and resuspended in 150 μ L of labeling buffer [40 mM HEPES, pH 7.5; 50 mM NaCl; 200 μ M IAF] per 200 mg of starting tissue. Due to the light sensitivity of IAF, this and all subsequent steps were carried out under dim light. The suspension was incubated for 10 minutes at room temperature, and then re-precipitated with 4 volumes of pre-chilled methanol. Again, the protein was washed 3 times (3 x 30 min) with the methanol and centrifuged after each wash (1 x 15 min; 2 x 5 min). After the final spin, the pellet was air-dried and resuspended in a small volume (\sim 50 μ L per 200 mg of starting tissue) of DeStreak buffer (GE Healthcare). Protein concentration was determined by CB-X assay (G-Biosciences).

Protein Separation by 2-DE. As above, all steps were carried out under dim light. To begin, 200 μ g of extracted protein was resuspended to a total volume of 180 μ L in DeStreak buffer and absorbed into a pH 4-7 gel strip. Isoelectric focusing in the first dimension was carried out at room temperature in a Proteon IEF cell using a four-step method: 1) 250 V, linear increase, 30 min; 2) 500 V, linear increase, 1 hr; 3) 8000 V, linear increase, 2.5 hr; 4) 8000 V, rapid increase to 35,000 until complete. Separation by molecular weight was in the second dimension at 150 V until the dye front reached the gel edge. Gels were then removed from their cassettes and imaged (λ_{ex} =488 nm and λ_{em} =520 nm) using a Typhoon 9410 (GE Healthcare) to detect IAF-labeled proteins.

After imaging, gels were bathed for 30 min in 100 mL of fixing solution [10% methanol; 7% glacial acetic acid] using an orbital shaker. The fixing solution was removed and the gels bathed overnight in 50 mL of Sypro Ruby (Biorad) protein stain. After staining, the gels were again bathed for 30 min in 100 mL of fixing solution. Following a wash step with MilliQ water to remove any excess stain, the gels were imaged again ($\lambda_{\text{ex}}=457$ nm and $\lambda_{\text{em}}=610$ nm) to detect total protein. For all gel replicates within a given set or sets to be compared, the same laser intensity (400 V for IAF images and 650-800 V for Sypro images) was used. After imaging was complete, gels were stored in MilliQ water at 4 °C until needed.

Gel Analysis, Spot Extraction and Mass Spectrometry. Replicate gels images were aligned using Progenesis SameSpots (Nonlinear Dynamics). Further alignment of replicate control and ozone-treatment gel images was carried out for each pairwise comparison. In order to quantify expression and thiol composition differences between the control and treated samples, spot volume was calculated and normalized for each spot in the aligned images. Those spots that different significantly in volume (ANOVA, $p<0.05$) between the averaged control and ozone treatment gels were then marked for identification.

Excision of significant spots from their respective gels was performed using a Gelpix System (Genetix) under high humidity (>85%) to prevent gel distortion or tearing. Gel plugs were dehydrated with 200 μL of acetonitrile (ACN) using a room-temperature table-top shaker (15 min; 900 rpm). ACN was then removed and the plugs washed five

times with 200 μL of 50mM NH_4HCO_3 ; 50% ACN (15 min; 900 rpm) with the liquid discarded after every wash to remove the Sypro Ruby stain. Following the last NH_4HCO_3 /ACN wash, the plugs further were washed with 100 μL of ACN (15 min; 900 rpm). After this, the liquid was discarded and the plugs air-dried. Once dry, the plugs were submerged in 20 μL of trypsin digestion buffer (50mM NH_4HCO_3 with 6 ng μL^{-1} trypsin) and rehydrated overnight at 37 °C. Next, 30 μL of 1% formic acid; 2% ACN was added to the digests, which were then shaken (30 min; 900 rpm). Following this wash, the supernatant was saved, the plug transferred to a new tube containing 24 μL of 60% ACN, and the tube shaken (30 min; 900 rpm). After this final wash, the supernatant was removed and added to that collected during the previous step, and the plugs discarded. The combined digest from each gel plug was lyophilized to dryness and then resuspended in 7 μL of 1% formic acid; 5% ACN.

Peptide separation and analysis were carried out as previously described (30). LC-MS/MS was conducted via an Eksigent nanoLC with a Dionex C18 PepMap100 column (75 μM id) coupled to a QSTARR XL MS/MS-TOF (Applied Biosystems). The peptide tandem mass spectra were processed using Analyst QS v1.1 (AB Sciex) and searched against the NCBI nr database (July 2010, 11368323 sequences) using an in-house version of MASCOT v2.20 (Matrix Science) with the following parameters: tryptic peptides with ≤ 1 missed cleavage site; precursor and MS/MS fragment ion mass tolerances of 0.8 and 0.8 Da, respectively; variable carbamidomethylation and fluoresceination of cysteine; and variable oxidation of methionine. The data was filtered using Scaffold 3 (Proteome

Software). Positive identification criteria was ≥ 2 peptide sequences, protein probability of 99.9%, and peptide probability of 80%.

Enzyme Assays. Standard spectrophotometric assays were used to determine activity of PGK (45), MDH (46), GS (47), GAPDH (48), FBA (49), and RuBisCO (50). To measure chitinase activity, a fluorescence-based kit (Sigma, CS1030) was used. Tissue extracts were prepared as above, and equal amounts of control and treated tissue extracts were added to tubes containing assay-appropriate buffer plus 1% plant protease inhibitors (Sigma).

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Figure 1. Representative 2-DE Gels. The gels shown contain proteins isolated from roots of soybean exposed to 60 ppb O₃. **(A)** 2-DE gel visualized for IAF-labeling of the redox proteome. Lines and corresponding numbers indicate spots which significantly differed ($p < 0.05$) in degree of oxidation as compared to control. **(B)** The same 2-DE gel from panel A, but with total proteome visualized with SYPRO ruby. Lines and corresponding numbers indicate spots which significantly differed ($p < 0.05$) in total expression as compared to control.

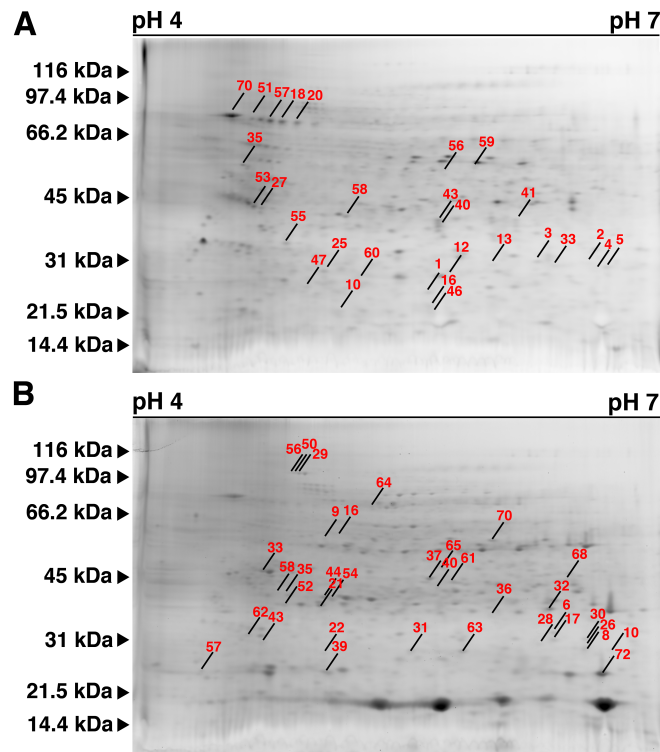


Figure 2. Venn diagram of proteins that differ in leaf and root tissues under 60 ppb and 115 ppb O₃ treatments compared to ambient conditions. **(A)** Distribution of proteins across all four combinations of tissue and O₃ concentration. Numbers in overlapping regions of the lobes indicate proteins found in more than one set of conditions. **(B-E)** Detailed breakdown of numbers of differentially oxidized (IAF) and/or expressed (SYPRO ruby) proteins between treated samples and controls. In each panel, the miniature Venn diagram in the top left corner indicates which lobe from panel A is analyzed.

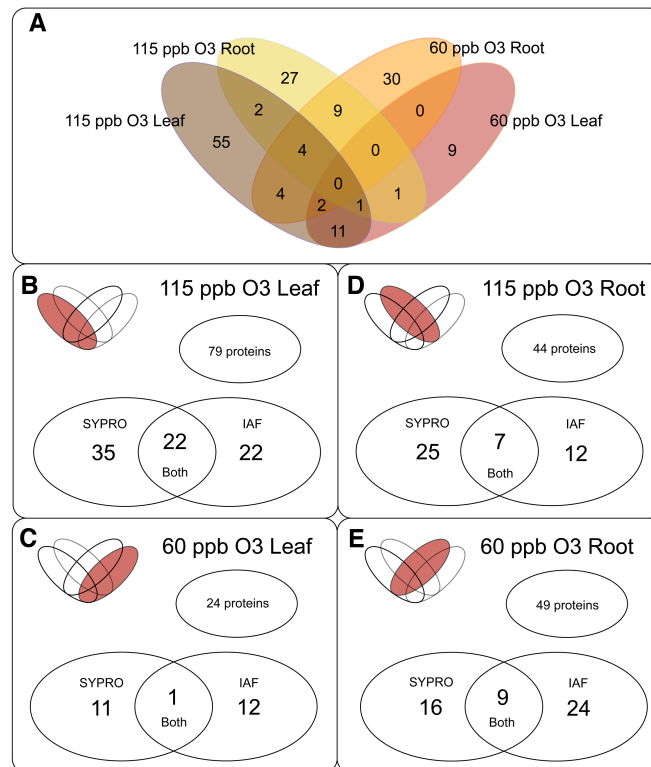


Figure 3. Summary of fold changes in total and redox proteomes of leaf tissue exposed to 115 ppb O₃. Fold changes, relative to ambient O₃ control, in oxidation state (IAF - fold change) and expression level (Sypro - fold change) for identified proteins identified are plotted. Names of representative proteins are shown with highly oxidized (orange box) and oxidized/expressed proteins (red) indicated. Additional information about the identified proteins is provided in Tables S2.

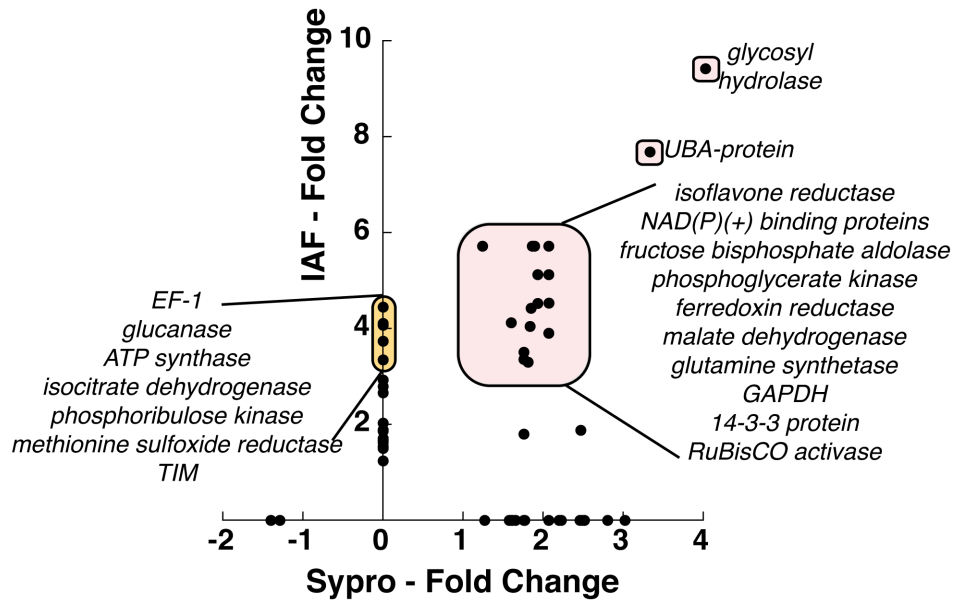


Figure 4. Metabolic Overview of Total and Redox Proteome Changes in Soybean Leaf Tissue Exposed to 115 ppb O₃. A schematic view of the different metabolic pathways identified is shown. Proteins that change in oxidation state (orange), expression level (red), or both oxidation state and expression level (red with black outline) are shown. Detailed information about the identified proteins is provided in Table S2.

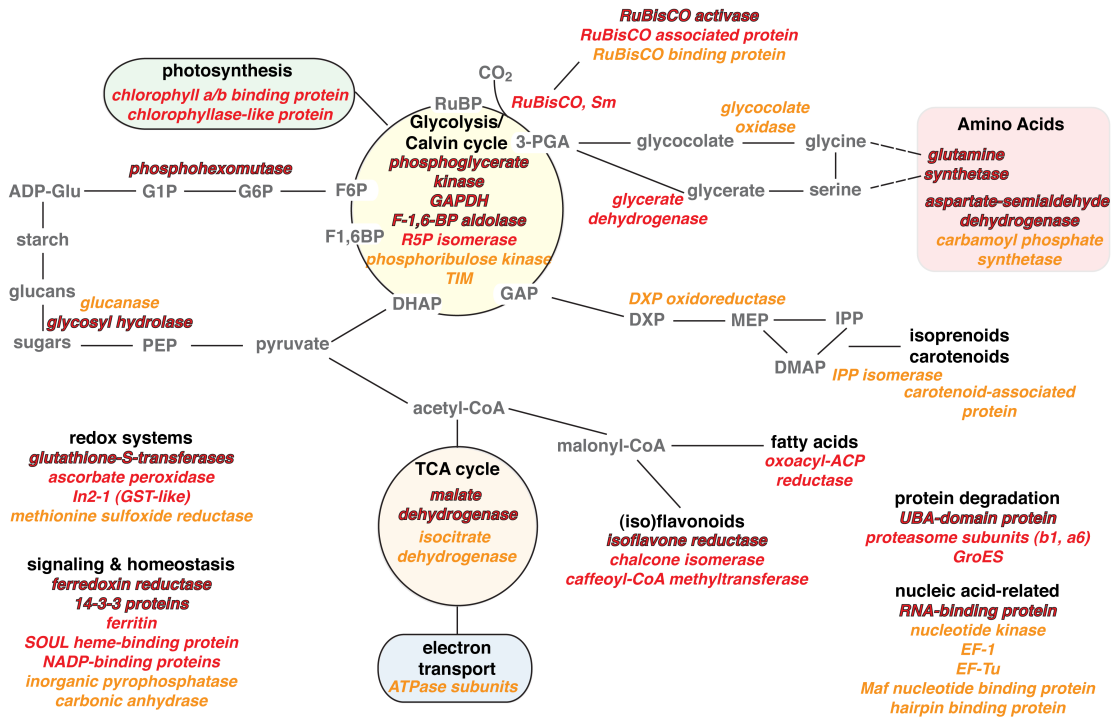


Table 1. Comparison of enzyme activities in leaf tissues exposed to ambient (40 ppb) and high (115 ppb) O₃. All assays were performed as described in the methods section. Values are averages ± standard deviations for n = 4-8. ND - no detected changes. Abbreviations are as used in the text.

	ambient ozone activity ($\mu\text{mol min}^{-1} \text{g FW}^{-1}$)	high ozone activity ($\mu\text{mol min}^{-1} \text{g FW}^{-1}$)	activity fold change	total protein (oxidation) fold change
PGK	715 ± 82	629 ± 68	0.9	1.8 (5.7)
FBA	31.6 ± 6.5	72.1 ± 9.9	2.3	1.8 (4.8)
GAPDH	183 ± 75	693 ± 301	3.8	1.7 (4.4)
MDH	72.0 ± 3.3	93.9 ± 10.4	1.3	2.0 (5.4)
GS	106 ± 9	246 ± 25	2.3	1.9 (4.5)
RuBisCO	11.1 ± 2.6	10.0 ± 4.1	0.9	ND (ND)
exochitinase	10.5 ± 0.8	17.2 ± 0.3	1.6	4.0 (9.4)
endochitinase	0.34 ± 0.02	1.40 ± 0.08	4.1	4.0 (9.4)
chitobiosidase	0.012 ± 0.046	0.133 ± 0.023	11.1	4.0 (9.4)

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Figure S1. Redox Proteome Labeling Approach. Proteins with free thiols (-SH), disulfide bonds (-S-S-), or modified cysteines (-S-mod) are incubated with N-ethylmaleimide (NEM) to block free sulfhydryl groups. Oxidized thiols are reduced with dithiothreitol (DTT). The resulting free thiols are labeled with 5-iodoacetamidofluorescein (IAF), and then the proteins are separated by 2-DE and identified by LC-MS/MS.

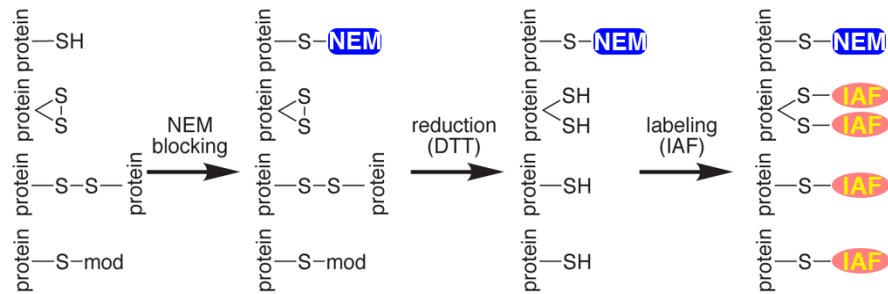


Figure S2. Additional Total and Redox Proteome Changes. Panels A-C show the fold changes in oxidation (IAF - fold change) and expression level (Sypro - fold change) relative to ambient controls for proteins identified by mass spectrometry in root and leaf tissues exposed to high and elevated O₃. Detailed information about the identified proteins is provided in Tables S2-S5.

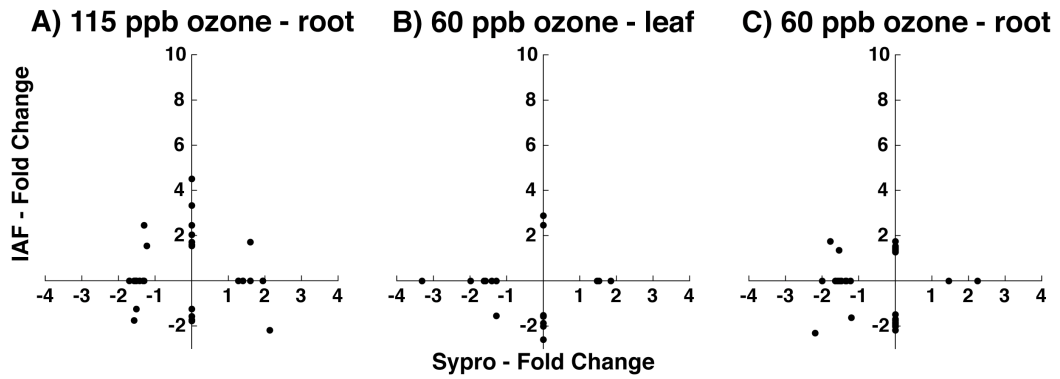


Figure S3. Immunoblot analysis of RuBisCO large subunit expression. Protein extracts from leaf tissue exposed to ambient (40 ppb) and high (115 ppb) O₃ were probed using anti-RuBisCO large subunit antibody. Lanes 1 and 3 contain 10 μg of total protein extract and lanes 2 and 4 contain 5 μg of total protein extract.

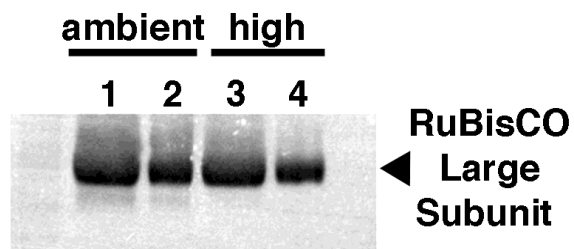


Table S1. Total number of spots detected and identified as either differentially expressed or oxidized across all experimental conditions. Differentially expressed/oxidized spots are further broken down into those with either single or multiple proteins.

O ₃ Level, Tissue, and Signal	Spots Detected	Spots Differentially Expressed/Oxidized	Spots with One Protein	Spots with Multiple Proteins	Total Spots Identified
115 ppb, leaf Sypro	154	29	9	17	26
115 ppb, leaf IAF	171	47	9	18	27
115 ppb, root Sypro	208	36	4	10	14
115 ppb, root IAF	196	29	5	6	11
60 ppb, leaf Sypro	196	44	7	4	11
60 ppb, leaf IAF	195	43	7	3	10
60 ppb, root Sypro	158	21	7	10	17
60 ppb, root IAF	177	28	9	15	24
Total	1455	277	57	83	140

Tables S2-S5. Data Summary Tables. These tables list peptides that are differentially expressed and/or oxidized from 115 ppb O₃ leaf tissue (Table S2), 115 ppb O₃ root tissue (Table S3), 60 ppb O₃ leaf tissue (Table S4), and 60 ppb O₃ root tissue (Table S5). The accession number of the protein, molecular weight, fold changes, identified peptides, and mascot ion scores are indicated. For each protein, the fold change and spot number are listed as [(spot number) fold change]. The magnitude of the fold change associated with each protein is indicated by color in the box, as follows: 1.2- to 3-fold, pale green; 3- to 5-fold, medium green; >5-fold, dark green; -1.2- to -3-fold, pale blue; and -3- to -5-fold medium blue.

Supporting Information Table 1. List of Peptides Identified from 115 ppb Leaf Proteins that are Differentially Expressed/Oxidized									
Protein	Accession Number(s)	Mol Weight	Fold Change IAF (spot number)	Fold Change SYPRO (spot number)	Classification	IAF Peptides (spot number)	Best Mascot Ion Score	SYPRO Peptides (spot number)	Best Mascot Ion Score
TIM phosphate binding protein (Glycine max)	gi 255640328	14 kDa	[(35)3.347], [(55)2.375], [(59)2.260]		glycolysis	[(35)] AAQDALLFR SNSLAQLGK YTAEGESEATR [(55)] AAQDALLFR ATPLQVADYTLK SNSLAQLGK YTAEGESEATR [(59)] AAQDALLFR SNSLAQLGK	56.3 60.8 70.3 67.8 64.4 66.1 87.9 71.2 59.3		
Nucleoside/nucleotide kinase (Glycine max)	gi 255639590	50 kDa	[(35)3.347]		nucleotide metabolism	[(35)] EANPGNALLELR VDPQLETWVK	55.6 59.1		
Hossmann-fold NAD(P)(+)-binding protein: 3-oxoacyl-(acyl-carrier-protein) reductase (Populus trichocarpa)	gi 224100059 gi 255547878 gi 255638092	28 kDa		[(13)2.805]	fatty acid metabolism			[(13)] ILETPLGR VESPVVVVTGASR	57.7 60.2
Hossmann-fold NAD(P)(+)-binding protein (Glycine max)	gi 255647108	42 kDa	[(21)4.533]	[(24)2.072]	unknown	[(21)] EADPSTDDIILGK FIGVFLSR	76.1 54.8	[(24)] EADPSTDDIILGK FIGVFLSR	76.1 54.8
Hossmann-fold NAD(P)(+)-binding protein (Glycine max)	gi 255635535	36 kDa	[(10)5.718], [(21)4.533]	[(40)1.655], [(25)2.070], [(24)2.072]	unknown	[(10)] DVSFLTNLPGASEK LLEDGYAVNTTIR LVDAGFEFK [(21)] DVSFLTNLPGASEK LLEDGYAVNTTIR	54.8 55.7 57.7 58.5 90	[(40)] DVSFLTNLPGASEK LLEDGYAVNTTIR [(25)] DVSFLTNLPGASEK LLEDGYAVNTTIR LVDAGFEFK [(24)] DVSFLTNLPGASEK LLEDGYAVNTTIR	59 66.6 54.8 55.7 57.7 58.5 90
Hossmann-fold NAD(P)(+)-binding protein (Glycine max)	gi 15778154 gi 26454611	29 kDa	[(89)1.667]	[(13)2.805]	unknown			[(13)] ALFSCQITTR QNIGAADDVIVGDIR	52.2 93.4
Hossmann-fold NAD(P)(+)-binding protein; possible isoflavone reductase (Glycine max)	gi 255648230 gi 2687724	36 kDa	[(10)5.718]	[(40)1.655], [(25)2.070]	secondary metabolism	[(10)] FFPSEFLDLDVDR NLAQIDITVPPR VFIQGDGNVK	51.7 60 55.4	[(40)] ILVLGPTGAGR VFIQGDGNVK [(25)] FFPSEFLDLDVDR NLAQIDITVPPR VFIQGDGNVK	66.9 57.1 51.7 60 55.4
14-3-3 protein [Nicotiana tabacum]	gi 15778154 gi 26454611	29 kDa	[(89)1.667]		signal transduction	[(89)] ENFVYVAK ERENFYVAK	56.6 55.8		

glutathione-s-transferase [Glycine max]	gi 2052029 gi 220683633 gi 255631159 gi 38679415	25 kDa	[(96)1.545]	redox homeostasis	FWADFDVNIK LLEEQLGDK	54.2 75.8		
glutathione-s-transferase [Glycine max]	gi 255637207	26 kDa	[(10)3.022]	redox homeostasis	GVEY ^[(10)] YKEENLR KVYDYWAVTK VVDYWAVTK		84 49.9 58.1	
proteasome beta type-1 subunit (Vitis vinifera)	gi 225453909 gi 255541320 gi 255627685 gi 255640620	25 kDa	[(20)2.226]	protein degradation	MSTGYN ^[(20)] ILTR TVFASATER		69.7 50.1	
ATP synthase beta subunit [Dioscorea communis]	gi 17224743 gi 21684883 gi 37720945 gi 42559020 gi 730692	52 kDa	[(26)4.107]	ATP-coupled proton transport	ESGVINEK ^[(26)] NAESK LAIFETGIK	68.1 70.4		
RuBisCO-associated protein (Glycine max)	gi 108712217 gi 125546535 gi 1134101 gi 225436538 gi 255587664 gi 297734943 gi 31193919 gi 5650721	31 kDa	[(16)2.460]	Calvin cycle	ILAVANE ^[(16)] YK VISDATASFPK VLVSI ^[(16)] GNIK		63.5 47.4 48.4	
RuBisCO subunit binding-protein alpha subunit, putative chloroplast precursor (Oryza sativa japonica)	gi 108712217 gi 125546535 gi 1134101 gi 225436538 gi 255587664 gi 297734943 gi 31193919 gi 5650721	50 kDa	[(87)1.711]	Calvin cycle	GILNVA ^[(87)] AIK GYSPQFV ^[(87)] TNPEK VGAATELEDR	49.4 50.4 94.5		
beta-form rubisco activase (Glycine max)	gi 255635315	49 kDa	[(46)1.600]; [(5)1.549]	Calvin cycle	GLAYDIS ^[(46)] DDQQDITR LLLYGNMLV ^[(46)] QEQENVK LVDTFPGOS ^[(46)] IDFFGALR MCALFIND ^[(46)] LDAGAGR MDGFYI ^[(46)] AFAMDK MGINPIM ^[(46)] MSAGELESGNAGEPAK VPIV ^[(46)] TGNDFSTLYAPLIR VFLILGI ^[(46)] WGGK VYDDE ^[(46)] VR WISGV ^[(46)] GVDSV ^[(46)] GK YLNEA ^[(46)] ALGNANEDA ^[(46)] IOR	93.4 101 51.7 55.7 53.8 114 62.6 85.9 60.5 57.4 66.6 93.1		

beta-form rubisco activase (Glycine max)	gi 255635315	49 kDa				Calvin cycle			<p>[[46]1.600]; [[51]1.549]</p>	<p>GLAYISDDQDITR LLYGNMLVQEQENWK LVDTFPGQSIDFFGALR MCALFINDLADAGGR MDGFYIAPAFMDK MGINPIMMSAGELESNGAGEPAK YPIIVTGNDFSTLYAPLIR VYDDEVIR WISGIVGDSVQK YLNEAALGNANEDAIOR</p>	<p>96.4 103 91.1 78.9 76.3 97.1 96.2 56.3 66.2 96.2</p>
Elongation factor 1 (Glycine max)	gi 255627339	25 kDa	[[22]4.449]			translation	<p>[[22]] KLEYLLPR LDEYLLPR SVQMEGLLWGASK</p>	<p>45.6 56.8 72.5</p>			
Elongation factor Tu, chloroplastic (Glycine max)	gi 2494261	52 kDa	[[26]4.107]			translation	<p>[[26]] ILDEALAGDNVGLLLR KYDEIDAAPEER NTTYVTGVEMFOK VGETVDLVGLR</p>	<p>59.9 78.6 51 76.1</p>			
phosphoglycerate kinase, putative (Ricinus communis)	gi 255544584	50 kDa				Calvin cycle/glycolysis			<p>[[35]1.771]</p>	<p>[[35]] ELDYLVGAVSSPK FVGVTEAIK GVTTIIGGGDSVAAVEK IGVIESLLEK LDLATSLLAK LSELLGIQVVK RPFPAIVGGSK</p>	<p>65 64.1 85.1 71.1 62.1 61.3 50.9</p>
phosphoglycerate kinase, putative (Populus trichocarpa)	gi 224109060	50 kDa				Calvin cycle/glycolysis			<p>[[35]1.771]</p>	<p>[[35]] ADLNVPLDDNQITDDTR FVGVTEAIK GVTTIIGGGDSVAAVEK IGVIESLLEK LSELLGIQVVK LVASLPDGGVLLLENVR RPFPAIVGGSK</p>	<p>75.3 64.1 85.1 71.1 61.3 70.1 50.9</p>
eukaryotic ferritin (Glycine max)	gi 255637227	29 kDa				redox homeostasis			<p>[[19]2.226]</p>	<p>[[19]] GDALYAMELALSLEK IAEYVTLR KIAEYVTLR NNDPQLADFIIESEFLYEQVK</p>	<p>61.1 57.2 67 67.4</p>
fructose-bisphosphate aldolase, putative (Ricinus communis)	gi 255575381 gi 255647273	39 kDa				Calvin cycle/glycolysis			<p>[[46]1.600]</p>	<p>[[46]] ANSEATLGTYSK AQEALLVR GILAADESTGTIGK</p>	<p>57.2 58.3 82.4</p>
fructose-bisphosphate aldolase (Glycine max)	gi 40457267	38 kDa	[[10]5.718]			Calvin cycle/glycolysis	<p>[[10]] GILAADESTGTIGK YVGGSGGLTSESLEYK</p>	<p>62.1 84.4</p>		<p>[[40]] GILAADESTGTIGK LASINVENIENR YVGGSGGLTSESLEYK</p>	<p>82.6 55.7 107</p>
inorganic pyrophosphatase (Glycine max)	gi 255640293	32 kDa	[[74]1.885]			ATP-coupled proton transport	<p>[[74]] APFVNDVDVVEK IVASLDDPK MEVATDESFTPIK</p>	<p>64.5 62.3 52.7</p>		<p>[[25]] GILAADESTGTIGK YVGGSGGLTSESLEYK</p>	<p>62.1 84.4</p>
Glycerate/hyde 3-phosphate dehydrogenase (Glycine max)	gi 255636463	48 kDa	[[25]4.123]			Calvin cycle/glycolysis	<p>[[25]] ILDNETTYDQK YDSMLGTQK</p>	<p>66.7 59.2</p>		<p>[[46]] GLTAEDVNAFFR ILDNETTYDQK YDSMLGTQK</p>	<p>60.5 81.8 55.4</p>

RRM (RNA recognition motif), also known as RBD (RNA binding domain) or RNP (ribonucleoprotein domain) protein (Glycine max)	gi 255645259	28 kDa	[(93)1.627],[66)2.105]		unknown	GNVSDGYGGGNYSR YGEWDAR	75.8 55.4	
RRM (RNA recognition motif), also known as RBD (RNA binding domain) or RNP (ribonucleoprotein domain) protein (Glycine max)	gi 255639723	30 kDa		[(10)3.022]	unknown		55.8 54.5	GFGEVTEGSPDEVK LFVGNLPEVSARSAR CFNGVELDGR SAIGSLDGVDLNGR
RRM (RNA recognition motif), also known as RBD (RNA binding domain) or RNP (ribonucleoprotein domain) protein (Glycine max)	gi 255636284	31 kDa	[(50)2.497],[33)3.357]		unknown	AEMFSGYELNGR LEQIFSEHGK VYVGNLPWEVDDAR	63.8 53 61.8	
phosphatase/phosphohexomutase (Populus trichocarpa x Populus deltoides)	gi 118488927 gi 224136009 gi 255540407 gi 255640791	35 kDa	[(74)1.885],[87)1.711]	[(36)1.760]	carbohydrate metabolism	AVEMFSGYELNGR VYVGNLPWEVDDAR IFAGDVVPR TELFMALEK	62.3 77.2 65.3 53.6	IFAGDVVPR TELFMALEK
plant basic secretory protein (Glycine max)	gi 255633510	27 kDa	[(68)2.030]		unknown	INLFVYEDR WDQGYDVTAR YVNDYGGDVK	59.5 56.2 67.1	
Chromoplast-specific carotenoid-associated protein, precursor (Cucumis sativus)	gi 62899808	35 kDa	[(92)1.627]		secondary metabolism - carotenoid	ALVDSFYGTDR GDGGSVFLK	61.3 51.8	
NADPH-specific isocitrate dehydrogenase (Glycine max)	gi 169989 gi 1706401 gi 255635311 gi 44921641 gi 479386	48 kDa	[(26)4.107],[28)4.034]		TCA cycle	LIDDMVAYALK LLDFTEK SFAEASMTALEK	61.1 53.2 54.2	
1-deoxy-D-xylulose 5-phosphate reductoisomerase (Pueraria montana var. lobata)	gi 35187000	51 kDa	[(26)4.107]		secondary metabolism - isoprenoid	LIDDMVAYALK SFAEASMTALEK AGGTMIGVLSAANEK AVEMFVEEK	53.4 73 82.4 50	

phosphoribulokinase .putative (Glycine max)	gi 255646270	45 kDa	[(35)3,347]([25)4,123]		Calvin cycle	ANDFLMYEQVK DLYEQLIASK FYGEVTCQMLK GVTALDPR KPDFEAYIDPOK LTSVFGAAEPPK [25] ANDFLMYEQVK FYGEVTCQMLK GVTALDPR KPDFEAYIDPOK LTSVFGAAEPPK [22] GSGIELIMDVAK IYYPDEEALQALR [35] EDGGYEVK TLPGPVSDPSELPK [(10)5,718] EDGGYEVK GDMWAGAGHTNYSTK TLPGPVSDPSELPK LDLTAELSEEK MELVDAAPFLK VLVVAHPANTNALIK [21] EDGGYEVK GDMWAGAGHTNYSTK TLPGPVSDPSELPK [45] IVTEILPAK SGIYYNDTQAR [103] GVAVSSVEEAR TDQVEDIAGK [98] AIALTVDTPR GVLTAEETR VPVFLDGGVR	76.2 49.5 71.8 61.5 55.7 69.1 71.8 73.8 61.8 58 75.2 68.9 71.8 49.3 57.7 65.9 59.9 73.8 52.3 63.2 58.2 80.2 59.6 59.9 58.7 64.6 60.1 57.9 54.2 59.7	60.8 53.1 57.1 57.7 65.9 59.9 73.8 52.3 63.2 58.2 80.2 59.6
beta-1,3-glucanase 3 (Glycine max)	gi 2921317	26 kDa	[[22)4,449]		carbohydrate metabolism		68.9 71.8	
cytosolic glutamine synthetase GSbet1 (Glycine max)	gi 10946357 gi 121336	39 kDa	[(35)3,347]([10)5,718]([21)4,513]	[(40)1,655]([25)2,070]([24)2,072]	amino acid metabolism	EDGGYEVK TLPGPVSDPSELPK [(10)5,718] EDGGYEVK GDMWAGAGHTNYSTK TLPGPVSDPSELPK LDLTAELSEEK MELVDAAPFLK VLVVAHPANTNALIK [21] EDGGYEVK GDMWAGAGHTNYSTK TLPGPVSDPSELPK [45] IVTEILPAK SGIYYNDTQAR [103] GVAVSSVEEAR TDQVEDIAGK [98] AIALTVDTPR GVLTAEETR VPVFLDGGVR	49.3 57.7 65.9 59.9 73.8 52.3 63.2 58.2 80.2 59.6 59.9 58.7 64.6 60.1 57.9 54.2 59.7	60.8 53.1 57.1 57.7 65.9 59.9 73.8 52.3 63.2 58.2 80.2 59.6
peptide methionine sulfoxide reductase (Glycine max)	gi 255631520	22 kDa	[(45)2,791]		redox homeostasis	IVTEILPAK SGIYYNDTQAR [103] GVAVSSVEEAR TDQVEDIAGK	59.9 58.7	
Carbamoyl-phosphate synthase L chain, ATP binding domain (Glycine max)	gi 25563622	19 kDa	[(103)1,240]		amino acid metabolism	GVAVSSVEEAR TDQVEDIAGK	64.6 60.1	
glycolate oxidase (Vitis vinifera)	gi 147789493 gi 1167961875 gi 189418957 gi 225462096 gi 255557255	41 kDa	[(98)1,502]		carbon metabolism	AIALTVDTPR GVLTAEETR VPVFLDGGVR	57.9 54.2 59.7	
Nucleotide binding protein Mtr (Glycine max)	gi 255642112	22 kDa	[(45)2,791]		nucleotide metabolism	ILGSSVAR LPVDDYLIK [69] ESELIDENALGVR YELLQOR	58.7 52.7	
isopentenyl diphosphate (IPP) isomerase (Glycine max)	gi 255635611 gi 255647098 gi 262036858 gi 6856554	27 kDa	[(89)1,667]		secondary metabolism - isoprenoid	ESELIDENALGVR YELLQOR	55.3 54	

S-adenosylmethionine-dependent methyltransferase (Glycine max)	gij255638590	27 kDa	[(75)1.880]	[(15)2.471]	unknown	[(75)] IDFIESPALPDK ITADINR KTYEVLPLVIK NPVILQSEDLTK TYEVLPLVIK YILETAVYPR	81.7 51 52.2 66.2 49.6 67.8	[(15)] IDFIESPALPDK ITADINR KTYEVLPLVIK NPVILQSEDLTK TYEVLPLVIK YILETAVYPR	81.7 51 52.2 66.2 49.6 67.8
lin-2-1 protein (Glycine max)	gij11385579 gij255637842	27 kDa		[(20)2.226]	redox homeostasis			[(20)] LATWFEELNK VDPQEIVDLFK YVDANFEGTLPFFSDFAK	59.3 60.3 62.4
SOUL_heme-binding protein (Glycine max)	gij255627213 gij255640875	26 kDa		[(10)3.022]	metal trafficking			[(10)] DSSVGEAAALK LFDYIOGK	59.4 47.9
putative chalcone isomerase 4 (Glycine max)	gij51039630	23 kDa		[(42)1.615]	secondary metabolism - flavonoid			[(42)1.615] GAQYGVQIETAVR YVIENANVVEAIK	60.3 81
chalcone isomerase A [Glycine max]	gij114199183 gij122725493 gij255625955 gij5921724 gij75305825	23 kDa		[(19)2.226]	secondary metabolism - flavonoid			[(19)] FTGIGVYLEDK SVGTYGDAAEAIEK	64.7 74.1
UBA-domain containing protein, Nascent polypeptide-associated complex NAC (Medicago truncatula)	gij217075454 gij255628877 gij67241023	22 kDa	[(5)6]8.14977 223	[(8)1]3.66/3 007	unknown	[(5)6] IEDLSSCLQTQAAEQFK NILFVSKPQVFK SPTSDDTYIIFGEAK	60.7 47.8 84.9	[(8)1] IEDLSSCLQTQAAEQFK NILFVSKPQVFK SPTSDDTYIIFGEAK	60.7 47.8 84.9
GTP-binding protein (Cicer arretinum)	gij10334503 (+61)	25 kDa		[(59)-1.401]	unknown			[(59)] LVIVDGGTGK NLQYYEISAK	62.8 48.9
chlorophyll a/b-binding protein (Glycine max)	gij1053216 gij255646685	28 kDa		[(42)1.615]	photosynthesis			[(42)] FGEAWFK QASSGSPWYGPDR	59.6 49.3
glycerate dehydrogenase (Cucumis sativus)	gij118564 gij118723307 gij1304042 gij1304044 gij13873334 gij147805559 gij1167963236 gij225428049 gij255581842 gij255639618 gij29293057 gij29293059 gij90761112	42 kDa		[(35)1.771]	carbon metabolism			[(35)] EGMATLAALNVLGK GQTVGVGAGR IVEADEFMR	53.8 57 49.5
putative aldol/keto reductase (Glycine max)	gij224555758 gij255637828	36 kDa		[(40)1.655]	unknown			[(40)] NILDONIGALAVK YIGLSEASPTIR	59.9 59.5
caffeoyl-CoA 3-O-methyltransferase (Amorpha fruticosa)	gij146782454 gij255656252	25 kDa		[(13)2.805]	secondary metabolism - flavonoid			[(13)] ENYELGLPLVIK ILAINDNR	50.4 59.5

aspartate-semialdehyde dehydrogenase (Glycine max)	gi 255669397	38 kDa		[(25)2,070]	amino acid biosynthesis			GTVVVDNSSAFR NAPGVVVDDHR	51.8 63.1
unknown (Glycine max)	gi 255644538	31 kDa		[(20)2,226]	unknown			DVADSVLADR YSSAAPLSPDAR	49.6 54.1
phosphatidylinositol transfer protein (PGP-TP) (Glycine max)	gi 255627393	17 kDa		[(65)-1,289]	lipid related			GQPATFSAATTGK GVEISPDPIAR	82.5 59.2
Glycosyl hydrolase family 18, acidic chitinase (Glycine max)	gi 255642487 gi 4835584	32 kDa	[(95)7,233;8,149];[(16)12,600];[(3)8,4,6,65];[(7)233];[(4,1)8,6,13];[(2,6,5)1]	[(5)3];[(4,445);4,600];[(3)8,4,6,65];[(8,1)13,660];[(3,00,7)	carbohydrate metabolism	ALNSFSSQR* YGGVMLWNR	75 61.5	ALNSFSSQR* YGGVMLWNR	64.9 52.8
ribulose 1,5-bisphosphate carboxylase small subunit precursor (Glycine soja)	gi 1079736 gi 10946975 gi 10946977 gi 10946979 gi 132096 gi 132113 gi 255625881 gi 255630492 gi 3914590	20 kDa		[(36)1,760]	Calvin cycle	ALNSFSSQR* YGGVMLWNR	64.8 51.2	ALNSFSSQR* YGGVMLWNR	64.8 51.2
Stress-induced protein SAM22 (Glycine max)	gi 134194 gi 2218276 gi 229597555	17 kDa		[(6)3,781];[(6,5)-1,289],				AIEAYLLAHPDYN ALVTDADNVIPK GDAEPNQDELK GVFTFEDEINSPVAPATLYK ITFLEDGETK SVENVEGGGPGTIK YETKGDAPNDELK	76.2 71.2 56.5 85.6 51.3 89.6 59.1
ribulose-1,5-bisphosphate carboxylase/oxygenase large subunit (Crotalaria zeyheri)	gi 1045644 (+1547)	50 kDa	[(74)1,655];[(26)4,107];[(10,5,7)6]	[(25)2,070]	Calvin cycle	DDENVNSQPFMR DITDLAAFR EITLGFVDLLR	49.8 48.3 50.2	EITLGFVDLLR LTYTTPDYETK	52.8 55.2

rubisco activase (Glycine max)	gi 290766485	49 kDa	[[25]4-123]([[103]1-240]([[2-1]4-533]	[[24]2-072]([[5]1]-549]	Calvin cycle	GLAYDISDDQQDITR LLTYGNMLVQEQENVK YLNEAALGNANEDAINR	80.8 85.1	GLAYDISDDQQDITR LLTYGNMLVQEQENVK YLNEAALGNANEDAINR	84.4 77.6 117
peroxiredoxin 5-like (Glycine max)	gi 255640689	25 kDa		[[65]-1-289]	redox homeostasis	GLAYDISDDQQDITR LLTYGNMLVQEQENVK WISGVGVDDGIGK YLNEAALGNANEDAINR	88.7 66.1 57.6 106	GLAYDISDDQQDITR LLTYGNMLVQEQENVK WISGVGVDDGIGK YLNEAALGNANEDAINR	96.4 91.1 102 78.9 97.1 66.3 59.1 96.2 56.3 66.9 124
carbonic anhydrase (Glycine max)	gi 255630357 gi 270342124	24 kDa	[[75]1-880]	[[15]2-471]([[55]-1-475]	carbon metabolism	EYEQAIIELOK NVANIVPPYDQSK	52.4 56.3	EYEQAIIELOK NVANIVPPYDQSK	52.4 56.3
Phosphoglycerate kinase, chloroplastic (Nicotiana tabacum)	gi 2499497	50 kDa	[[10]5-718]	[[25]2-070]([[35]1]-771]	Calvin cycle/glycolysis	ELDYLVGAVSNPK GVTTIIGGGDSVAAVEK IGVIESLLEK YSLAPLVPR	64.1 54.5 64.1 53.6	ELDYLVGAVSNPK GVTTIIGGGDSVAAVEK IGVIESLLEK YSLAPLVPR	64.1 54.5 64.1 53.6
CYPOR-like ferridoxin reductase (Populus trichocarpa)	gi 224074257	40 kDa	[[10]5-718]	[[25]2-070]	redox homeostasis	EGOSGIVIPDGIDK GIDDIMVSLAAK LDFAVSR LYSIASSAIGDFGDSK	50.1 64.7 57.6 79.6	EGOSGIVIPDGIDK GIDDIMVSLAAK LDFAVSR LYSIASSAIGDFGDSK	50.1 64.7 57.6 79.6
fructose- bisphosphate aldolase, chloroplastic (Glycine max)	gi 1168411 gi 217072476 gi 22633 gi 84468410	42 kDa	[[65]2-975]([[10]5-718]	[[46]1-600]([[25]2-070]	Calvin cycle/glycolysis	EAWGLAR LASIGLENTAANR SNSIAQLGK	54.4 95.7 66.1	INDVLEQNIIVPGIK LASIGLENTAANR	59.2 75
						ALFSQITTR QNIGAADDVIVGDIR	52.2 93.4	ALFSQITTR QNIGAADDVIVGDIR	52.2 93.4

glyceraldehyde-3-phosphate dehydrogenase A subunit (Glycine max)	gi 77540210	43 kDa	[[35]3,347]],[125]4,123]],[140]1,655]],[20]5,718]],[21]4,513]],[24]2,072]	Calvin cycle/glycolysis	DSPLDVAINDTGGVK TFAEVNAAFR VVDLADIVANK	51.5 49.7 64.2	[[46]1,600]],[40]1,655]],[20]5,718]],[21]4,513]],[24]2,072]	DSPLDVAINDTGGVK TFAEVNAAFR VVDLADIVANK	56.3 59.7 55.5
pentapeptide repeats-containing protein (Medicago sativa)	gi 217071608	26 kDa	[[65]1,289]	unknown	DSPLDVAINDTGGVK VVDLADIVANK	63.3 68.8	[[25]1,289]	AVALVPLTK TFAEVNAAFR VVDLADIVANK	56.5 62.3 72.4
14-3-3-like protein, protein kinase C inhibitor homologue (Oenothera elata subsp. hookeri) [®]	gi 1168195 gi 15778152 gi 235641583 gi 28434607 gi 2879818 gi 3912948	29 kDa	[[50]2,497]],[33]3,577]],[22]2,444]	signal transduction	EAAESTLSAYK IISSEIQK	55.9 61	[[36]1,760]	DSTLIMQLLR EAAESTLLAYK IISSEIQK TVEVEELTVEER YEEIMVEFMEK	58.2 81.6 83.7 69.7 67.7
glutathione-S-transferase (Glycine max) [®]	gi 255627415 gi 255640468	23 kDa	[[19]2,226]],[20]2,226]],[10]3,022]	redox homeostasis	EAAESTLSAYK IISSEIQK	65.9 56.2	[[19]2,226]],[20]2,226]],[10]3,022]	VPVVLFDGK VTAVDLSLAPK	59.4 59
ascorbate peroxidase 2 (Glycine max) [®]	gi 1336082 gi 1420938 gi 16304410 gi 187962070 gi 217072458 gi 235636684 gi 37196687 gi 4406539	27 kDa	[[13]2,805]],[20]2,226]	redox homeostasis	VPVVLFDGK VTAVDLSLAPK	51.1 58.6	[[10]2,805]],[13]2,226]	VPVVLFDGK VTAVDLSLAPK EGLLQLPSDK TGGPPFGTIK	50.8 56.2 56.6 56.7
			[[20]2,226]		ALLSDPVRFLVEK EGLLQLPSDK TGGPPFGTIK	60.9 52.3 48.5	[[20]2,226]		

ribose 5-phosphate isomerase type A (RPLA) subfamily protein (Glycine max) ¹⁾	gi 255640161				[66]1.271]	Calvin cycle			[66] FVWDDTK LAADKAVESVK LGALLASGQLSDIVGVPTSK LQELFKEEGVEAK SGMVLGLGTGTAAFVWAK SLGIPLSVLDNPNR	56.1 71.1 92.7 71.2 89.4 101
14-3-3 family protein (Vitis vinifera) ²⁾	gi 225451995 gi 255638346 gi 3023194	30 kDa	[(39)3.357]		[(36)1.760]	signal transduction		[33] EAAESTLLAYK IISIEQK IISIEQKEESR	58.2 54.8 56.5	
dieneolone hydrolase family protein: chlorophyllase-like (Glycine max) ³⁾	gi 255629231 gi 255645019	17 kDa			[(10)3.022]	chlorophyll metabolism: photosynthesis			[10] DDTTFDAYVVGK ISQLGSGFK	58.2 81.6 63.7 69.7 67.7
polyketide cyclase 2 superfamily protein (Glycine max) ⁴⁾	gi 255628305	17 kDa			[(65)-1.289]	secondary metabolism		[65] AIVLDASNVFPK LTLAEGILGVYK LVATPDGGSVK SVETIEGDGGPGTIK	53.5 61.8 51.7 84.7	
proteasome subunit alpha type-6 (Glycine max) ⁵⁾	gi 12229897 gi 255648341	27 kDa			[(13)2.805]	protein degradation		[13] FTYGYEMVVDVLAK LFQVEYAFK YLGLLATGMITADAR	63.1 56.5 54.3	
malate dehydrogenase (Glycine max) ⁶⁾	gi 5929964	36 kDa	[(10)5.718]([21)4.533]		[(25)2.070]([24)2.072]	TCA cycle		[10] ALEGADVVIIPAGVPR LFGVTTLDVWR VAVLGAAGGIGQPLSLMLK [21] ALEGADVVIIPAGVPR LFGVTTLDVWR VAVLGAAGGIGQPLSLMLK	72.5 83.1 69.3 67.3 74.4 71.1	
cytosolic malate dehydrogenase (Cicer arietinum) ⁷⁾	gi 10334493 gi 217073248 gi 27462762 gi 27462764 gi 77999077 gi 83283965	35 kDa	[(10)5.718]		[(25)2.070]	TCA cycle		[10] LDLTAEELSEEK MELVDAAPFLK VLVVPANPANTNALLK	73.8 52.3 63.2	

triosephosphate isomerase (Glycine max)	gjl77540216	27 kDa	[(33)3.577]([96)1.545]([1.222]([75)1.880])	[(15)2.471]([36)1.760]([1.1)1.88])	<p>[(33)] EAGTTTAVVAEQTK IVTTLNEAK</p> <p>[(96)] EAGTTTAVVAEQTK IVTTLNEAK QLLINESNEFVGDK VAYALQQGLK</p> <p>[(104)] AEFVDIINAATVK EAGTTTAVVAEQTK IVTTLNEAK QLLINESNEFVGDK VAYALQQGLK</p> <p>[(75)] AEFVDIINAATVK IVTTLNEAK QLLINESNEFVGDK VAYALQQGLK</p>	<p>[(15)] AEFVDIINAATVK IVTTLNEAK QLLINESNEFVGDK VAYALQQGLK</p> <p>[(36)] IVTTLNEAK VAYALQQGLK</p> <p>[(31)1.88] AEFVDIINAATVK QLLINESNEFVGDK VAYALQQGLK VIACIGETLEQR</p>	<p>62 57.8</p> <p>65.8 51.9 72.1 56.2</p> <p>57.4 61.2 65.6 68 56.4</p> <p>70 65.5 69 49.1</p>	<p>70 65.5 69 49.1</p> <p>59.2 56</p> <p>75.7 74.2 52.4 72.9</p>
Chaperonin 10 kD subunit (cpn10 or GroES) (Glycine max)	gjl255645102	27 kDa	[(28)1.958]([42)1.615]([1.0)3.022])	protein folding	<p>[(28)] DDDVGILTTDDIK DGSDYITLR TSGGLLLEATK YAGTEVDFDGTK</p> <p>[(42)] DGSDYITLR TSGGLLLEATK YAGTEVDFDGTK</p> <p>[(10)] DGSDYITLR TSGGLLLEATK</p> <p>[(6)] ALVTDADNVIPK GITFEDETTSPVAPATLYK ITFVEDGESK SVENLEGNNGGPGTIK</p> <p>[(65)] ITFVEDGESK SVENLEGNNGGPGTIK</p>	<p>56.3 56.1 83.7 87</p> <p>45.8 60.9 59.8</p> <p>56.4 60.8</p> <p>71.2 52.4 51.2 83.3</p> <p>54.8 76.9</p>		
polyketide cyclase 2 superfamily protein (Glycine max)	gjl18643 gjl255640867	17 kDa	[(6)3.781]([6.5)1.289])	secondary metabolism	<p>[(6)] ALVTDADNVIPK GITFEDETTSPVAPATLYK ITFVEDGESK SVENLEGNNGGPGTIK</p> <p>[(65)] ITFVEDGESK SVENLEGNNGGPGTIK</p>	<p>56.4 60.8</p> <p>71.2 52.4 51.2 83.3</p> <p>54.8 76.9</p>		

Supporting Information Table 2. List of Peptides Identified from 115 ppb Root Proteins that are Differentially Expressed/Oxidized

Protein	Accession Number(s)	Mol Weight	Fold Change IAF (spot number)	Fold Change SYPRO (spot number)	Classification	IAF Peptides (spot number)	Best Mascot Ion Score	SYPRO Peptides (spot number)	Best Mascot Ion Score
TM phosphate binding protein (Glycine max)	gi 255634120	20 kDa		[(40)-1.573]	Calvin cycle/glycolysis			[(40)] EAGTTTAVSEGTK IVITLNEAK QLLNESNEFVGDK	64.6 64.8 90.2
polyketide cyclase 2 superfamily protein (Glycine max)	gi 255631546	17kDa	[(47)1.717]		secondary metabolism	[(47)] ALVTDADNIIPK ITFVEDGETK SVENVEGNGGFGTIK	71.8 63 61.8		
polyketide cyclase 2 superfamily protein (Glycine max)	gi 255627117	18 kDa	[(13)2.461]	[(63)-1.303]	secondary metabolism	[(13)] FIFQAIDNDHGHGTIIK HWTYTIDGK LFSGDIDHNYK	40.7 59.7 62.9	[(63)] ITTEIGVHATATK LFSGDIDHNYK	52.2 52.3
Hossmann-fold NAD(P)(+)-binding protein, isoflavone reductase-like (Glycine max)	gi 255637531	34 kDa	[(59)1.544]		secondary metabolism - flavonoid	[(59)] AGNPTFALVR VILGDGNPK	60.3 63		
similar to heat shock cognate 70 kDa protein 1 (Vitis vinifera)	gi 225434984 gi 225434986 gi 225434984 gi 225434996	71 kDa		[(44)-1.514]	protein folding			[(44)] FSDSSVQSDIK NCQVANNPNTVEDAK TTFPSYVAFTDIER VEIANDQGNR	69.1 86.1 56.7 50
chaperonin GroEL (Bradyrhizobium japonicum USDA 110)	gi 27377170 gi 27380737 gi 27382090	58 kDa		[(64)1.273]	protein folding			[(64)] LAGGVAVIR LENVTLNMLGR SVAAGMNPMDLK VGGATEVEVK	52.9 67.6 55.2 55.7
glutathione-S-transferase (Glycine max)	gi 255629025 gi 255646535	24 kDa	[(59)-1.573]		redox homeostasis	[(59)] FSGAAAPAEAAPAK SIEMPGLLWGASK	58.7 59.7		
F1 ATPase (Pisum sativum)	gi 2116558	60 kDa		[(64)1.273]	ATP-coupled proton transport			[(64)] ITDEFTEGK TIAMDATEGVVR	58.4 71.4
F0-F1 ATP synthase subunit beta [Rhodospseudomonas palustris BisA53]	gi 115522308 gi 27375551 gi 283839557 gi 39933253 gi 75674631 gi 85713719 gi 86747391 gi 91975039 gi 92116150	51 kDa		[(64)1.273]	ATP-coupled proton transport			[(64)] FTOAGSEVSAALLGR FVDLADTIK VVDLLAPYAK	68.8 56.3 66
ferric leghemoglobin reductase-2 precursor (Glycine max)	gi 3309269	53 kDa	[(70)-1.250]	[(44)-1.514]	nodule metabolism	[(70)] AIDNAEGLVK VVGVDTSGDGVK VSSSTGALALTEIPK	55.8 74.9 72.9	[(44)] AIDNAEGLVK VVGVDTSGDGVK	52.1 57.4

cytophlin (Phaseolus vulgaris)	gi 145049729 gi 254047060 gi 289780455 gi 829119	18 kDa	[(13):2..461]		protein folding	IVFELYADVTPR VFFDMITGGQPAGR	67.2 62.5		
RubisCO large subunit-binding protein subunit alpha (Brassica napus)	gi 1351030 gi 464727	58 kDa	[(20):2..187]	[(9):2..132]	Calvin cycle/glycolysis	LLVEFENAR VGAATETELED	58.9 75.9	49.2 69.6	
dirigent-like protein (Glycine max)	gi 255629177	21 kDa		[(63):-1..303]	secondary metabolism - lignin			58.1 79.4	
plant basic secretory protein (Glycine max)	gi 255627793	25 kDa		[(52):-1..423]	unknown				
ascorbate peroxidase (Glycine max)	gi 110590276 gi 110591017 gi 161761102 gi 161761104 gi 29726917 gi 310561 gi 37196683	28 kDa		[(44):-1..514]	redox homeostasis			62.6 74.2 56.1	
NADPH-dependent FMN reductase (Glycine max)	gi 255630927	22 kDa		[(32):-1..704]	redox homeostasis			54.8 100 77	
NADPH-dependent oxidoreductase, putative (Glycine max)	gi 255644585	38 kDa	[(59):1..544]	[(70):-1..228]	redox homeostasis	AYTFGSPPLIGYGVSK ESDMNVESTITLK LPQGSNDVLLK	58.1 64.5 55	75.2 57.3	
caireticulin-1 (Glycine max)	gi 117165712	48 kDa	[(70):-1..250]		protein folding	FYAISA EYPEFSNK KPEGYDDIPK QTGSLYSDWDLPPK SGTLFDNVLITDPEYAK TEAGEDTKEEGVHDEL YVGIELWQVVK	78.1 66.2 65.6 111 58.7 56.5		
transaldolase-like (Glycine max)	gi 255646850	48 kDa	[(53): 1..573],[(27): 1..983]		secondary metabolism	AISSNAYNDQFR LADDTTEGTEAAK	80.5 83.2		
reversibly glycosylated protein (Phaseolus vulgaris)	gi 38194918	40 kDa		[(16):1..942]	unknown			55.3 57.8	
reversibly glycosylated protein (Glycine max)	gi 255645037	42 kDa		[(16):1..942]	unknown			64.6 60.3	

peptidase M17 (Oryza sativa japonica)	gii115449199 gii238013218 gii242066822 gii293332992 gii75261364	33 kDa	[[64]]1.273]	proteolysis			FDMGGSAAVFGAAK TIEVNTDAEGR	57.6 64.6
ADP-Ribosylation Factor 1 Complexed With GDP, Full Length Non- Myristoylated (Homo sapiens)	gii1065361 (+113)	21 kDa	[[62]]-1.308]	vesicle transport			ILMVGDAAGK QDLPNAMNAEITDK	58.9 58.3
pyruvate dehydrogenase (acetyl)-transferring)	gii255635250 gii255635914	39 kDa	[[16]]1.942]	TCA cycle			EGISAEVNLR VLSPYSSEDFAR	58.4 51.9
alcohol dehydrogenase 1 (Pisum sativum)*	gii113361 gii113366 gii229464655 gii229464657 gii229464659 gii237512195 gii255638965 gii23773061 gii4039115 gii452767 gii452769 gii1587338 gii1651638	41 kDa	[[16]]1.942] [[9]]2.132]	carbon metabolism			FGVNEFVNPK IIGVDLVSSR FGVNEFVNPK IIGVDLVSSR	59.1 72.5 58.4 66.6
molecular chaperone DnaK, provisional (Vitis vinifera)*	gii118488840 gii147860809 gii166919372 gii224120086 gii224129424 gii225456004	74 kDa	[[56]]1.397]	protein folding			FGVNEFVNPK IIGVDLVSSR FGVNEFVNPK IIGVDLVSSR	59.1 72.5 58.4 66.6

proteasome subunit alpha type-5 (Glycine max)*	gii12229923 gii217071324 gii217071540 gii224119566 gii225441983 gii255584432 gii255626321 gii255641541 gii255647525 gii255647791 gii297742930	26 kDa	[(53)-1.573]		protein degradation	AIGSGSEGADSSLOEQYNK GVNTFSPEGR ITSPLLEPSSVEK	135 62.9 68.1	
copper chaperone homolog CCH (Glycine max)*	gii6525011	14 kDa	[(62)-1.308]		metal trafficking	AIGSGSEGADSSLOEQYNK GVNTFSPEGR ITSPLLEPSSVEK	135 62.9 68.1	
copper chaperone homolog CCH (Glycine max)*	gii6525011	14 kDa	[(62)-1.308]		metal trafficking			61.2 71.2 58 51
eukaryotic UGIFase (Glycine max)*	gii255635072	51 kDa	[(64)1.273]		nucleotide metabolism			88.4 80.1 72.3
cystathionine beta-synthase domain-containing protein (Glycine max)*	gii255631750	23 kDa	[(2)4.512]		amino acid biosynthesis	AIAGITER GMVGMVSGDVVR LITVTPDK SMTQNNVGALVVVK VLQAMQLMTDK	62.9 65 54.6 81.1 58.8	
cyclophilin (Glycine max)*	gii17981611 gii255628137	18 kDa	[(13)2.461]. [(4)4.207]		protein folding	IVMELYADVTPR VFFDMTIGGGSAGR	67.2 53.7	
YjGF YER057c UK114 family	gii255629976 gii255632812	20 kDa	[(39)1.600]		unknown	IVMELYADVTPR VFFDMTIGGGSAGR AGGSYSSVVK TTIMLADLK	74.8 65.5 66.9 55.5	

ATP synthase subunit alpha, mitochondrial (Helianthus annuus)	gii114404 gii114411 gii15429015 gii224365668 gii22742 gii231585 gii2645893 gii28276712 gii296040811 gii357982 gii5305369 gii543866 gii74181742 gii903732	55 kDa	[(48)-1.730]	ATP-coupled proton transport	AAEITLLESR TAIADTILNOK WVSVGDGIAR	61 61.9 50.6	
14-3-3-like protein, protein kinase C inhibitor homologue (Oenothera elata subsp. hookeri)	gii1168195 gii15778152 gii255641583 gii26454607 gii2879818 gii3912948	29 kDa	[(33)1.659]([52)-1.423]	signal transduction	DSTLMIQLLR EAAESTLSAYK IISIEIQEESR KEAAESTLSAYK LVPSAAAGDSK VSSAAESELTVEER YEEWVEFMEK	58.1 69.1 72.9 56.8 61.3 90.2 56.2	
glutathione-S-transferase (Glycine max)	gii255627415 gii255640468	23 kDa	[(48)-1.753] [(40)-1.573]	redox homeostasis	VPVVLFDGK VTAVDLSLAPK	57.4 67	
translationally controlled tumor protein (Elaeis guineensis)	gii192910898 gii192912974 gii192912976 gii20140683 gii75222628	19 kDa	[(62)-1.308]	cell expansion	LDAEQQLFK LQEQAPDKK QFVTFMK VVDIVDTFR	47.6 57 48.5 49.3	
dirigent-like protein (Glycine max)	gii255626867	21 kDa	[(63)-1.303]	secondary metabolism - lignin	AQGLFGLASLEDR EMPIVGGTGVFR IIEPSASEVR	82.3 60 75.8	
Universal stress protein family protein (Glycine max)	gii255626103 gii255628729	18 kDa	[(63)-1.303]	stress response	EKIVEAVGDLK IGVALDFSK IVEAVGDLK KIGVALDFSK	50.7 53.2 64.7 51.4	
poly(ADP-ribose) cyclase 2 superfamily protein (Glycine max)	gii255628305	17 kDa	[(47)1.717] [(39)1.600]	secondary metabolism	AVLDAASNVFPK GDEOLAEEYVK LTLAEGLYVK SVETIEGGGGFTIK	66.4 56.2 57.8 70.5	
proteasome subunit alpha type-6 (Glycine max)	gii12229897 gii255648341	27 kDa	[(61)-1.326]	protein degradation	AAGITSIGVR ATEIEVGWVR YLGLLATGMTADAF	88.1 51.4 84.7	

malate dehydrogenase (Glycine max)	gii15929964	36 kDa	[(59)]1.544]	[(70)-1.228]	TCA cycle	ALEGADVWIIIPAGVPR SEVWGYQGEELGK	ALEGADVWIIIPAGVPR DDLFINAGIVK LFGVTTLLDVVR SEVWGYQGEELGK TODGGTEVVEAK	57 60.6 77.1 87 54.9
cytosolic malate dehydrogenase (Cicer arifelinum)	gii10334493 gii217073248 gii27462762 gii27462764 gii77999077 gii83283965	35 kDa	[(59)]1.544]		TCA cycle	ALGQISER EFAPSIPEK LDLTAEELSEEK MELYDAAFLLK VLVVA NPANTNALIK		62.8 56.1 78.3 55.3 55.4
polyketide cyclase 2 superfamily protein (Glycine max)	gii118643 gii255640867	17 kDa		[(39)]1.600]	secondary metabolism		ALVTDADNVPK AVEAYLLANPHYN ITFVEDGESK KITFVEDGESK SVENLEGNNGGFGLIK	49 62.5 59 61.5 71.7

Supporting Information Table 3. List of Peptides Identified from 60 ppb Leaf Proteins that are Differentially Expressed/Oxidized

Protein	Accession Number(s)	Mol Weight	Fold Change IAF (spot number)	Fold Change SYPRO (spot number)	Classification	IAF Peptides (spot number)	Best Mascot Ion Score	SYPRO Peptides (spot number)	Best Mascot Ion Score
polyketide cyclase 2 superfamily protein (Glycine max)	gi 255630540	17 kDa	[(79)2.467]		secondary metabolism	[(79)] LTFVEDGQTK LVADPNGGSIK	50.7 69.7		
dehydrogenase [NADP+], chloroplastic (Pisum sativum)	gi 462579	48 kDa	[(128)-1.547]		TCA cycle	LASGEVFGPDDPALK TEAELLAEK	69.6 55.8		
ribulose-1,5-bisphosphate carboxylase/oxygenase large subunit (Ostryopsis davidiana)	gi 5731964	52 kDa		[(53)1.520]	Calvin cycle			[(53)] DTDILAAFR XKDDTDLAAFR	46.1 54.4
peroxiredoxin (PRX) family protein, 2-Cys (Glycine max)	gi 255641409	22 kDa		[(50)-1.620]	redox homeostasis			[(50)] SGGLGLNYPPLISDVTK SYGVLIPDQGIAR	92 54.8
translation elongation factor-TU (Glycine max)	gi 2548952	27 kDa		[(77)-1.281]	translation			[(77)] ILDEALAGDNNVGLLLR LMDEVDDYIPFQR NITTVIGVEMFQK VGETVDLVGLR	79.4 63.4 49.1 83.6
plastid high chlorophyll fluorescence 136 precursor (Zea mays)	gi 148251625 gi 218198925 gi 222636266 gi 224096552 gi 225423755 gi 242097170 gi 75252730	43 kDa		[(64)-1.407]	photosynthesis			[(64)] AADNAAANLYSVK SIPSAEDEDFNFR	71.8 62.8
thiamine pyrophosphate (TPP) family protein (Populus trichocarpa)	gi 224063766	81 kDa	[(127)-1.575];[(135)-1.468]		cofactor metabolism	[(127)] ALPTYTPEPADATR FLAIDAVEK KYSEEAELK NGNTGYDEIR NLQQNLNALVK YSEEAELK [(135)] KYSEEAELK NGNTGYDEIR NLQQNLNALVK VLPGLLGGADLASSNMTLK YSEEAELK	46.1 52.3 81.4 57 48.1 49.5 69.9 66.7 75.8 74.9 65.9		

23 kDa OEC protein (Salsicornia veneta)	gii14855011 gii197691941 gii253635846	22 kDa	[(75)-2.599]		photosynthesis	EVEYFGQVLR QYYSITVLTIR	51.9 48.4		
glutamine synthetase precursor (Glycine max)	gii13877511 gii255648131	48 kDa	[(128)-1.574]		amino acid metabolism	AAEIFSNPK EDGGFEVIKK LEGLLNLDITPFDK	72.1 56 58.6		
ribulose 1,5- bisphosphate carboxylase small subunit precursor (Glycine soja)	gii1079736 gii10946375 gii10946377 gii10946379 gii132086 gii132113 gii255625881 gii255630492 gii3914590	20 kDa		[(37)1.845]	Calvin cycle	FETLSYLPDLDDAQLAK IIGFDNVR SPGYDGR			87 67.1 53.1
Stress-induced protein SAM22 (Glycine max)	gii134194 gii22218276 gii229597555	17 kDa		[(30)-1.984]	stress response				62.3 68.5 49.1 49 51.9 60.3 77.3
ribulose-1,5- bisphosphate carboxylase/oxygen ase large subunit (Orothamnus zeyheri)	gii1045644 (+1547)	50 kDa		[(61)1.421],[53]1.520]	Calvin cycle				56.9 60.1 46.1 54.4
rubisco activase (Glycine max)	gii290766485	49 kDa	[(128)-1.547]		Calvin cycle	GLAYDISDDQDDTR LLTYGNMLVQEQENVK MGINPIMMSAGELESNGAFEP AK VPLJLGIWGGK YLNEAALGNANEDAINR	98.2 85.2 89.6 56.4 107		
peroxiredoxin 5-like (Glycine max)	gii255640689	25 kDa	[(79)2.467]		redox homeostasis	VNEEVLLISDGGNGTFTK YALLAEDGVVK	56.2 67.1		
carbonic anhydrase (Glycine max)	gii255630357 gii270342124	24 kDa		[(8)-3.313]	carbon metabolism	EYEQAIIEELQK NVANIVPPYDOSK			49.7 47.9

Phosphoglycerate kinase, chloroplastic (Nicotiana tabacum) ¹	gi 2499497	50 kDa	[(128)-1.574],[(133)-1.513]	[(77)-1.281]	Calvin cycle/glycolysis	ADLNVPDINQINITDDTR ELDYLGVAVSNPK GVTTIIGGDSVAAVEK IGVIESLLEK LSELLGIQWVK	68.5 72.4 73.2 71.3 54.2	ELDYLVGAVSNPK GVTTIIGGDSVAAVEK YSLAPLVPR	56.3 62.7 58.5
CYPOR-like ferridoxin reductase (Pennislvs tricarum) ¹	gi 224074257	40 kDa		[(64)-1.407]	redox homeostasis			GIDDIWVSLAAK LYSIASSAIGDFGDSK	78 87.4
bisphosphate aldolase chloroplastic (Glycine max) ¹	gi 1168411 gi 217072476 gi 22633 gi 84468410	42 kDa	[(109)-1.369]		Calvin cycle/glycolysis	EAAWGLAR LASIGLENTANR SNSLAQLGK	57 91.3 68.4		
glyceraldehyde-3-phosphate dehydrogenase A subunit (Glycine max) ¹	gi 77540210	43 kDa	[(112)-1.811],[(82)-2.207]		Calvin cycle/glycolysis	AVALVPLTK VVDLADIVANK	53.5 77.2		
pentapeptide repeats-containing protein (Medicago truncatula) ¹	gi 217071608	26 kDa	[(79)2.467]		unknown	TFAEEVNAAFR VVDLADIVANK	54.4 76.5		
transcriptionally controlled tumor protein (Elaeis guineensis) ¹	gi 192910898 gi 192912974 gi 192912976 gi 20140683 gi 7522628	19 kDa		[(52)-1.581]	cell expansion	FDGADMTEVMSK GVDFSNVAIDLR	60.7 68.3	LDAEQEELFK VVDIVDTR	66.1 69.6
triosephosphate isomerase (Glycine max) ¹	gi 77540216	27 kDa	[(38)1.834],[(8)-8-3.313],[(82)1.210]		Calvin cycle/glycolysis			EAGTTTAVVAEQTK IVTTLNEAK QLLNESENFYGDK VAYALQQGLK	48.8 49.7 61.7 53.4
Chaperonin 10 kD subunit (cpn10 or GroES) (Glycine max) ¹	gi 255645102	27 kDa	[(75)-2.599]		protein folding	DGSDYITLR TSGGLLLEATK YAGTEVDFDGTK	74.7 70.3 67.7	IVTTLNEAK QLLNESENFYGDK VAYALQQGLK	52.3 53.6 61.9 63.7 78.3 52

polyketide cyclase 2 superfamily protein (Glycine max)	gi 18643_gli235640667	17 kDa	[(67)2.888]	secondary metabolism	[(67)] ALVTDADNVIPIK AVEAYLLANPHYN ITFVEDGESK	66.4 46.3 51.7	
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Supporting Information Table 4. List of Peptides Identified from 60 ppb Root Proteins that are Differentially Expressed/Oxidized

Protein	Accession Number(s)	Mol Weight	Fold Change IAF (spot number)	Fold Change SYPRO (spot number)	Pathway	IAF Peptides (spot number)	Best Mascot Ion Score	SYPRO Peptides (spot number)	Best Mascot Ion Score
triosephosphate isomerase (Glycine max)	gi 255645535	33 kDa	[(44):1.750]	[(33)-1.776]	Calvin cycle/glycolysis	[(44)] GGAFTGEISAEQLK GPEFATVNSVTSK IEISAQNSWVGK IYGGSVNGGSAELAK LVADLNSAK	75.2 82.3 57.3 56.6 68.5	[(33)] AEFVDINAATVK EAGTTTAAVAEQTK QLLNESNEFYGDK VAYALQQGLK	74.4 60.8 82.3 54.7
methionine synthase (Glycine max)	gi 33325957	84 kDa	[(13)-3.129] [(67):1.276]	[(58)-1.223] [(48)-1.539]	amino acid biosynthesis	[(13)] IPPTEEIADR YLFAGVVDGR	66.4 65.3	[(58)] IPPTEEIADR YLFAGVVDGR	50.3 71.7
putative fructokinase 2 (Vitis vinifera)	gi 225453918	35 kDa		[(58)-1.223]	carbohydrate metabolism	[(67)] IPPTEEIADR IVEVNALAK YLFAGVVDGR	58.8 54.2 52.8	[(48)] IVEVNALAK YLFAGVVDGR	57.5 84.3
nucleoside diphosphate kinase 1 (Glycine max)	gi 2498078 gi 26245395	16 kDa	[(3):4.559]	[(36)-1.719] [(64):1.748]	nucleotide metabolism	[(3)] GLIGEISR IIGATNPAGSEPGTIR	64.2 60.9	[(36)] GLIGEISR IIGATNPAGSEPGTIR KIIGATNPAGSEPGTIR LVTVDPRPFAEK	68.6 87.2 57.6
Hosmann-told NAD(P)+-binding protein, possible isoflavone reductase homolog 2 (Glycine max)	gi 255637547 gi 255640090 gi 6573171	34 kDa	[(59):1.471]		secondary metabolism - flavonoid	[(59):1.471] FIVEASAK ILFIGGTGYIGK	52.7 63	[(34)] GDFALDGR GLIGEISR IIGATNPAGSEPGTIR KIIGATNPAGSEPGTIR LVTVDPRPFAEK	57.3 66.3 80.5 91.9 66.7
dimethylmenaquinone methyltransferase (Glycine max)	gi 255630950	18 kDa	[(27)-2.309]	[(16)-2.191]	secondary metabolism	[(27)-2.309] QVFSGPVTLK QVFNINAGTR VFEDNLVTR VLVDVGGASLR	48 56.8 62.2 69	[(16)] QVFSGPVTLK VFEDNLVTR VLVDVGGASLR	48 65.3 53.7

transaldolase-like protein (Glycine max)	gi 255646850	48 kDa	[(46)-1.736],[(31)-2.181],[(33)-2.034]	[(46)-AISSSNAYNDQFR LADDTTEGTEAAK VTSVASFFVSR [(31)-AISSSNAYNDQFR GVTSNPAIFOK LADDTTEGTEAAK VTSVASFFVSR [(33)-AISSSNAYNDQFR GVTSNPAIFOK IGTVPALNLR LADDTTEGTEAAK	secondary metabolism	85.7 107 60.8	83.1 54.2 66.6 73.3	90.3 54.4 49 92	[(60)-FSDSSVQSDIK NOVAMNPINVFDAK TTPSVVGFDTIER VEIANDQGNR [(57)-FSDSSVQSDIK NOVAMNPINVFDAK TTPSVVGFDTIER [(62)-FSDSSVQSDIK TTPSVVGFDTIER [(64)-FSDSSVQSDIK NOVAMNPINVFDAK TTPSVVGFDTIER [(46)-IVSSIEQKEEGR VVVGSTPASELTVEEER	83.1 80.7 60.3 67.2	71.6 57.2 54.6	
cell-autonomous heat shock cognate protein 70 (Cucurbita maxima)	gi 26885223	71 kDa	[(56)-1.493],[(47)-1.696],[(62)-385],[(54)-1.558]	[(60)-1.204],[(57)-1.283]	protein folding	54.6 55.5	51.4 57.5 50.3	54.6 55.5	[(56)-NOVAMNPINVFDAK TTPSVVGFDTIER [(47)-FSDSSVQSDIK NOVAMNPINVFDAK TTPSVVGFDTIER [(62)-FSDSSVQSDIK TTPSVVGFDTIER [(64)-FSDSSVQSDIK NOVAMNPINVFDAK TTPSVVGFDTIER	83.1 80.7 60.3 67.2	71.6 57.2 54.6	
14-3-3-like protein B; SGF14B (Glycine max)	gi 3023195	28 kDa	[(46)-1.736]		signal transduction	63.1 90.2						
DHAR class glutathione transferase DHAR2 (Glycine max)	gi 255627415 gi 255640468	23 kDa		[(50)-1.514]	redox homeostasis				[(50)-VPVLFDDGK VTAVDLSLAPK	51.5 62.9		
glutathione-s-transferase (Glycine max)	gi 255625731	25 kDa		[(60)-1.204],[(50)-1.514]	redox homeostasis				[(60)-VLDVYEEER VYGPTYGSPK [(50)-VLDVYEEER VYGPTYGSPK	53.8 52.2		
proteasome beta type-6 subunit (Vitis vinifera)	gi 225429850 gi 255574159 gi 255625747 gi 298081796	25 kDa	[(47)-1.696]		protein degradation	76.5 60			[(47)-SGSAAQDSQIVSDYVR TVINSEGVTR	57.7 65.2		
proteasome alpha type 4 subunit (Glycine max)	gi 255637272	28 kDa		[(51)-1.473],[(33)-1.776]	protein degradation				[(51)-AAAIGANNQAAQSILK LELAEVFLSPSGK [(33)-AAAIGANNQAAQSILK LELAEVFLSPSGK	105 57.2		

S-adenosyl-L-homocysteinylase (Petunia x hybrida)	gi 32867697 gi 464734	53 kDa	[(57)-1.485],[(34)-2.014]	amino acid biosynthesis	LVGVSEETTTGVK VAVVAGYGDVGK	70.1 49.2		
protein disulfide isomerase (Glycine max)	gi 171854980 gi 49257109	59 kDa	[(57)-1.485]	redox homeostasis	LVGVSEETTTGVK VAVVAGYGDVGK	62.8 62.7		
peroxidase 1 precursor (Glycine max)	gi 255645056 gi 5002342	37 kDa	[(57)-1.485]	redox homeostasis	SADDEATAFIGENK VAIVGVFPK	67 65.2		
Aspartate aminotransferase P2, mitochondrial (Lupinus angustifolius)	gi 112979 gi 169915 gi 255648095 gi 25990362 gi 2605932 gi 3378163 gi 7548843	50 kDa	[(23)-1.998]	amino acid biosynthesis	GLDVVNOIK MGNIGVLTGSQGEIR	50.8 86.8	ATAPELLGADNPANIK LNLGVGAYR VATVQGLSGTGSLR	66.3 52.5 70.2
chain A, mutant beta-amylose (W55) (Glycine max)	gi 149241163 gi 157850459 gi 157850496 gi 169913 gi 231544 gi 46015332 gi 46015333 gi 46015334 gi 46015336 gi 46015825 gi 46015829 gi 50513924 gi 62122629 gi 62122631 gi 62122633 gi 62122635 gi 62738229 gi 62738230 gi 62738231 gi 71673371 gi 902938	56 kDa	[(49)-1.676],[(34)-2.014]	carbohydrate metabolism	LSDDLQK LSMFGVTVLR	57.5 76.5	LSDDLQK LSMFGVTVLR VAGENALPR	61 54.1 61.2
vacuolar ATPase subunit B1 (Triticum aestivum)	gi 125744990 gi 2493131 gi 2493132 gi 255660497 gi 6715512 gi 126236	54 kDa	[(23)-1.998]	ATP-coupled proton transport			AVVGVFEGTSGDINK TFVSLDMLGR TVSGVAGPLVLDK	74.9 56.3 54.3
leghemoglobin C1; nodulin-50 (Glycine max)	gi 126236	15 kDa	[(62)1.385]	nodulin metabolism	AVTDPOFVVK QEALVSSSEAFK	53.6 85.3		
NADPH:quinone reductase and related Zn-dependent oxidoreductase TRX family protein, possibly m-type (Glycine max)	gi 2556638934 gi 255631161	35 kDa 20 kDa	[(51)-1.473] [(56)1.493]	redox homeostasis	LNTDESPSTATR TTLTSSIEK	56.9 55	VGDEVYGDINKV VIGSLAEYTAEEER	61.2 67.3

pyridine nucleotide- disulphide oxidoreductase (Glycine max) actin (<i>Cryza sativa</i> japonica)	gi 255636578 gi 115484337 gi 168472715 gi 217072994 gi 225448323 gi 296085677 gi 9965319	44 kDa 42 kDa	[(56)-1..338] [(56)-1..338]	redox homeostasis cell structure	GTVAVGFTINSDDGEVK LTDVFGVGGADAK AGFAGDDAPR GEYDESGPAVHR	63.5 69.8 63.3 60.9
secretory peroxidase (Glycine max)	gi 255638280	38 kDa	[(67)1..276]([61)1..387]	redox homeostasis	[(67)] EIVNQFASDOK QLGGPFDVPLGR VSQLDVITDR [(61)1..387] QGVFTSDDDIAGSPK VSQLDVITDR	57.2 51.7 63.7 66.2 62.3
Glc EDJ BRP like 8 family (Glycine max)	gi 255628635	17 kDa	[(55)1..531]	unknown	[(55)] AAFGAEV(GR ANDAILFK	66.8 59.4
unknown (Glycine max)	gi 255647164	12 kDa	[(59)1..424]	unknown	[(59)] LVAEAAQSALK VADAAAGDLLDAAGK	59.3 85.9
Phosphoglucosyl transferase e. cytoplasmic (<i>Pisum sativum</i>)	gi 12585296 gi 12585316 gi 12585330 gi 12643355 gi 15220668 gi 15223226 gi 224099523 gi 22411476 gi 238479031 gi 238479033 gi 255573724 gi 26185954 gi 297841819 gi 297845376 gi 40233152 gi 62321043 gi 9295686	63 kDa	[(62)1..385]	carbohydrate metabolism	[(62)] LSGTGSEGATIR YLFEDGSR	88.1 51.6
predicted translation initiation factor 2B subunit glyoxalase I	gi 255647295 gi 255637721	23 kDa 32 kDa	[(67)1..276] [(39)-1..905]	translation carbon metabolism	[(67)] AIGSYGASIIQQQTEK GSLQLLDQR [(39)] EPGPVGGGTTVIAFVK SAEIVNQVIK TTSLFLDPDGWK	72.9 51 65.2 53.8 57.2

alcohol dehydrogenase 1 (Pisum sativum)*	gii113361 gii113366 gii229464655 gii229464657 gii229464659 gii237512195 gii255638965 gii29373061 gii4039115 gii452767 gii452769 gii51587338 gii61651638	41 kDa	[(67)1.276]([66)1.299]([58)1.477]([39)1.905]	[(48)-1.539]	carbon metabolism	[(67) FGVNEFVNPK IIGVDLVSSR [(66) FGVNEFVNPK IIGVDLVSSR [(58) FGVNEFVNPK IIGVDLVSSR [(39) FGVNEFVNPK IIGVDLVSSR	57.8 54.2 52.3 58.8 69.9 66.7 57.6 58.2	[(48)-1.539] FGVNEFVNPK IIGVDLVSSR	63.3 54.8
molecular chaperone DnaK, provisional (Vitis vinifera)*	gii118488840 gii147860809 gii166919372 gii224120086 gii224129424 gii225456004	74 kDa	[(52)1.460]	[(52)1.460]	protein folding	GPDGDVIDADFTDSK IAGLEVLR NQADSVYQTEK QFAAEISAOVLR	101 74.8 72.2 83.7		
proteasome subunit alpha type-5 (Glycine max)*	gii12229523 gii217071324 gii217071540 gii224119566 gii225441983 gii255584432 gii255626321 gii255641541 gii255647525 gii255647791 gii297742930	26 kDa	[(31)-2.181]	[(31)-2.181]	protein degradation	[(31) GVNTFSPGGR ITSPLLEPSSVEK LGSTAIGLK	63.6 68.8 53		
copper chaperone homolog COH (Glycine max)*	gii6525011	14 kDa	[(55)1.531]	[(55)1.531]	metal trafficking	[(55) AAFGAEEVGR ANDAILFFK	66.8 59.4		
eukaryotic UGPase (Glycine max)*	gii255635072	51 kDa	[(41)-1.629]	[(41)-1.629]	nucleotide metabolism	[(41) ISVANFGVK LDALLSQGK SAVAGLNEISEK	56 53.2 78.5		
cystathionine beta-synthase domain-containing protein (Glycine max)*	gii255631750	23 kDa	[(14)2.246]	[(14)2.246]	amino acid biosynthesis	[(14) AIAGITR GMVGMVSIQDVVR SMTNNNGALVVVK	60.2 58.2 96.5		

polyketide cyclase 2 superfamily protein (Glycine max)	gi 255628305	17 kDa	[(27)-2,309],[1(32)-2,215],[1(56)-493]		secondary metabolism	<p>[(27)] AIVLDASNVFPK GDEQLAEYYVK SVETIEGGDGGPTIK</p> <p>[(32)] AIEDFGANPDYN AIVLDASNVFPK GDEQLAEYYVK SVETIEGGDGGPTIK SVETIEGGDGGPTIK</p> <p>[(56)] GDEQLAEYYVK SVETIEGGDGGPTIK</p>	54.8 55.7 71.6 52 54.6 67.5 84.7 62.4 64.6 69.3		
proteasome subunit alpha type-6 (Glycine max)	gi 12229897 gi 255648341	27 kDa		[(44)-1,568]	protein degradation	<p>[(44)] AAGTTS[GVR ATEIEVGVVR</p>	74.9 66.7		
malate dehydrogenase (Glycine max)	gi 5929964	36 kDa	[(67)1,276],[1(58)1,471],[1(66)1,1,387],[1(66)1,238]	[(48)-1,539]	TCA cycle	<p>[(67)] ALEGADVVIIPAGVPR DDLFININAGIVK LFGVTTLDVVR</p> <p>[(58)] ALEGADVVIIPAGVPR DDLFININAGIVK LFGVTTLDVVR SEVVGYGDEELGK VAVLGAAGGIGGPLSLLMIK</p> <p>[(61)] ALEGADVVIIPAGVPR ANLDDDDVIK LFGVTTLDVVR SEVVGYGDEELGK</p> <p>[(66)1,298] DDLFININAGIVK SEVVGYGDEELGK</p> <p>[(58)1,471] KLDLTAEEELSEEK LDLTAEEELSEEK MELVDAAFPLK VLVVANPANTNALIK</p>	83.2 54.2 78.2 90.8 64.6 88 75.1 57 81.3 51.5 72.4 80.6 70.1 86.9		64 51.1 65.9
cytosolic malate dehydrogenase (Cicer arietinum)	gi 10334493 gi 217073248 gi 27462762 gi 27462764 gi 77999077 gi 83283965	35 kDa	[(58)1,471]		TCA cycle	<p>[(58)1,471] KLDLTAEEELSEEK LDLTAEEELSEEK MELVDAAFPLK VLVVANPANTNALIK</p>	62.9 79.1 60 63.8		

triosephosphate isomerase (Glycine max)	gi 77540216	27 kDa	[(66)1,298]([54)-1,558]([44)1,750]([60)-1,419]	[[67]-1,263]([51)-1,473]([39)-1,776]([47)-1,539]	alvin cycle/glycoly	[[68] AEFVDINAATVK EAGTTTAVAEQTK IVTLLNEAK QLLNESNEFVGDK [[64] EAGTTTAVAEQTK IVTLLNEAK QLLNESNEFVGDK [[44] EAGTTTAVAEQTK IVTLLNEAK QLLNESNEFVGDK [[60] EAGTTTAVAEQTK QLLNESNEFVGDK	55.7 54.4 49.3 83.3 70.8 52.1 84.1 72.9 52.8 71.5 81.2 64.9	[[67] EAGTTTAVAEQTK VAYALQQGLK [[61] AEFVDINAATVK EAGTTTAVAEQTK QLLNESNEFVGDK VAYALQQGLK [[33] AEFVDINAATVK EAGTTTAVAEQTK QLLNESNEFVGDK VAYALQQGLK [[47] EAGTTTAVAEQTK IVTLLNEAK QLLNESNEFVGDK VAYALQQGLK	52.9 51.9 71.5 70.8 86.3 57.5 74.4 60.8 82.3 54.7 62.4 80.5 82.8 55.6
Chaperonin 10 Kd subunit (cpn10 or GroES) (Glycine max)	gi 255645102	27 kDa	[[60]-1,204]([50)-1,514]	protein folding	[[60]-1,204] DGGSDYITLR TSGGLLITEATK [[50)-1,514] DGGSDYITLR GKDGSDYITLR KPLSVTPGNTVLYSK TSGGLLITEATK YAGTEVDFDGTK YTAIKPLGDR	61.2 55.9 58 56.3 77.6 80 80.6 56.3			

CHAPTER 5

CONCLUSIONS AND FUTURE DIRECTIONS

The goal of my thesis work was to obtain new insights into plant redox biology, with an emphasis on novel forms of redox-response due to oxidative stress. Specifically, I first sought to shed new light on the use of the glutathione homolog homoglutathione in legumes through structural and kinetic analyses of the substrate specificity responsible for the different modes of synthesis. Secondly, I have examined the regulation of the soybean thiol-redox proteome in response to changes in real-world oxidative conditions, namely field-exposure to increasing concentrations of tropospheric ozone.

In the preceding chapters, I described the methodologies used and results obtained in my efforts to meet my experimental goals. Based on my research results, I determined that synthesis of homoglutathione (hGSH), instead of glutathione (GSH), by the enzyme homoglutathione synthetase is largely specified by the replacement of two alanine residues in the alanine-rich loop with a leucine and proline (Chapter 2). In Chapter 3, preliminary efforts to test the methodology necessary for detecting differences in protein expression and thiol redox-state led to the conclusion that it was sound for application in future larger-scale experiments. Additionally, by comparing the changes in protein expression and thiol redox-state between *B. juncea* roots treated with BSO and H₂O₂, I was able to conclude that differing sources of ROS (exogenous versus endogenous) led to different protein redox responses. The tested methodology was then applied to tissue from soybean plants grown under various concentrations of tropospheric ozone in natural field setting (Chapter 4). From the results obtained, I was able to identify widespread and large-fold changes in the expression of key proteins which, contrary to what has been shown in acute exposure experiments, are largely involved in carbon fixation and flux

(glycolysis, Calvin cycle, and the TCA cycle). As indicated by their formatting, Chapter 2 and Chapter 3 were previously published in the journals *Plant Cell* and *Proteomics*, respectively. Final adjustments to Chapter 4 are currently underway, and it will be submitted to PNAS shortly.

Homoglutathione Synthetase and Molecular Diversity of Plant Glutathione Biosynthesis

As described in the preface to Chapter 2, it has been known for more than 20 years that some plants - specifically legumes and grasses - produce thiol-containing tripeptides besides GSH. Of these tripeptides, hGSH (from legumes) has been the most thoroughly investigated. Metabolite studies on the localization of hGSH indicate that its distribution varies across different tissue types in a species dependent manner [Moran et al, 2000; Matamoros et al., 1999]. In nodules, the most common site of localization, hGSH content further varies as a function of time till senescence and in response to changes in stressor concentrations [Loscos et al., 2008].

Like GSH, hGSH is synthesized in two ATP-dependent steps. While the first step's enzyme, γ -glutamylcysteine synthetase, is shared between the GSH and hGSH biosynthesis pathways, each pathway has a dedicated second enzyme. Although the activity necessary for hGSH synthesis was easily isolated, very little was known about the homoglutathione synthetase (hGS) enzyme itself [Macinol, 1987]. Beginning in 1993, a series of glutathione synthetase (GS) crystal structures were solved and published, indicating that the enzymes fell within the ATP-grasp superfamily of proteins [Yamaguchi et. al, 1993; Kato et. al, 1994; Polekhina et. al, 1999; Gogos and Shapiro,

2002]. Because of high homology between hGS and GS sequences, as well strong structural identity between ATP-grasp family members, it became apparent that hGS would look quite similar to GS overall. However, hGS used a different substrate (β -alanine) than GS (glycine); the subtle differences in residues and residue placement required for this shift in substrate specificity could not be determined by homology modeling and necessitated the acquisition of an hGS crystal structure.

Chapter 2, “Structural Basis for Evolution of Product Diversity in Soybean Glutathione Biosynthesis”, describes three x-ray crystal structures of hGS: the open-form apoenzyme, the open-form enzyme with γ -glutamylcysteine bound, and the closed-form enzyme with hGSH, ADP, and a sulfate ion bound. From these structures, in conjunction with the available hGS and GS sequences, it was concluded that two residues in the active site are primarily responsible for dictating substrate specificity between the two types of enzymes. In GS, terminal contacts with the carboxyl tail of glycine are provided by two alanine residues as part of the larger alanine-rich loop region. However, in soybean (and most other sequenced species) hGS, these two alanines are replaced by a leucine and proline; based on the hGS structure, these replacement residues pull the alanine-rich loop outwards by several Å, allowing the larger β -alanine molecule to be accommodated in the active site. The critical role of the leucine and proline residues in conferring substrate specificity to hGS was further confirmed by way of site-directed mutagenesis. Normally, hGS is not capable of catalyzing the synthesis of GSH at a physiologically relevant rate. However, the presence of one of two mutations - either leucine to alanine or proline to alanine, improves GSH catalysis by a factor of 10 or 100,

respectively. With the simultaneous inclusion of both mutations, GSH catalysis improves by nearly 1000-fold, demonstrating that the two mutations act synergistically to improve enzyme efficiency. While a 1000-fold improvement is significant, kinetic analysis of the *Arabidopsis* GS enzyme reveals a rate of glycine turnover nearly 10-fold higher than that of the mutated hGS. Thus, while the AA/LP site is clearly important for dictating substrate specificity, it can be concluded that other, as yet unknown residues also play a small but significant role in optimizing turnover rate.

A further conclusion obtained from the three solved hGS crystal structures, centers on the apparent domain movements required for enzyme catalysis. Based upon “snapshots” of various enzymes with various substrates bound, it was suspected that GS-type proteins went through some degree of domain movement as part of their catalytic cycle. However, as structures from a complete cycle were not available for a single enzyme, this suspicion could not be confirmed. By solving the structure of hGS in three forms - the apo enzyme, the intermediate enzyme- γ -glutamylcysteine complex, and the post-reaction complex - a more complete picture of the GS-type domain-movement cycle has been provided. Two regions - the lid domain (which includes the glycine-rich loop) and the previously described alanine-rich loop, appear to be the most dynamic and mobile portions of the hGS enzyme. Prior to substrate binding, these two domains are pulled back and away, allowing exposure of the active site to the surrounding environment. No change in domain structure is evident following γ -glutamylcysteine binding. When all three of γ -glutamylcysteine, ATP (represented here by ADP and a sulfate), and are in the active site, the lid domain swings inward to cover the ATP moiety, while the alanine rich

loops provides stabilizing contacts with both β -alanine and ATP. Once the reaction is complete, ADP and hGSH are released from the active site, and the enzyme resets itself into the open apoenzyme form.

As described above, x-ray crystal structures of hGS were solved in three forms. While these three structures represent key points in the reaction cycle, they do not portray every step in the mechanism. Because GS, and by extension hGS, utilize a random Ter-reactant mechanism, there are actually nine different possible combinations of reactants and products [Jez and Cahoon, 2004]. While some of these combinations are energetically unfavorable and unlikely to occur, others are part of the most-likely mechanism. In particular, the formation of an enzyme-ATP complex, and an enzyme-ATP- γ -glutamylcysteine complex is strongly favored and, in the case of the latter, necessary for completion of the reaction. In order to fill in the missing pieces of the reaction mechanism and provide insight into additional domain movements, it would be beneficial to obtain crystal structures of both of these hGS complexes in addition to those already solved. Specifically, a structure of the enzyme•non-hydrolyzable ATP analog• γ -glutamylcysteine complex could confirm whether, as indicated for the yeast GS, movement of the lid domain occurs before introduction of β -alanine into the active site pocket [Gogos and Shapiro, 2002].

There also exists the potential for a number of follow-up experiments involving the hGSH. Because hGSH has only been found in six legume species - none of which are closely related to one another - an obvious question to ask is whether hGSH is found in other legumes [Wojciechowski. et al., 2004]. This line of inquiry would necessitate the

use of HPLC-based profiling methods for various legume tissue types, and leads to questions regarding the evolutionary origins of hGSH. Because the available legume genome sequences show signs of at least one duplication event, it is likely that many other legumes contain multiple GS genes. However, there is no guarantee that these copies evolved into hGS in every species. For example, in broad bean, which does not produce hGSH, the evolution may never have occurred and/or the hGS gene(s) may be silenced. If only some legumes contain hGS genes, then it is likely that either hGS evolution was an independent event in each species or hGS evolved and then was subsequently lost in many legumes. The conclusion that hGS evolution was independent of course leads to a new query: why do all of the independently-evolved hGS enzymes utilize β -alanine over some other amino acid? This question would require an answer in two parts. First, it would be necessary to consider the availability of β -alanine (produced from uracil degradation or from spermine/spermidine) as compared to other amino acids [plantcyc.org]. Second, the specific properties, such as solubility, effectiveness as a redox buffer, and transportability, of hGSH versus other (artificially) synthesized tripeptides would need to be assessed. These experiments will also hopefully shed light on hGSH's role relative to GSH in legumes.

Redox Proteomics: Platform development with Brassica juncea (Indian Mustard)

Based on previous experiments [Jez et al., 2004; Hicks et al., 2007], it was determined that glutamate-cysteine ligase (GCL), the first enzyme in the GSH and hGSH biosynthesis pathways, is redox-regulated through the formation of two reversible

intramolecular disulfide bonds. One disulfide bond controls dimerization, while the second controls access to the active site; together under reducing conditions they inactivate the enzyme. Although GCL is not the only enzyme to utilize a thiol-based redox regulatory mechanism, it is one of the first - besides NPR1 - to be identified in plants. Prior work has largely focused on targets of the redox-regulatory proteins thioredoxin and glutaredoxin; while the column chromatography-based techniques employed are sound, they select for proteins with strong interactions, meaning that low abundance targets or targets that bind to a different redox-regulatory protein are missed [Balmer et al., 2003; Motohashi et al., 2001; Yano et al., 2001]. To circumvent this problem, an alternative non-column-based dual redox labeling strategy was developed. In short: the free thiol groups of proteins from a plant extract are first labeled with an alkylating agent. Next, oxidized thiols are reduced and labelled with a thiol-labile fluorescein derivative. The protein mixture is then separated by isoelectric point and molecular weight via 2D-SDS-PAGE. Protein spots containing oxidized thiols can be visually identified, as fluorescein derivatives fluoresce under a narrow range of wavelengths. The gel spots can then be excised, proteins digested, peptides analyzed by LC-MS/MS, and spectra matched via MASCOT to yield a likely peptide identity. Using this approach, the effect of different oxidizing treatments upon protein thiol state can be compared. If the gels are additionally exposed to a total protein stain, then changes in total protein expression as a result of treatment can be compared as well.

Chapter 3, “Redox-regulatory mechanisms induced by oxidative stress in *Brassica juncea* roots monitored by 2-DE proteomics”, describes a series of experiments in which,

B. juncea roots were exposed to different oxidizing agents, and the resulting redox and total protein expression changes were measured. Following 2D-SDS-PAGE, treatment of roots with either BSO or H₂O₂ yielded 50 and 59 gel spots, respectively, that differed significantly in thiol redox-status between the control and treated samples. A further 40 and 27 spots, respectively, displayed significant and non-overlapping changes in protein expression between the control and treated samples. From these initial results, it was concluded that both BSO and H₂O₂ directly affect both thiol-redox status and expression of root proteins. A closer look at the significant spots detected revealed that a subset of the detected spots - approximately 11-17 depending on the treatment combination - contained a single protein, while the remainder contained multiple proteins that co-migrated on the 2D-gel. Because of the difficulty in assigning definitive redox and expression fold change values to a single protein within a multi-protein spot, only spots containing a single protein were analyzed in greater detail. Of those 52 single protein spots, a large number were found to contain proteins involved in maintaining redox homeostasis. Given that BSO and H₂O₂ are a source of endogenous and exogenous ROS, respectively, this result was not surprising. Perhaps more surprising however was the simultaneous identification of proteins involved in pathways less commonly associated with ROS management: namely glycolysis, carbohydrate metabolism, and amino acid biosynthesis, among others. Widespread upregulation/increased oxidation of metabolic enzymes such as O-acetylserine sulfhydrylase and malate dehydrogenase suggest that existing front-line redox mechanisms may be insufficient to combat ROS exposure, and

that metabolic reallocation may be required to support the synthesis of additional redox compounds to maintain cellular state.

An additional conclusion from these experiments stems from the identification of proteins associated with the H₂O₂ and BSO tissue treatments: of the proteins from the 52 single protein spots, not one was identified in more than one combination of treatments (H₂O₂/BSO) and detection methods (SYPRO/IAF). This result indicates that, in *B. juncea* at least, ROS initiates different redox-response mechanisms depending on whether the source is endogenous or exogenous. The lack of overlap between SYPRO- and IAF-significant proteins further indicates a disconnect between protein expression changes and redox state: for the proteins identified, a change in redox-regulation or a change in expression - never both - may alter flux through their respective redox response pathways.

Among the proteins identified in this series of experiments, several - including 3-phosphoshikimate 1-carboxyvinyltransferase (a component of the shikimate pathway leading to tyrosine, tryptophan, and phenylalanine biosynthesis) and 5-methyltetrahydropteroyltri-glutamate-homocysteine S-methyltransferase (from the methionine biosynthesis pathway), have not been previously identified as containing redox-sensitive thiol groups. Due to the nature of the dual labeling strategy, it is not possible to determine from the existing data whether these proteins contain reversible disulfides or are modified by glutathione, etc. As these proteins may represent regulatory control points in their respective pathways, it would be beneficial to identify possible redox-regulatory mechanisms through the use of a more targeted functional studies.

One of the goals of this set of experiments was to develop a methodology that could be used to probe the thiol-redox proteome in a variety of plant species and tissue types under varying conditions. Based on the results as previously described, that goal has been met. The next step, application to different plants under other oxidative stress conditions has already been initiated, and the first round of results from those experiments are described in Chapter 4.

Connecting Proteomes and Climate Change: Ozone-Induced Changes in the Total and Redox Proteomes of Glycine max (Soybean)

Once a suitable thiol-labeling strategy (described in Chapter 3), had been developed, the next step to employ it with a more agriculturally relevant and redox-sensitive crop, such as soybean. To do this, we established a collaborative effort with the laboratory of Lisa Ainsworth in the USDA-ARS group at the University of Illinois-Urbana-Champaign. The Soybean Free Air Concentration Enrichment (SoyFACE) facility at UI-UC uses a ring-based ozone exposure setup to maintain concentrations over crops by way of a computer controlled system. Because the soybean plants are sown directly in the soil, problems obtaining sufficient light and water while being exposed are avoided, and as an added benefit the plants are able to grow in a natural, exposure-realistic environment.

As described in Chapter 4, “From crops to climate change: redox proteomics of soybean ozone responses,” soybean plants were field grown under three different ozone concentrations using FACE technology, and differences in expression and thiol-content

assessed via 2D-SDS-PAGE and LC-MS/MS. To confirm that the observed expression changes translated into appreciable changes in protein activity, activity assays were also conducted for a number of implicated proteins.

Although previous studies have investigated the effects of acute ozone on the soybean proteome, none have reported on the consequences of chronic exposure over a growing season. Moreover, this work is the first to use plants grown in the field with relevant day-night ozone exposure cycles versus growth chamber experiments in which plants are constantly exposed to high ozone levels for a few days. This work indicates that a number of proteins, particularly those involved in primary metabolism, redox homeostasis, and amino acid metabolism, experience large shifts in their expression and redox-state. While a subset of these proteins have been previously identified in other ozone experiments, many represent new additions to the redox-responsive proteome in plants. In this section, I will focus on the protein classifications in which some of the largest and/or most abundant fold changes in expression and redox-state took place. I will also discuss observations of both redox-regulation and redox-response.

i. Ozone induces changes in primary metabolism and amino acid biosynthesis

At concentrations greater than 40 ppb, tropospheric ozone begins to negatively affect soybean growth and yield. Above ground, these changes visibly manifest themselves as decreased shoot and pod biomass, fewer pods produced, and premature leaf senescence. Within the leaves themselves, ozone entering through the stomata is converted to ROS, which irreversibly oxidizes the plasma membrane and photosystem

components, and results in the degradation of chlorophyll. Although no comparable work has been done with soybeans, acute exposure of clover to 75 ppb ozone resulted in slower root tip formation and elongation [Vollsnes et al., 2010]. Likewise, in potatoes, chronic exposure to 80-120 ppb O₃ resulted in necrosis and vascular damage above ground, and decreased tuber size and yield below ground [Asensi-Fabado et al., 2010].

Given that the primary site of plant O₃ exposure is the leaves, it follows that many of the largest fold changes in the proteome occur there. Typically, in response to oxidative damage, photosynthetic output and Calvin cycle activity are downregulated. To compensate, plants draw on reserve energy stores, resulting in a decrease in leaf starch concentrations and a corresponding increase in the sucrose concentration [Ahsan et al., 2010]. Accordingly, enzymes involved in sugar catabolism must also be upregulated; previous studies have identified several primary metabolism proteins - among them malate dehydrogenase and phosphoglycerate mutase - that are increasingly expressed under ozone stress [Bohler et al., 2007]. My results are consistent with this view; in leaves exposed to 115 ppb O₃, there is upregulation and/or increased oxidation of multiple primary metabolism enzymes including malate dehydrogenase, phosphoribulokinase (PRK), and glyceraldehyde-3-phosphate dehydrogenase (G3PDH). The latter two enzymes are thought to form a complex that is redox-regulated by thioredoxin in response to changes in light intensity [Howard et al., 2008]. Although ozone-produced ROS damage cellular light-harvesting capacity, they also lend themselves to an increasingly oxidizing environment. In this work, PRK and G3PDH are both approximately 4-fold more oxidized in the 115 ppb-treated tissue relative to the

control. Likewise, activity assays indicate that G3PDH is approximately 3.7-fold more active in the 115 ppb-treated tissue relative to the control. Like PRK and G3PDH, malate dehydrogenase (MD) is an important regulatory control point, serving as the rate-limiting step of the TCA cycle. Previous studies have shown that cytosolic MD is redox-regulated, with reversible inactivation occurring under oxidizing conditions [Hara et al., 2006]. In this study, MD was detected in a number of different tissue-O₃ concentration combinations, with the highest upregulation (2-fold) and oxidation (5-fold) occurring in leaves exposed to 115 ppb ozone. Though oxidation is thought to decrease MD activity, this study sees greater activity in leaves exposed to 115 ppb ozone as compared to leaves exposed to ambient conditions. This is likely due to the fact that, even under oxidizing conditions, reduced protein makes up a much higher percentage of the total protein than oxidized protein, and thus a 2-fold change in expression results in a proportionally larger percentage of reduced - and in this case active - protein being available.

As expression of key proteins increases in response to oxidative stress, the demand for amino acids used in protein synthesis also increases. In leaves exposed to 115 ppb ozone, the upregulation and/or increased oxidation of a number of proteins involved in amino acid biosynthesis, including aspartate-semialdehyde dehydrogenase (ASADH), glutamine synthetase (GS), and phosphoglycerate kinase (PK), was observed.

The first enzyme, ASADH, is of particular importance in plants, as it represents the primary control point in the biosynthetic pathways responsible for the production of isoleucine, methionine, lysine, and threonine. Experiments conducted in *E. coli* suggest that one of the enzyme's catalytic cysteine residues can be reversibly reduced to control

enzyme activity; however no comparable redox analysis has been conducted using a plant variant of the enzyme [Alvarez et al., 2003]. This study is the first to identify the soybean enzyme as a possible redox-regulation target, and as a component of the plant oxidative-response mechanism.

Like ASADH, GS represents a major metabolic control point in plants. The enzyme, along with its partner enzyme glutamate synthetase, is part of the GS-GOGAT cycle and is necessary for assimilation of ammonium in plants. GS itself catalyzes the ATP-dependent fixation of ammonium to glutamate to form glutamine, while glutamate synthetase catalyzes the synthesis of two molecules of glutamate from one molecule of glutamine and one molecule of 2-oxoglutarate [Bernard and Habash, 2009]. The active sites of both cytosolic and plastidic isoforms of GS are known to contain one or more reaction-critical cysteine residues; redox modification of these residues presents a likely means for attenuating GS enzyme activity [Bernard and Habash, 2009; Choi et al., 1999].

In the 2D-gel analysis, PK was observed to be upregulated and more oxidized in response to oxidative stress. Although this enzyme is a key component in the serine biosynthesis pathway (which leads to production of the redox-critical amino acid cysteine), I was unable to detect a difference in enzyme activity between the control and 115 ppb ozone-treated extracts. This is likely due to the fact that the isoforms of the enzyme were all identified in spots containing several other proteins, including fructose-bisphosphate aldolase (FBA). Since activity assays confirmed a difference in FBA activity between the control and treated extracts, it is likely that this (and other enzymes) account for the activity in those spots and that PK is a false positive.

ii. Ozone affects redox homeostasis and induces stress response proteins

Given that tropospheric ozone is a source of ROS, it is not surprising a number of proteins involved in maintaining cellular redox homeostasis were differentially regulated and/or oxidized in response to ozone exposure. Across several different tissue-type/ozone concentration combinations, multiple different isoforms of glutathione-s-transferase (GST) were detected. In plants, GSTs comprise a large family of enzymes that, as their name suggests, conjugate glutathione to a variety of substrates to aid in xenobiotic detoxification or to serve a redox-protective function (Dixon et al., 1998). In our study different isoforms were both up or down regulated, more oxidized or more reduced in the treated tissues relative to the controls, but this is to be expected given the diversity of reactions that GSTs catalyze. In the roots of soybean plants treated with 60 ppb ozone, I detected an uncharacterized isoform of thioredoxin that was upregulated relative to the control tissue (Table 2d). By homology (BLAST), this thioredoxin is likely an m-type, meaning that it is localized to the chloroplasts. Given that the ROS generated from ozone degradation arguably have their greatest effect against redox-sensitive chloroplast components, the upregulation of such a thioredoxin is logical. This thioredoxin was not observed as being differentially regulated/expressed in the roots of plants exposed to 115 ppb ozone, nor was it observed in leaf tissue, suggesting that its expression may be tied to the degree of oxidative damage experienced by components of the “dark reactions” of carbon fixation.

Methionine sulfoxide reductase (MSR) also appears to be an important component of the soybean ozone response and was observed to be 2.7-fold more oxidized

in this study following exposure to 115 ppb ozone. This enzyme is responsible for recycling the limiting amino acid methionine through the reduction of methionine sulfoxide. The reaction mechanism for the bacterial enzyme proceeds through the formation of an intramolecular, thioredoxin-mediated, disulfide bond [Antoine et al., 2003]. Although it follows that the enzyme would be more oxidized in an ozone-rich environment, additional work is needed with the plant enzyme to confirm that the same reaction mechanism applies.

Like MSR, the enzyme ascorbate peroxidase (AP) also plays an important redox-protective/corrective role. Ascorbate, as described in the introduction, is one of the three major redox couples found in plants; AP, as part of the larger ascorbate-glutathione cycle, utilizes ascorbate as a substrate in order to detoxify hydrogen peroxide and other peroxides produced from oxidative bursts [Noctor and Foyer, 1998]. In this study, total AP (isoform 2) expression was up-regulated approximately 2.5-fold in leaves exposed to 115 ppb ozone, while AP (isoform 1) was down-regulated 1.5-fold in 115 ppb-exposed roots. The cause of these differing expression profiles is unclear, though it may relate to the redistribution of resources from more moderately oxidized (roots) to more severely oxidized (leaves) tissue types.

iii. Ozone induces both a redox response and changes in redox-regulation

Because ozone, as a denser-than-air gas, is ground-hugging but not ground-penetrating, it follows that the largest effects of high tropospheric ozone concentrations would be found in aerial plant tissues. In this study, the largest fold-changes in protein

expression and redox state were found in leaves exposed to 115 ppb ozone, and overall more than half of the significant proteins were detected in leaf tissue. However, a sizable minority of proteins were identified exclusively in root tissue. Given that this tissue was not exposed to ozone, several hypotheses exist concerning why local changes in expression and redox state were observed. One possibility is that ozone-induced ROS act as a propagated signaling molecules that directly prompt changes in protein redox state and regulation. Ozone, besides its capacity for degradation into multiple ROS, is also known to prompt the rapid release of H₂O₂ into the cell apoplast as a result of even short term exposure [Rao and Davis, 2001]. H₂O₂ is able to diffuse into cells through oxidative-gated aquaporins, and can function as an intercellular signal for activating the hypersensitive response and other plant defense mechanisms [Henzler et al., 2004; Henzler and Steudle, 2000]. In this situation, ROS can not only directly regulate the activity of anti-oxidation enzymes, but can also instigate a secondary redox response through the activation of transcription factors and other protein expression machinery. However, given their high reactivity, it is unlikely that most ROS would be able to diffuse more than one or two cells from their site of origin before being consumed [Murphy et al., 2001]. Although it is theoretically possible that ROS could serve as a long-distance signaling molecule by propagating from cell at a time, the majority of evidence indicates that it is confined to localized intracellular responses.

Another hypothesis for explaining the observed root response to ozone is that another small molecule besides the various ROS is responsible for signal transduction. Out of the other numerous hormones and peptides known to play a role the plant defense

response, calcium is the most likely candidate. Besides the advantage it gains by being easily diffused, calcium is also normally maintained in cells for controlling channel flux, and has previously been implicated as a secondary messenger in several other signal transduction networks. Prior work in *Arabidopsis* has indicated that roots and shoots independently undergo large changes in calcium flux as a result of ozone exposure; however, it is unclear how much of that flux crosses the shoot/root boundary [Evans et al., 2005]. Assuming that signal transduction does at least in part proceed from the exposed shoots to the roots, then the changes in expression we see in soybean root tissue are likely a secondary redox response, with changes in protein thiol status brought on by secondary local calcium-induced oxidative bursts.

A third possibility for explaining the observed root response to ozone, is that the roots themselves are sensing the small amount of ozone able to diffuse from the surface, and are responding independently of the shoot tissue. Previous studies have confirmed that exposed roots can mount their own response to ROS independent of the aerial tissue if the two portions are physically separated [Rentel and Knight, 2004]. If this is the case, then our observed changes expression and thiol status may again be a result of primary redox regulation of local proteins, with secondary redox responses occurring beyond the sites of exposure. In all actuality though, it is likely that a combination of all three methods of signal propagation are occurring in soybeans as a result of ozone exposure; further local real-time analysis may be required to fully tease out the differing degrees of regulation and response observed, particularly in the below-ground tissues.

iv. Future Work

In chapter 4, 159 proteins were identified that differentially respond and/or are regulated by changes in tropospheric ozone concentration. Seventy-nine of these proteins saw significant changes in their expression and/or oxidation state following exposure to ozone at a concentration of 115 ppb. Many of these proteins (Figure 4 of Chapter 4) represent critical control points for the maintenance of carbon metabolism. The identities and distribution of proteins identified in soybean plants exposed to 115 ppb ozone, as compared to plants exposed to a lower ozone concentration, suggests that the observed metabolic changes may result from acclimation of the plant to chronic ozone exposure. Although altered total and redox proteomes were examined in soybean, it is unclear if these changes correspond to changes in carbon flux and sugar mobilization. That is to say, are more metabolites being pushed through the system, and from where is the carbon being obtained?

To examine the effect of chronic ozone exposure on soybean metabolism it would be beneficial if subsequent analyses of the soybean ozone-redox proteome were carried out in conjunction with analyses of the soybean metabolome. Using standard GC- and HPLC-based methodologies, it would be possible to compare the relative pool sizes of key metabolites (glycolysis, TCA cycle, and amino acid precursors, etc.) from ozone-exposed and ambient-exposed soybean tissues. To quantify changes in storage metabolites, fatty acids from crude soybean extract could be converted to fatty acid methyl esters (so as to make them more polar and volatile), and separated by GC [www.gerstel.com]. Metabolite identity and quantity could then be determined via

comparison to the retention time and known make-up of a standard mixture; identity could also be determined via GS-MS [Lehmann et al., 2009]. For other primary and secondary metabolites (including amino acids), both polar and non-polar extractions from the initial tissue would be carried out. Metabolites in the polar phase would be separated by HILIC (Hydrophobic Interaction Liquid Chromatography) so as to improve their volatility for downstream electrospray ionization, while metabolites in the non-polar phase would be separated by traditional reverse phase chromatography [Grumbach et al., 2004]. As above, metabolite identity and quantity could be determined in both cases using a standard injection mixture; identifications could also be confirmed via LC-MS. The resulting data from this series of experiments would indicate whether observed upregulation of protein expression corresponded to changes in carbon metabolism, and at what stage of metabolism reallocation of resources was occurring to fuel *de novo* protein synthesis.

To complement the above approach, a C¹⁴-based feeding study could also be carried out. In short, leaf disks from ambient-exposed and high ozone-exposed soybeans would be soaked in buffer containing one of several C¹⁴-labeled metabolites. At pre-determined time-points, the leaf disks would then be flash frozen, and leaf metabolites extracted and fractionated by TLC. The radioactive plates would then be imaged using x-ray film, and radioactive counts for individual spots would be determined by liquid scintillation [Katahira and Ashihara, 2009]. Alternately, C¹³ could be used in place of C¹⁴ in conjunction with fractionation and separation via GS-MS to improve both metabolite separation and overall sensitivity [Feng et al., 2010, Tang et al., 2010]. Because C¹³

occurs naturally at approximately 1 out of every 100 possible carbon atoms, care must be taken to avoid confusing background C13 with that resulting from labeling. To obtain absolute quantification, GC metabolite profiles for both ambient- and high ozone-treated tissues could be compared against a standard injection containing known quantities of metabolites of interest. By repeating this experiment for different lengths of time and with different starting metabolites (in particular different sugars, starches, and photosynthetic precursors), it should be possible to determine not only which metabolic pools are changing, but also from where the carbon necessary for observed metabolite and protein changes is being sourced.

For the work described in Chapter 4, tissue from soybean exposed to three different ozone concentrations (115, 60, and 40 ppb) was harvested at the R3 stage of development for further analysis; however, acclimation to chronic ozone exposure is likely a time-dependent process and further analysis should involve investigation of changes at various soybean developmental stages across the growing season. While the initial experiments (Chapter 4) provide new insights into ozone-induced changes to the soybean proteome, in particular the existence of a response threshold between 60 and 115 ppb, the results only reflect a snapshot along a continuum of time- and dose-dependent responses. By the R3 stage of development, the expression and activity of a number of metabolomic proteins has been altered, while other ozone-damage marker proteins (in particular RuBisCO and PEPC) remain as yet unchanged. However, it is unclear how early during soybean growth and exposure the observed changes are manifested, and whether or not the changes remain consistent as the growing season progresses. By

harvesting and analyzing tissue from additional timepoints, a more complete picture of the soybean temporal redox response, including changes in the specific activities of key proteins, could be established. As with the additional timepoints, the collection and analysis of tissues exposed to alternate ozone concentrations, in particular those above 115 ppb and those between 60 and 115 ppb, would provide new and valuable insights. While the existing data implies the presence of a ozone response threshold between 60 and 115 ppb, it is unclear at what ozone concentration the transition occurs. Better knowledge of the threshold for changes in the ozone proteomic response may allow future farmers to enact preventative measures as per predictions for daytime and seasonal tropospheric ozone concentrations.

Analysis of tissues exposed to ozone concentrations above 115 ppb would allow for better understanding of how ozone affects the proteome under future-predicted conditions. Growth chamber studies examining continuous or acute exposure to have indicated that acute exposure to ozone concentrations in excess of 100 ppb results programmed cell death [Pell et al., 1997; Chen et al., 2009; Overmyer et al., 2003]. A better understanding of physiological effects resulting from the natural diurnal ozone cycle and the presence of variable weather conditions may allow field grown soybean to better mitigate the toxic effects of ozone. Moreover, proteomic and metabolic strategies may also help in the identification of molecular targets for engineering of ozone tolerant strains of soybean.

v. To Build a Better Soybean Plant ...

Because current soybean cultivars are not ozone tolerant, and because traditional breeding takes more time than can be afforded, biotechnology to engineer soybeans presents the most reasonable avenue for generating ozone resistant varieties, but this requires significant understanding of the molecular responses to this abiotic stress. Based on the completed experiments, it is clear that not just individual proteins, but rather whole pathways are involved in maintaining the soybean ozone response. While it would be arguably advantageous to re-engineer all of proteins of glycolysis to increase their substrate affinities under high stress conditions, it is much more so within the realm of practicality to focus on one or two proteins at a time. Because many of the proteins identified in this work are redox regulated, or may yet be identified as such, one possible avenue is to increase the availability of reducing equivalents to help redox-regulated proteins maintain an active redox state during exposure to ROS.

The majority of proteins identified as differentially oxidized following ozone exposure are known targets of thioredoxin or glutaredoxin; thus the “doxins” may represent a starting point for manipulating the soybean ozone response. Unfortunately, both thioredoxins and glutaredoxins exist as protein families within a given species (for example, *A. thaliana* has at least 19 and 22 different thioredoxins and glutaredoxins, respectively), making the choice of one or two to alter difficult in the absence of detailed interaction data [Buchanan and Balmer, 2005]. Because both glutaredoxins and thioredoxins ultimately derive their reducing power from NADPH, another option for increasing the availability of reducing equivalents is to target chloroplast ferredoxin.

Like the other doxins, ferredoxin is a member of a large protein family; however, plant species tend to contain relatively few ferredoxin-encoding genes (*A. thaliana* has 4), a subset of which encode chloroplast-specific ferredoxins [Hanke et al., 2004; Fukuyama, 2004]. Although NADPH production is ultimately tied to light intensity, the presence of additional ferredoxin activity may aid in the scavenging of electrons that would otherwise be lost from photosystem I under high light, and additionally help prevent the formation of additional ROS.

As an alternative to providing cells with additional reducing equivalents to combat ozone stress, it may be possible to engineer plants to better neutralize ROS at their point of entry - prior to the induction of damaging oxidative cascades. Ozone enters a plant primarily via the stomata, and in the apoplastic space it is rapidly converted to various toxic ROS. Prior research in broad bean has indicated that ascorbate and dehydroascorbate concentrations rise rapidly following ozone exposure [Luwe and Heber, 1995; Kangasjarvi et al., 2005]. As ascorbate is one of the three major redox couples employed by plants, this observation suggests an effort by the plants to quickly stave off the creation of damaging ROS. In line with this observation, we also noted the upregulation of ascorbate peroxidase - an enzyme that utilizes ascorbate to detoxify peroxides - in leaves exposed to 115 ppb ozone. While ascorbate has been implicated as a major player in apoplasmic detoxification and remains a focus of ongoing studies [Burkey et al., 2003; Ainsworth et al., 2008], other redox reactive molecules may provide more subtle contributions. Many other compounds, including flavonols, flavones, isoflavonoids, and anthocyanins, also display ROS scavenging capabilities, and have

been found to be localized in part to the apoplast. For example, the presence of the flavonol quercetin has been noted in the oxidized outer scales of brown onion, and the isoflavonoids daidzein and sojagol have been identified in ozone-stressed soybean leaves [Takahama and Hirota, 2000; Keen and Taylor, 1975]. Despite clear benefits associated with many of these compounds in response to oxidative damage, work in the area has been slow, due in part to inherent metabolite and metabolite-derivative toxicity at higher concentrations [Didyk and Blum, 2006; Bais et al., 2003; Parvez et al., 2004].

In the work described in Chapter 4, we noted that expression of two isoforms of chalcone isomerase were upregulated relative to the control in leaves exposed to 115 ppb ozone. This enzyme presents an interesting target for improving plant ozone tolerance in that it is directly upstream of the enzymes responsible for isoflavone, flavonone, and anthocyanin biosynthesis, among other ROS scavenging metabolites. While based on our work and available crystal structures chalcone isomerase does not appear to be redox regulated at the protein level, it is clear that the enzyme responds in some capacity to oxidative cues [Ferrer et al., 2008]. Limited investigations in duckweed (*Lenma gibba*) have indicated that chalcone isomerase transcript accumulation is tied to the inhibition of chloroplastic electron transport, though it remains unclear whether other signaling factors are involved [Akhtar et al., 2010]. Accordingly, further investigation of the chalcone isomerase promoter and its redox responsiveness is warranted. Depending on the motifs and modes of regulation found during study of the chalcone isomerase promoter, it may be possible to adjust expression of the enzyme so that its production is more tightly linked to increases in atmospheric ozone concentration. In that eventuality, the effect of

greater chalcone isomerase availability on downstream redox metabolite accumulation, and in turn their ability to disarm invading ozone-based ROS, would also be prime targets for further investigation.

An additional strategy could be to target the enzymes of carbon metabolism in an effort to maintain core metabolism at levels that maintain sufficient plant growth and seed yield. As another target for improving soybean ozone tolerance, RuBisCO at first glance appears to be an excellent candidate. The enzyme, as the lynchpin of the Calvin cycle, is responsible for the carboxylation of ribulose-1,5-bisphosphate, and accounts for 30-50% of the soluble protein present in leaves [Feller et al., 2008]. While higher soluble RuBisCO concentrations are unlikely to be achievable due to limiting nitrogen, common sense would dictate that by increasing the enzyme's catalytic efficiency, it should be possible to increase overall carbon flux [Parry et al., 2003]. Extensive efforts have been undertaken to increase CO₂ specificity via directed evolution, as well as through the formation of hybrid cross-species holozymes; however, it appears that CO₂ specificity is closely tied to O₂ specificity (RuBisCO also displays alternate oxygenase activity), and thus no significant improvements have been made [Parry et al., 2003; Mueller-Cajar and Whitney, 2008].

In light of these difficulties, efforts to improve carbon flux have also been directed at key RuBisCO-interactor: RuBisCO activase (RA). RA catalyzes the removal of the inhibitor carboxyarabinitol 1-phosphate (CA1P) from the RuBisCO active site in the absence of a key carbamylation modification required for catalysis. As a regulator of RuBisCO activity, RA is in turn regulated via several distinct mechanisms. Because the

enzyme displays ATPase activity (as ATP hydrolysis is required for CA1P removal), it is sensitive to the ratio of ADP:ATP present in the chloroplast. At a ratio of 1:1, common during the nighttime hours, RA activity is minimal. As the light reactions of photosynthesis become active however, this ratio increases to 1:2 or 1:3, and RA activity increases accordingly [Zhang and Portis, 1999; Kallis et al., 2000]. In addition to ADP:ATP ratio, some isoforms of RA are also regulated by temperature and the formation of a reversible disulfide bond. In plants, some species contain a gene encoding only a single, short form of RA, while others produce an additional long form splice variant or express a longer isoform from a distinct gene [Portis, 2003]. While the larger isoform appears more sensitive to the ADP:ATP ratio, it is also generally more thermo-stable as well [Shen et al. 1991]. The larger isoform additionally is regulated through the formation of a thioredoxin-mediated disulfide bond at the C-terminus of the protein [Zhang and Portis, 1999]. The isoforms appear to be co-expressed, and there is limited evidence to suggest that redox regulation of the larger isoform can effect the activity of the smaller isoform, though the extent of their interaction remains unclear [Zhang et al., 2001].

In this study, two different isoforms of soybean RA, (one of which was differentially oxidized), were identified in leaves treated with 115 ppb ozone. Because the large isoform is already redox regulated, it presents an excellent target for improving carbon flux and oxidation tolerance in crop plants. In *Arabidopsis*, directed evolution of the small isoform has resulted in several variants that display increased thermo-tolerance and secondarily improve CO₂ assimilation rates and seed yield [Kurek et al., 2007].

Since tropospheric ozone concentrations are tied to temperature, improvements to the thermo-tolerance alone of the soybean large RA isoform could confer some benefit.

Using additional directed evolution-based strategies, such as the blending of isoforms from various species, it may be possible to generate an RA variant that improves soybean ozone tolerance and helps maintain seed yield.

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APPENDIX I

STRUCTURAL AND FUNCTIONAL EVOLUTION OF ISOPROPYLMALATE

DEHYDROGENASES IN LEUCINE AND GLUCOSINOLATE

PATHWAYS OF *ARABIDOPSIS THALIANA*

Classification: Biological Sciences - Plant Biology, Biochemistry

**Structural and Functional Evolution of Isopropylmalate Dehydrogenases in
Leucine and Glucosinolate Pathways of *Arabidopsis thaliana***

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Running Title: Structure and function of Arabidopsis isopropylmalate dehydrogenases

Keywords: crystal structure, functional evolution, isopropylmalate dehydrogenases, leucine
biosynthesis, glucosinolate biosynthesis, plant

Data deposition: The atomic coordinates and structure factors for AtIPMDH2 have been
deposited in the Protein Data Bank, www.rcsb.org (PDB ID code 3R8W).

Author contributions: JMJ and SC designed research; YH, AG, QY, JMS and SB performed
research; YH, AG, JMJ and SC analyzed data; and YH, JMJ and SC wrote paper.

The authors declare no conflict of interest.

1 **Abstract**

2 The methionine chain-elongation pathway required for aliphatic glucosinolate biosynthesis in
3 plants is thought to have evolved from leucine biosynthesis. In *Arabidopsis thaliana*, three 3-
4 isopropylmalate dehydrogenases (AtIPMDHs) play key roles in either methionine chain-
5 elongation for aliphatic glucosinolate biosynthesis (AtIPMDH1) or in leucine synthesis
6 (AtIPMDH2 and AtIPMDH3). Here we elucidate the molecular basis underlying the metabolic
7 specialization of these enzymes. The crystal structure of AtIPMDH2 was solved to provide the
8 first detailed molecular architecture of a plant IPMDH. Modeling of 3-isopropylmalate binding in
9 active site of the crystal structure and sequence comparisons of prokaryotic and eukaryotic
10 IPMDH suggest that substitution of one active site residue in AtIPMDH1 may lead to altered
11 metabolic function. Site-directed mutagenesis of Phe137 to a leucine in AtIPMDH1 (AtIPMDH1-
12 F137L) reduced the enzyme activity toward 3-(2'-methylthio)ethylmalate by 200-fold, but
13 enhanced catalytic efficiency with 3-isopropylmalate to levels observed with AtIPMDH2 and
14 AtIPMDH3. Conversely, the AtIPMDH2-L134F and AtIPMDH3-L133F mutants enhanced
15 catalytic efficiency with 3-(2'-methylthio)ethylmalate ~100-fold and reduced activity for 3-
16 isopropylmalate. Furthermore, the altered *in vivo* glucosinolate profile of an *Arabidopsis ipmdh1*
17 T-DNA knockout mutant could be restored to wild-type levels by constructs expressing
18 AtIPMDH1, AtIPMDH2-L134F, or AtIPMDH3-L133F, but not by AtIPMDH1-F137L. These results
19 demonstrate that a single amino acid substitution results in functional divergence of IPMDH *in*
20 *planta* to affect substrate specificity and contribute to the evolution of specialized glucosinolate
21 biosynthesis from ancestral leucine biosynthesis.

1 \body

2 **Introduction**

3 To compensate for their sessile nature, plants evolved mechanisms to cope with rapid
4 environmental changes and challenges (1). The production of specialized metabolites is one of
5 the important mechanisms for the survival and fitness of plants (2). The molecular diversity of
6 these specialized compounds arises from differential modification of common backbone
7 structures, which necessitates the evolution of homologous enzymes with varied specificities
8 (1). In plants, glucosinolates constitute a diverse group of sulfur-containing specialized
9 metabolites (3-4). Biosynthesis of methionine-derived glucosinolates is initiated by the
10 sequential addition of methylene groups to produce chain-elongated methionine derivatives via
11 an iterative three-step chain-elongation process that mimics the chemistry of leucine synthesis
12 (Fig. 1A).

13

14 To date, all the genes involved in the methionine chain-elongation process have been
15 identified and characterized in *Arabidopsis thaliana* (5-14). The different enzymes of the
16 methionine chain-elongation pathway for glucosinolate synthesis appear to have evolved from
17 leucine synthesis either by gene duplication and neo-functionalization of one of the duplicated
18 genes or by sub-functionalization via differential temporal and spatial expression of gene copies
19 (14-15). For example, four genes in *Arabidopsis* encode isopropylmalate synthases (IPMS) with
20 two (IPMS1 and IPMS2) serving in leucine biosynthesis and the other two genes encoding
21 methylthioalkylmalate (MAM) synthases (MAM1 and MAM3) catalyzing the committed step in
22 methionine chain-elongation (5-6, 16). A recent study showed that loss of a C-terminal
23 regulatory domain and a few amino acid exchanges can convert IPMS into MAM (14).
24 Specialization of the *Arabidopsis* isopropylmalate isomerases (IPMI) for different metabolisms
25 occurs by changes in the oligomeric composition of these enzymes. IPMI are heterodimeric

1 enzymes consisting of a large subunit encoded by a single gene and a small subunit encoded
2 by one of three genes (8-9, 12). Metabolic profiling of the large subunit mutant revealed
3 accumulation of intermediates in both the leucine pathway and the methionine chain-elongation
4 pathway, demonstrating the dual function of this subunit in both leucine and glucosinolate
5 biosynthesis (10). In contrast, the small subunits are specialized to either leucine biosynthesis
6 or methionine chain-elongation (2, 10, 12). Furthermore, among the six branch-chain
7 aminotransferases (BCATs) in *Arabidopsis*, BCAT4 in the cytosol is specifically involved in
8 glucosinolate biosynthesis, whereas BCAT3 in the plastids functions in both amino acid and
9 glucosinolate biosynthesis (7, 9). However, the changes that tailor BCAT activity are unclear.

10

11 Previously, we showed that *Arabidopsis thaliana* isopropylmalate dehydrogenase 1
12 (AtIPMDH1) catalyzes the oxidative decarboxylation step in the methionine chain-elongation of
13 glucosinolate biosynthesis and that AtIPMDH2 and AtIPMDH3 are primarily involved in leucine
14 biosynthesis (Fig. 1B) (11, 13). These studies highlight the functional specialization of these
15 isoforms, but do not reveal how these activities evolved. Here we examine the molecular basis
16 for the functional evolution of the IPMDH family in *Arabidopsis*. The crystal structure of
17 AtIPMDH2, the first determined for a plant IPMDH, reveals an active site structure similar to that
18 of the bacterial enzymes and provides a template for modeling substrate binding in the active
19 site. Analysis of the AtIPMDH2 structure, sequence comparisons, and site-directed mutagenesis
20 demonstrate that a single residue difference in the active site drastically alters substrate
21 specificity of the AtIPMDH isoforms both *in vitro* and *in vivo*. This work demonstrates the basis
22 for functional divergence of an AtIPMDH isoform for glucosinolate biosynthesis from those
23 involved in leucine biosynthesis.

24

25 **Results**

1 **Sub-functionalization and AtIPMDH Metabolic Specialization.** The three *IPMDH*
2 genes in Arabidopsis have overlapping, yet distinct expression patterns. *AtIPMDH1* (At5g14200)
3 is highly expressed in leaves and roots; *AtIPMDH2* (At1g80560) is weakly expressed throughout
4 the plant; and *AtIPMDH3* (At1g31180) is constitutively expressed at high levels in all tissues
5 (11, 13, 17). To test the possible contribution of differential expression to the specialization of
6 *AtIPMDHs*, each gene was placed under control of the native *AtIPMDH1* promoter and then
7 transformed into an *atipmdh1* mutant line (11). As shown in Fig. 2, the altered glucosinolate
8 profile of the *atipmdh1* mutant could only be rescued by expression of *AtIPMDH1*. The results
9 indicate that subfunctionalization may not be the cause of AtIPMDH specialization.

10

11 **Structure of AtIPMDH2.** To determine the molecular architecture of a plant IPMDH, the
12 2.25 Å resolution x-ray crystal structure of AtIPMDH2 was solved by molecular replacement
13 (Table S1). AtIPMDH2 is a dimeric protein with each monomer consisting of two domains (Fig.
14 3A). Domain 1 contains seven α -helices (α 1-4 and α 9-11) and five β -strands (β 1-3 and β 11-12),
15 along with the N- and C-termini. Four α -helices (α 5-8) and seven β -strands (β 4-10) comprise
16 domain 2. Between the two domains, β 4 and β 5 form the interdomain region. The second
17 domain also serves as the dimerization interface with β 6 and β 7 of each monomer as part of an
18 inter-subunit β -sheet and α 7 and α 8 of each monomer forming a four-helix bundle at the dimer
19 interface. The overall structure of AtIPMDH2 is similar to those of the IPMDH from various
20 bacteria, including *Salmonella typhimurium* and *Thermus thermophilus* (18-20), with a root
21 mean square deviation of 1.3-1.7 Å² over ~350 residues. Because the plant and bacterial
22 IPMDH share ~50% sequence identity, conservation of key residues defines the active site
23 region situated in a cleft between the two domains of each monomer (Fig. 3A).

24

25 The active site (Fig. 3B) is roughly delineated by α 8 at the bottom and with α 4 of one
26 monomer and α 7 of the adjacent monomer forming opposite sides of the site. Within the active

1 site, all of the residues previously identified in structures of bacterial IPMDH in complex with
2 isopropylmalate and Mg^{2+} are also conserved in AtIPMDH2 (18, 20). Because efforts to obtain a
3 structure of AtIPMDH2 in complex with ligands did not yield crystals, 3-isopropylmalate and
4 Mg^{2+} were modeled into the plant enzyme based on the positions of these ligands observed in
5 the bacterial structures (Fig. 3B and C) (18, 20). This comparison shows that Asp264* (asterisk
6 denotes adjacent monomer), Asp288, and Asp292 are positioned to interact with a catalytically
7 essential divalent metal (i.e., Mg^{2+} or Mn^{2+}) and that a trio of arginines (Arg136, Arg146, and
8 Arg174) is poised to form charge-charge interactions with the carboxylate groups of the
9 substrate. Residues corresponding to Leu132, Leu133, Tyr181, Lys232*, Asn234*, and Val235*
10 form a largely hydrophobic region around the isopropyl group of the substrate.

11

12 Although all of these amino acids are invariant in the bacterial and plant IPMDH involved
13 in leucine biosynthesis, the side-chain corresponding to Leu133 is replaced with a
14 phenylalanine in AtIPMDH1 (Fig. 3D), which is the isoform previously shown to be primarily
15 involved in glucosinolate synthesis in Arabidopsis (11). Mechanistically, the conversion of 3-
16 isopropylmalate to 4-methyl-2-oxovalerate in leucine synthesis and the conversion of 3-malate
17 derivatives (e.g., 3-(2'-methylthio)ethylmalate) to 2-oxo acids (e.g., 5-methylthio-2-oxopentanoate)
18 in glucosinolate synthesis likely use a common metal-dependent reaction (Fig. S1); however,
19 different substrate side-chains of 3-malate derivatives (Fig. 1B) must fit in the plant IPMDH
20 active site for production of aliphatic glucosinolates with six different chain lengths (C3-C8).
21 Thus, we hypothesize that this single amino acid exchange from the leucine found in AtIPMDH2
22 and AtIPMDH3 to the phenylalanine in the active site of AtIPMDH1 may contribute to the
23 functional divergence of this isoform for glucosinolate biosynthesis.

24

25 **Biochemical Analysis of Wild-Type and Mutant AtIPMDH.** Previous studies on the
26 AtIPMDH demonstrate that each isoform accepts 3-isopropylmalate as a substrate (11, 13), but

1 a kinetic comparison with a glucosinolate pathway substrate has not been reported. Using both
2 3-isopropylmalate and 3-(2'-methylthio)ethylmalate, the steady-state kinetic parameters for each
3 AtIPMDH were determined (Table 1 and Fig. 4). Comparison of the catalytic efficiencies shows
4 that AtIPMDH2 and AtIPMDH3 favor 3-isopropylmalate over 3-(2'-methylthio)ethylmalate by
5 14,900- and 29,600-fold, respectively. Moreover, these isoforms were ~20-fold more active with
6 the leucine biosynthesis substrate than AtIPMDH1. In comparison, AtIPMDH1 accepts both
7 substrates with comparable k_{cat}/K_m values, but was ~500-fold more efficient with the
8 glucosinolate substrate than the other two isoforms. These catalytic efficiencies agree with the
9 observed *in vivo* roles of the AtIPMDH isoforms in glucosinolate and leucine synthesis pathways
10 (11-13).

11

12 To investigate the significance of the active site difference in the AtIPMDH, a series of
13 point mutants (AtIPMDH1-F137L, AtIPMDH2-L133F, and AtIPMDH3-L134F) were generated.
14 Kinetic analysis of these mutants demonstrates the critical role of this active site change in
15 determining substrate specificity (Table 1 and Fig. 4). In AtIPMDH1, substitution of Phe137 with
16 a leucine reduced the k_{cat}/K_m of the mutant for 3-(2'-methylthio)ethylmalate to values
17 comparable to those observed for AtIPMDH2 and AtIPMDH3. This was also accompanied by
18 improved catalytic efficiency with 3-isopropylmalate, as the AtIPMDH1-F137L mutant was only
19 2- to 3-fold less efficient with this substrate than AtIPMDH2 and AtIPMDH3. The complementary
20 mutation in either AtIPMDH2 (L133F) or AtIPMDH3 (L134F) yields mutant enzymes that were
21 ~30-fold less active with 3-isopropylmalate than the corresponding wild-type proteins, but still
22 comparable to wild-type AtIPMDH1. Moreover, AtIPMDH2-L133F and AtIPMDH3-L134F
23 displayed nearly a 100-fold improvement in activity with 3-(2'-methylthio)ethylmalate as a
24 substrate to k_{cat}/K_m values that were 4- and 7-fold less than those observed with AtIPMDH1.
25 These results demonstrate the critical role of the residue at position 133 (AtIPMDH2 numbering)

1 in the evolution of AtIPMDH1 for the methionine chain-elongation reactions of glucosinolate
2 biosynthesis.

3

4 ***In vivo* Analysis of AtIPMDH Mutant Function.** To test whether the amino acid
5 substitution that occurred in AtIPMDH1 contributes to its specific function *in vivo*, *atipmdh1*
6 mutant plants were transformed with each of the mutant *AtIPMDH* genes driven by the
7 *AtIPMDH1* promoter. After isolation of homozygous lines, the glucosinolate profile in each
8 mutant was examined. In comparison to the results shown in Fig. 1, the pronounced
9 glucosinolate phenotype in the *atipmdh1* mutant could not be rescued by AtIPMDH1-F133L
10 (Fig. 5), indicating that the active site substitution impaired AtIPMDH1 function for glucosinolate
11 synthesis *in vivo*. In contrast, the glucosinolate phenotype could be restored to the wild-type
12 profile by expression of either AtIPMDH2-L133F or AtIPMDH3-L134F (Fig. 5). The *in planta*
13 findings corroborate the conclusion drawn from the biochemical analysis of recombinant
14 proteins and provide evidence for the evolution of AtIPMDH1 by gene duplication and a single
15 critical amino acid substitution.

16

17 **Discussion**

18

19 The evolution of specialized metabolism from primary metabolism is a common theme
20 across biochemical pathways in plants (and microbes). Here we explored the molecular basis
21 underlying the divergence of biological function in the IPMDHs of Arabidopsis. Although all three
22 AtIPMDH accept 3-isopropylmalate, AtIPMDH1 is less efficient than the other isoforms (11, 13).
23 Previous work also showed that knockout mutants of *AtIPMDH1* result in reduced levels of C4
24 to C8 aliphatic glucosinolates (11). In contrast, knockout mutations of the other isoforms did not
25 alter glucosinolate levels but reduced leucine content (11, 13). Interestingly, a double mutation

1 of *AtIPMDH2* and *AtIPMDH3* in Arabidopsis plants led to defects in pollen and embryo sac
2 development, suggesting that leucine synthesis is essential for gametophyte formation. Using a
3 combination of structural and functional analysis, this work demonstrates that a single amino
4 acid change in the AtIPMDH active site leads to functional specialization of these enzymes in
5 leucine synthesis (primary metabolism) and aliphatic glucosinolate synthesis (specialized
6 metabolism).

7

8 Possible sub- and neo-functionalization processes can drive the evolution of specialized
9 metabolism (14-15). To evaluate if altered expression of *AtIPMDH* isoforms underlies functional
10 specialization, each isoform gene was expressed under control of the *AtIPMDH1* promoter in an
11 *atimpdh1* mutant background (Fig. 2). Because the glucosinolate profile in the mutant was
12 rescued only by expression of *AtIPMDH1*, it appears that neo-functionalization, involving gene
13 duplication and subsequent mutation to a new function, may be the underlying evolutionary
14 mechanism.

15

16 The three-dimensional structure of AtIPMDH2 (Fig. 3) and functional analysis (Table 1
17 and Fig. 4) of the AtIPMDH provides insight on the specific changes required to alter the
18 metabolic roles of these enzymes. A common chemical transformation is required to convert 3-
19 isopropylmalate to 4-methyl-2-oxovalerate in leucine synthesis and 3-malate derivatives to 2-
20 oxo acids in glucosinolate synthesis (Fig. 1). The AtIPMDH active site includes invariant
21 residues for binding of either Mg²⁺ or Mn²⁺ (Asp288, Asp292, Asp264*) and for charge-charge
22 interactions with the substrate carboxylate groups (Arg136, Arg146, and Arg174). Likewise,
23 Tyr181 and Lys232*, which are proposed to perform general acid-base chemistry in the reaction
24 mechanism (21), are conserved. For both 3-isopropylmalate (leucine synthesis) and 3-malate
25 derivatives (glucosinolate synthesis), the overall reaction (Fig. S1) involves oxidation of the
26 alcohol by deprotonation and hydride transfer to NAD⁺. This is followed by spontaneous

1 decarboxylation, stabilization of the resulting enolate by the metal ion, and protonation to yield
2 the final product.

3

4 Leucine and glucosinolate synthesis requires the same chemistry, but the AtIPMDH
5 active site must accommodate reactants with different side-chains (i.e., isopropyl versus
6 elongated methionine side-chain groups). The AtIPMDH2 structure and sequence analysis
7 reveals a single amino acid difference of a leucine (AtIPMDH2 and AtIPMDH3) versus a
8 phenylalanine (AtIPMDH1) in the active site. This difference occurs in the set of residues
9 proposed to form the substrate interaction surface in the bacterial and plant IPMDH (18-20).
10 Both *in vitro* and *in vivo* functional analysis of AtIPMDH1-F137L, AtIPMDH2-L133F, and
11 AtIPMDH3-L134F demonstrates that switching this amino acid in each isoform is sufficient to
12 interconvert catalytic efficiency (Table 1 and Fig. 4) and to change the aliphatic glucosinolate
13 profiles in transgenic plants (Fig. 5). These results suggest that gene duplication of AtIPMDH
14 followed by mutation of one active site residue in AtIPMDH1 leads to its specialized role for
15 glucosinolate synthesis in Arabidopsis.

16

17 The structure-function analysis of the AtIPMDH provides insight on the molecular basis
18 for altered function, but it is unclear how the leucine to phenylalanine mutation allows AtIPMDH1
19 to accommodate the growing methionine chain in subsequent iterations of the glucosinolate
20 synthesis reactions (Fig. 1A). Multiple structures of IPMDH from bacteria indicate that the
21 structural features around the active site are flexible and that active site dynamics likely plays a
22 potential role in substrate recognition and catalysis (22). Moreover, the effect of the longer side-
23 chain on the kinetics of the various glucosinolate biosynthesis pathway enzymes (i.e., BCAT,
24 MAM, IPMI, and IPMDH) has not been explored. In Arabidopsis, multiple lines of evidence
25 strongly support the evolution of methionine chain-elongation process of glucosinolate
26 biosynthesis from leucine biosynthesis (5-8, 11); however, the molecular underpinnings for this

1 evolution are only beginning to be understood. For example, the substrate specialization of the
2 heterodimeric IPMI is determined by which small subunit associates with the large subunit (2, 8,
3 10, 12). More recently, the changes needed to convert IPMS from leucine synthesis into a MAM
4 was demonstrated to involve the loss of a C-terminal regulatory domain responsible for
5 feedback inhibition by leucine and a series of amino acid mutations (14). In contrast to large
6 remodeling of protein structure in IPMS and MAM, the substrate specificity of IPMDH requires
7 one amino acid difference.

8

9 Interactions between Arabidopsis and its environment may have driven the co-evolution
10 of the pathways needed to synthesize the core glucosinolate structure and the elongation of the
11 methionine side-chain. The biosynthesis of the glucosinolates has been suggested to have
12 evolved from the prevalent system of cyanogenic glucoside biosynthesis (23-24). Evidence for
13 this includes the wide distribution of cyanogenic glucosides in plants and arthropods, and the
14 conservation of cytochrome P450s in the biosynthesis of glucosinolates and cyanogenic
15 glucosides. In addition, metabolic engineering using cytochromes P450 involved in cyanogenic
16 glycoside biosynthesis allows for the generation of acyanogenic plants that also display altered
17 glucosinolate profiles (24-25). It is evident that when environmental challenges such as insect
18 herbivores present themselves, specialization of enzymes from different pathways contributes to
19 the evolution of methionine-derived glucosinolates for plant survival.

20

21 In summary, we have determined the molecular changes responsible for the recruitment
22 of AtIPMDH from leucine biosynthesis for the specialized synthesis of glucosinolates. Future
23 studies need to explore protein level changes in other glucosinolate enzymes to understand
24 how the entire glucosinolate pathway evolved.

25

26 **Materials and Methods**

1

2 **Plants and Growth.** Seeds of *Arabidopsis thaliana* ecotype Columbia (Col-0) and SALK
3 mutant *atipmdh1* (Salk_063423C) were obtained from the Arabidopsis Biological Resource
4 Center (ABRC). Seed germination and plant growth conditions were as previously described
5 (11, 13).

6

7 **Plasmid Construction and Plant Transformation.** Oligonucleotides used in this study are
8 listed in Table S2. The full-length coding sequences of *AtIPMDH1*, *AtIPMDH2* and *AtIPMDH3*
9 were amplified using the Platinum Pfx DNA Polymerase (Invitrogen) with appropriate primer
10 pairs. PCR products were firstly cloned into pSC-B-amp/kan vector using StrataClone Blunt
11 PCR Cloning Kit (StrataClone), and then sequenced. Correct fragments were subcloned into the
12 *AtIPMDH1*pro::GUS vector (11) to generate constructs for each isoform and/or mutant under
13 control of the *AtIPMDH1* promoter. The resulting constructs were introduced into *Agrobacterium*
14 *tumefaciens* strain C58C1 followed by transformation into *atipmdh1* plants. Transgenic plants
15 were selected for hygromycin resistance and homozygous plants used for subsequent analysis.

16

17 **Glucosinolate Analysis.** Rosette leaves of 4-week-old plants and mature seeds were used
18 for glucosinolate analysis. Glucosinolates were analyzed using HPLC–mass spectrometry as
19 previously described (11, 13).

20

21 **Protein Expression, Purification, Assays, Crystallization, and Structure**
22 **Determination.** Expression and purification of wild-type and mutant AtIPMDH as histidine-
23 tagged proteins for functional analysis was performed as previously described (11). IPMDH
24 assay conditions using either 3-isopropylmalate or 3-(2'-methylthio)ethylmalate as a substrate
25 and the analysis of steady-state kinetic parameters were as previously described (11). For
26 crystallization of AtIPMDH2, the histidine-tag was removed by thrombin digestion and the

1 protein further purified using size-exclusion chromatography (27). Crystals of AtIPMDH2 were
2 obtained in 5 μ L hanging drops of a 1:1 mixture of protein and crystallization buffer (0.16 M
3 ammonium sulfate, 0.08 M sodium acetate trihydrate, 20% PEG 4000, 20% glycerol) at 4 °C
4 over a 0.7 mL reservoir. Data collection (100 K) was performed at the beamline 19-ID at the
5 Advanced Photon Source Argonne National Laboratory. Diffraction data was integrated and
6 reduced using HKL3000 (28). The structure of AtIPMDH2 was solved by molecular replacement
7 performed with PHASER 29) using the structure of IPMDH from *Salmonella typhimurium* (19) as
8 a search model. Model building was performed in COOT (30) and all refinements were
9 performed with Phenix (31).

10

11 **Site-Directed Mutagenesis.** Site-directed mutagenesis was performed using the
12 QuikChange PCR method (Stratagene). Bacterial expression vectors for each *AtIPMDH* (11, 13)
13 were used as templates with specific oligonucleotide pairs (Table S2). Protein expression,
14 purification, and assays were performed as described above for wild-type protein.

15

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17

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26

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- 12

Figure Legends

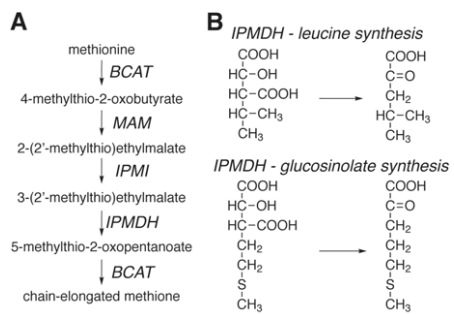
Fig. 1. (A) Overview of the methionine chain-elongation pathway of aliphatic glucosinolate biosynthesis in *Arabidopsis thaliana*. Note that a chain-elongated 2-oxo acid can serve as a substrate for MAM1 and MAM3 in subsequent rounds through the pathway to yield longer side-chain products. (B) IPMDHs catalyze the conversion of 3-isopropylmalate to 4-methyl-2-oxovalerate in leucine synthesis and the conversion of 3-(2'-methylthio)ethylmalate to 5-methylthio-2-oxopentanoate in glucosinolate synthesis.

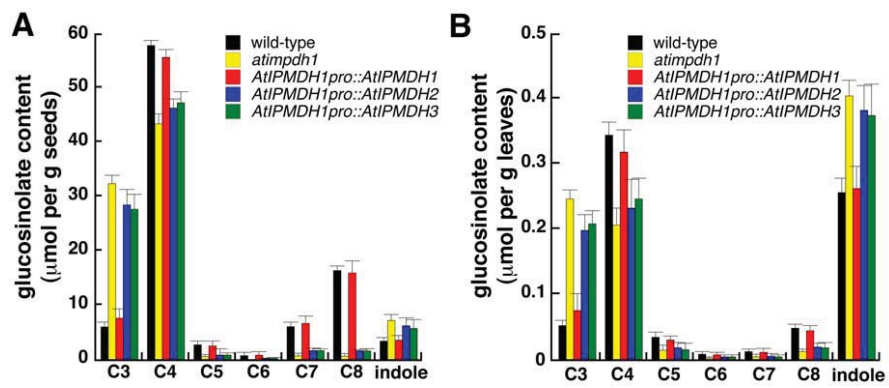
Fig. 2. Glucosinolate profiles in seeds (A) and leaves (B) from wild-type, *atipmdh1* mutant, and transgenic plants harboring each *AtIPMDH* driven by the *AtIPMDH1* promoter. Levels of aliphatic glucosinolates with varied methylene chain length (C3-C8) are shown. All indole glucosinolates are combined into a single group. Data are mean \pm standard deviation (n=3).

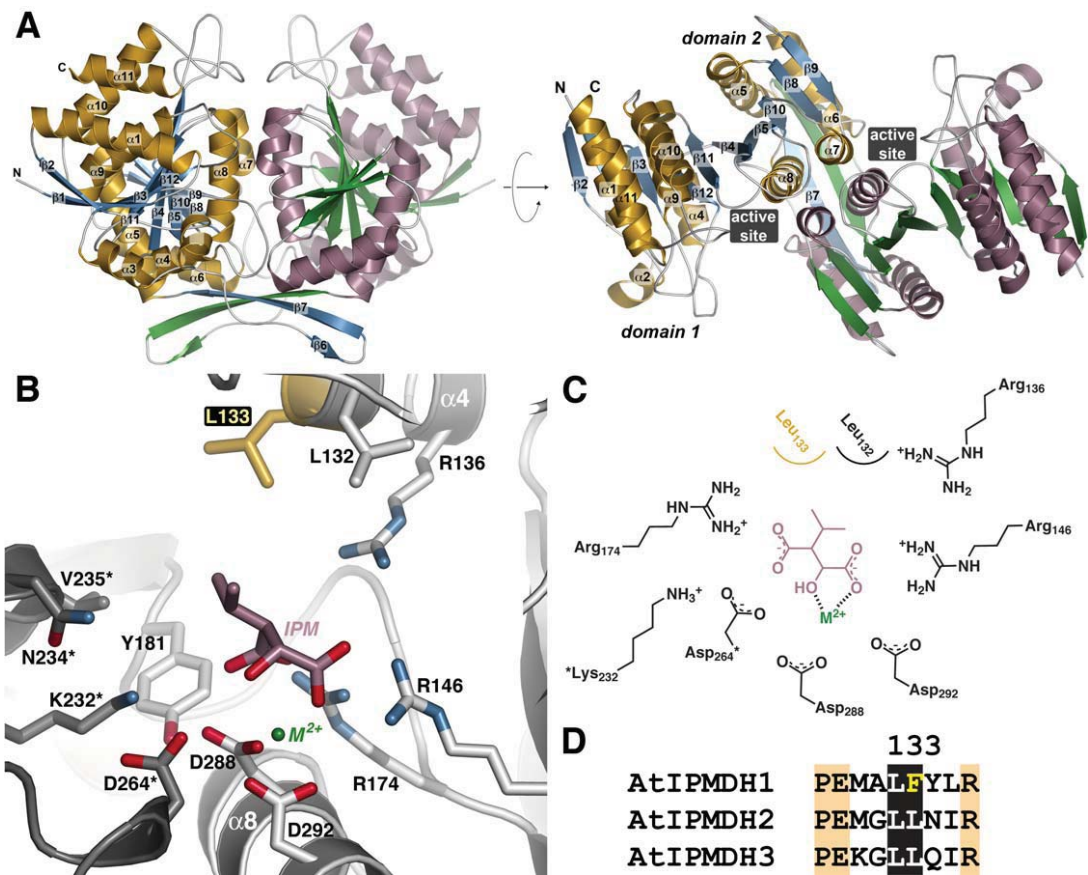
Fig. 3. Structure of AtIPMDH2. (A) Ribbon diagrams of the AtIPMDH2 dimer. Monomer A is shown with gold α -helices and blue β -strands and monomer B is drawn with rose α -helices and green β -strands. Secondary structure features are labeled on the A monomer. The left view shows the dimer down the 2-fold axis. The right view is rotated 90° to show the two domains of each monomer. The position of the active site cleft is indicated. (B) Active site view and model of 3-isopropylmalate (IPM) and divalent metal (M^{2+}). Side-chains of active site residues are shown with those from the adjacent monomer (grey) indicated by an asterisk. The positions of the substrate and metal are modeled based on the bacterial structures (18-20). The active site difference among the AtIPMDH isoforms is highlighted in gold. (C) Schematic of the active site model. (D) Sequence comparison of the region including residue 133 (AtIPMDH2 numbering).

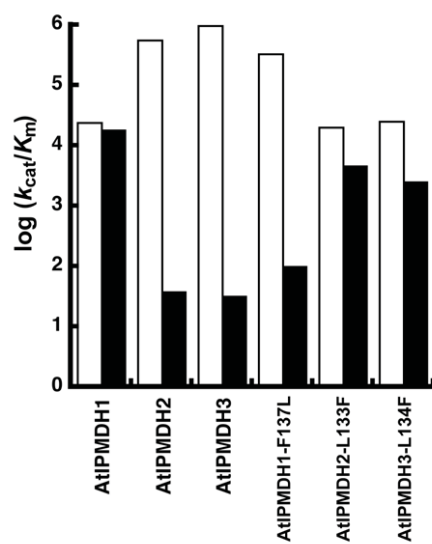
Fig. 4. Comparison of the catalytic efficiencies (k_{cat}/K_m) of wild-type and mutant AtIPMDH using 3-isopropylmalate (white bars) and 3-(2'-methylthio)ethylmalate (black bars).

Fig. 5. Glucosinolate profiles in seeds (A) and leaves (B) of wild-type, *atipmdh1* mutant, and transgenic plants expressing AtIPMDH1-F137L, AtIPMDH2-L133F and AtIPMDH3-L134F driven by *AtIPMDH1* native promoter. Levels of aliphatic glucosinolates with varied methylene chain length (C3-C8) are shown. All indole glucosinolates are combined into a single group. Data are mean \pm standard deviation (n=3).









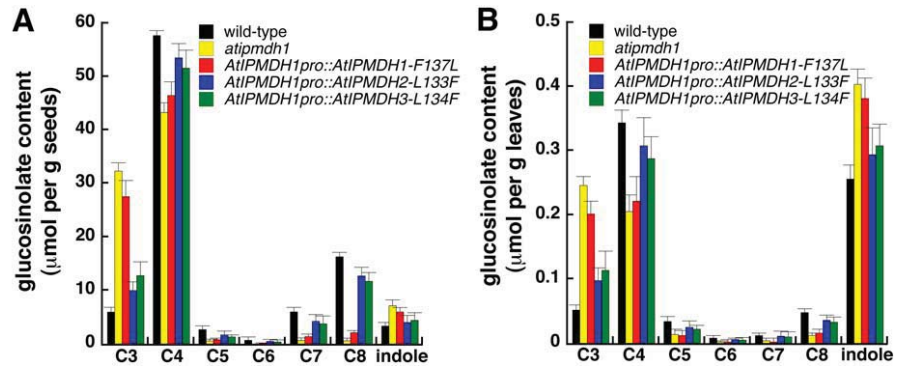


Table 1. Kinetic parameters of wild-type and mutant AtIPMDHs.

	<i>3-isopropylmalate</i>			<i>3-(2'-methylthio)malate</i>		
	k_{cat} (min^{-1})	K_{m} (μM)	$k_{\text{cat}}/K_{\text{m}}$ ($\text{M}^{-1} \text{s}^{-1}$)	k_{cat} (min^{-1})	K_{m} (μM)	$k_{\text{cat}}/K_{\text{m}}$ ($\text{M}^{-1} \text{s}^{-1}$)
AtIPMDH1	37 ± 4	25.2 ± 2.3	24,471	51 ± 5	45.3 ± 3.6	18,763
AtIPMDH1-F137L	230 ± 14	11.4 ± 1.7	336,257	2.0 ± 0.2	323 ± 21	103
AtIPMDH2	373 ± 33	10.9 ± 1.3	570,336	1.0 ± 0.2	435 ± 32	38.3
AtIPMDH2-L133F	37 ± 5	30.3 ± 2.5	20,352	22 ± 1	77.0 ± 8.5	4,761
AtIPMDH3	543 ± 36	9.2 ± 1.4	983,696	1.0 ± 0.1	502 ± 35	33.2
AtIPMDH3-L134F	44 ± 5	28.5 ± 1.5	25,731	16 ± 1	103 ± 10	2,589

Data are means ± standard error (n = 3).